

Multidisciplinary Aspects of Developing Small Sensing Devices for Monitoring Chemicals and Biochemicals

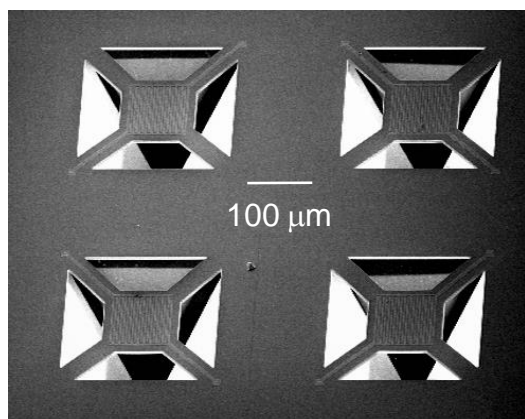
Steve Semancik

Biomolecular Measurement Division

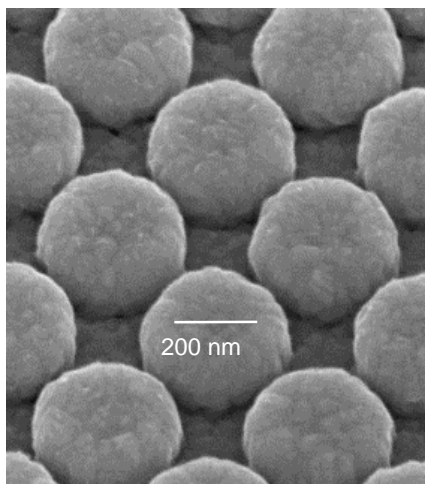
National Institute of Standards and Technology

Gaithersburg, Maryland USA

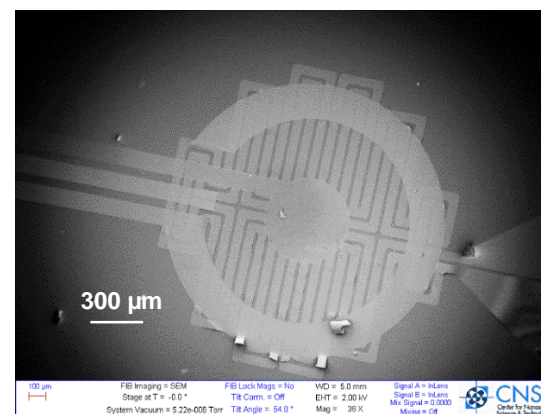
stephen.semancik@nist.gov



**Chemiresistive
Gas Sensing**



**Plasmonic Gas
& Solution Sensing**



**Electrochemical
DNA-Based Monitoring**

small device platforms, nanomaterials, temperature variation

National Institute of Standards and Technology (NIST)



Gaithersburg, Maryland

Contributors to efforts at NIST:

Materials

Kurt Benkstein
Carlos Martinez*
Josh Hertz*

Platforms

John Suehle
Mike Gaitan
Richard Cavicchi

Gas Sensing

Doug Meier*
Phil Rogers*
Barani Raman
Tekin Kunt
Yangyang Zhao

Biosensing

Charles Choi*
Herman Sintim
Zuliang Shen
Sarah Robinson

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Growing Demand for Sensors: Examples

iPhone and tablet apps, wearables



Fitbit

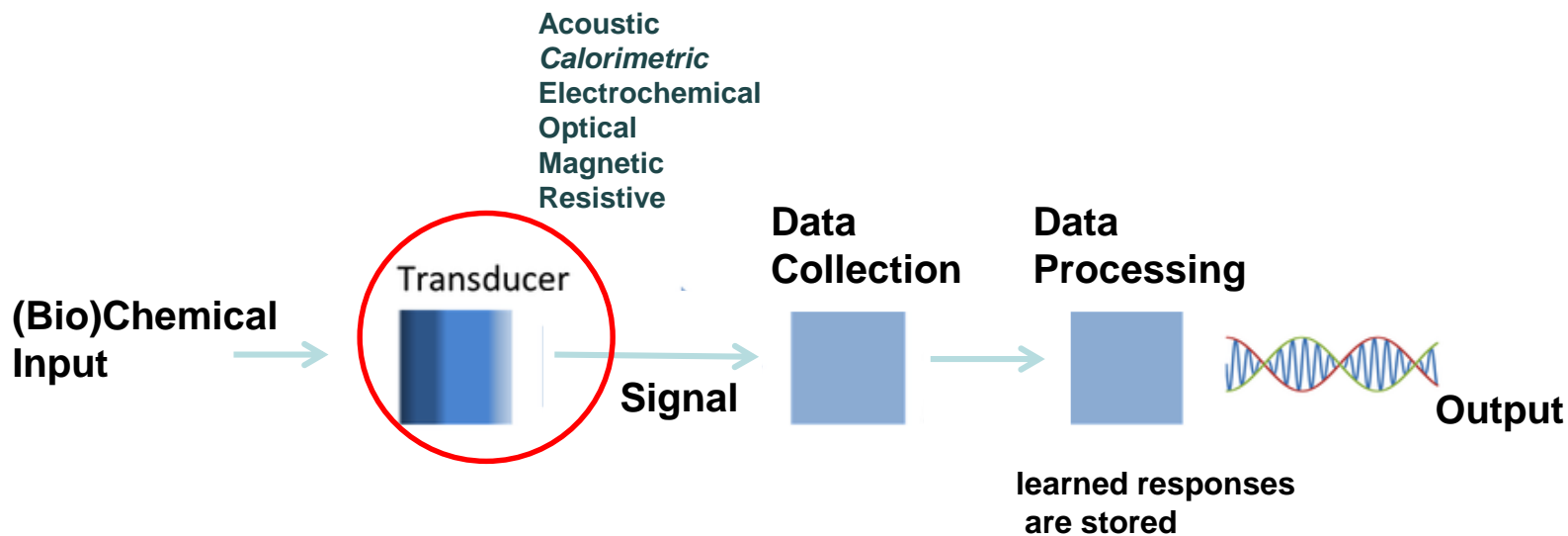


- Electronics/interfaces/communication components are much more mature (low \$\$)
- Physical sensing is more prominent in commercial devices – easier (fewer degrees of freedom)
- Need good sensor devices for chemicals/biochemicals - reliable “chemical signals”

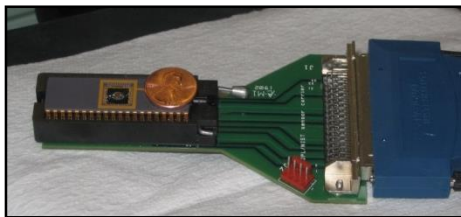
chemical/biochemical interactions produce a richness that is much more challenging (sensitivity, interferences)

Microdevice Sensing Technology

device-based electronic sensing can add convenience at low cost



Themes



- **Functional microplatforms**
- **Integrated nanomaterials** for transduction
- **Signal processing** of data streams
- **Operational concepts** and enabling technology

Bio-Inspiration for Chemical Detection

Insects



tracking low molecular conc's. of target molecules to locate food or a mate



evolutionary success (biological olfaction) challenges "electronic noses" (artificial olfaction)

Trained Dogs



reliable detection of drugs, explosives and disease [not overly convenient or network friendly]



"system" is sensitive and fast

Measurement Choices: Sensors vs Instruments

Instruments

larger, more expensive, "gold standard"

Direct measurement of chemical target

Mass spectrometry

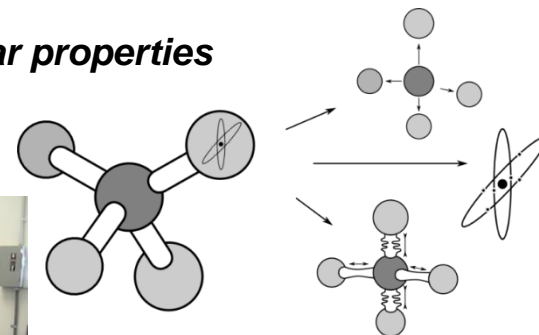
IR-vis-UV/fluorescence spectroscopy

photo-electron spectroscopy

NMR spectroscopy



intrinsic molecular properties



Orbitrap LC/MS
~ \$1M

Sensors

smaller, cheaper, less precise, screening/networks

Indirect detection of chemical target

Electrochemical

Colorimetric/Opto-chemical

Bio-molecular (assays)

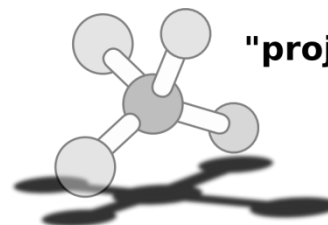
Solid State



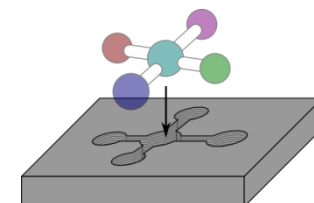
< 1 cm, \$10s to \$100s

effects from interactions

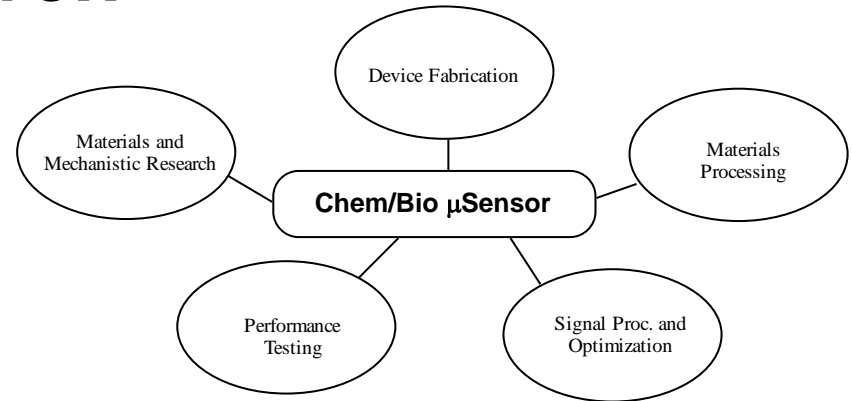
lock and key



"projection"



Multidisciplinary Research



Chemistry

- kinetics, thermodynamics
- surface reactions
- etching/micromachining reactions

Physics

- semiconductor/insulator properties
- electron transfer

Engineering

- design and fabrication of microdevices
- testing equipment



Materials Science

- thin films
- nanomaterials

Math

- data collection
- signal processing

Biology

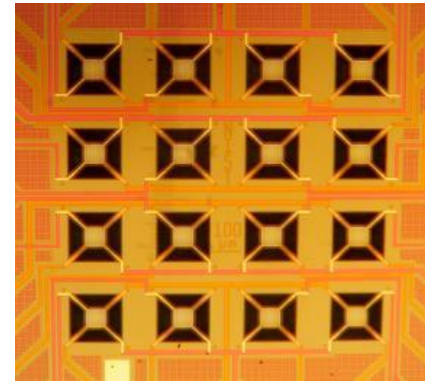
- biomolecule binding
- medical screening

Outline



- **Chemiresistor Array**

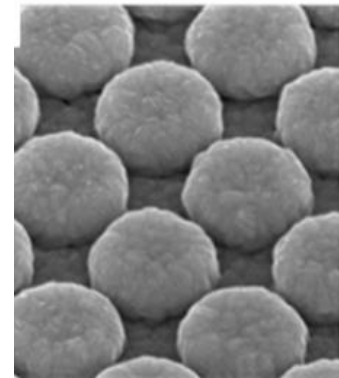
application-adaptable gas sensing



100 μm
elements

- **Plasmonic Optical Sensor Platform**

gas-phase and condensed-phase sensing

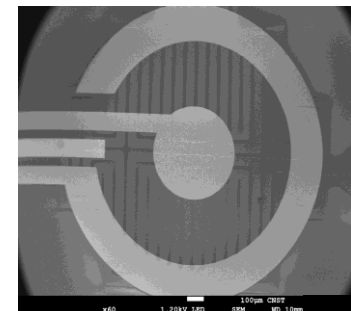
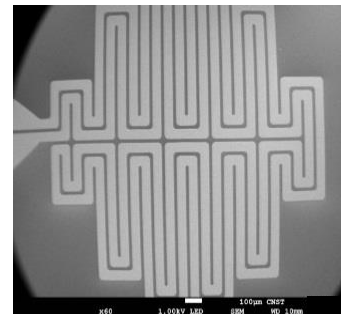


380 nm
features

- **Microscale Electrochemical Device**

biochemical characterization

DNA on
500 μm
working
electrode

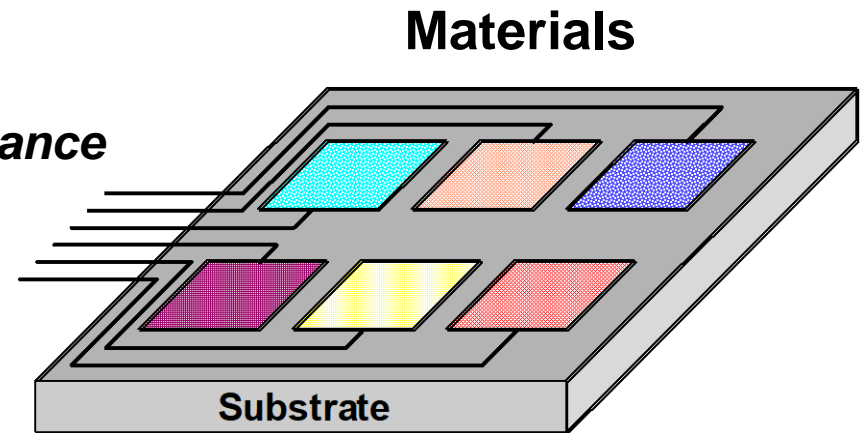


Chemical Sensor Array Concepts: *Adaptable Platform*

Solid State **Chemiresistor** Devices

gas-induced changes in electrical conductance

- semiconducting oxides
- conducting polymers
- chemical and electronic modifiers
- nanotextured and nanostructured films

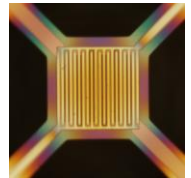
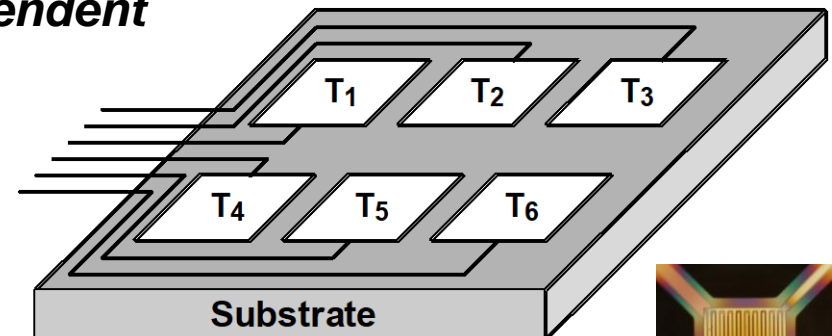


materials-dependent and temperature-dependent surface interactions

- adsorption $f_1(T)$
- desorption $f_2(T)$
- coadsorbate reactions $f_3(T)$

T(t) programming controls these phenomena in time

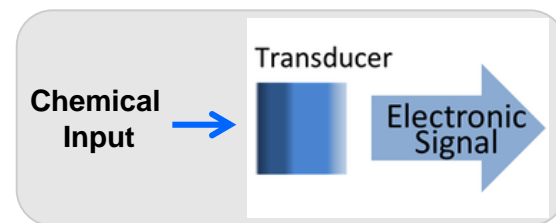
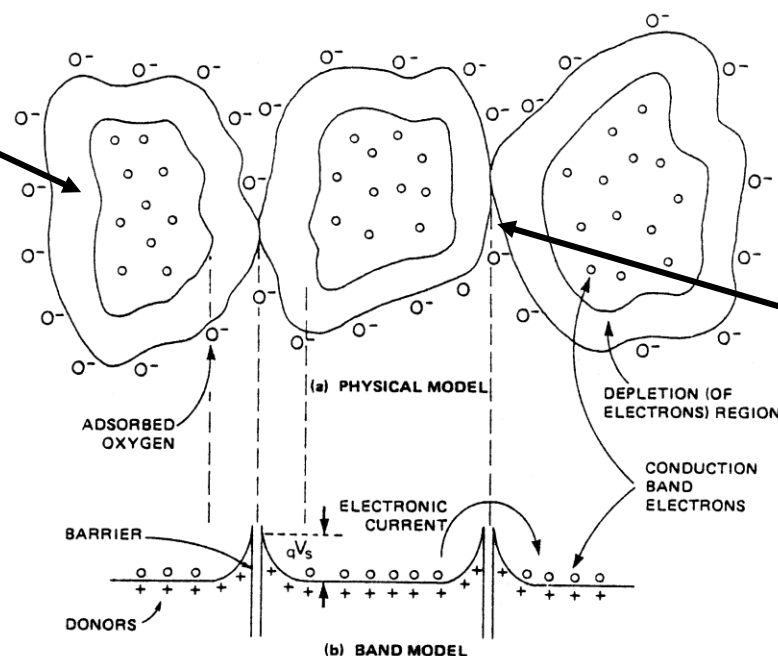
Operating Temperature $T_n(t)$



approach produces analytically rich datastreams and allows a tunable technology for varied applications

Chemiresistive Sensing: Mechanistic Effects

Variable electron depletion



Interparticle percolation

- Oxygen vacancy defects - set base conductance levels for oxide films
- Surface charge transfer to bonded adsorbates
- Adsorbates react to control adsorbed oxygen (air background)

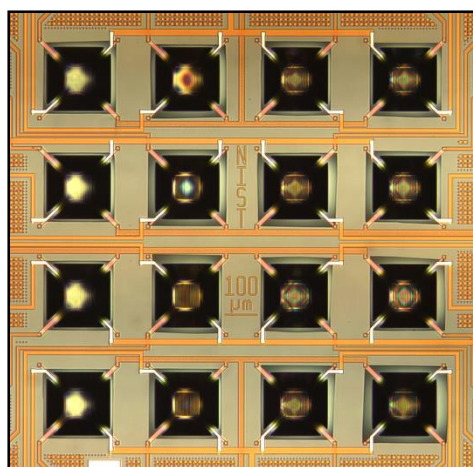
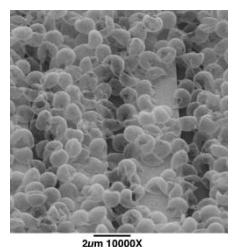
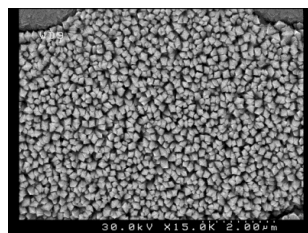
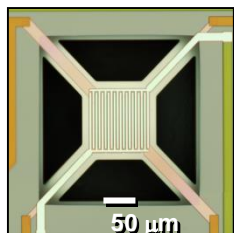
T and microstructure are critical to transduction/performance (chemical and electronic interfacial properties)

sensitivity, selectivity, stability, speed

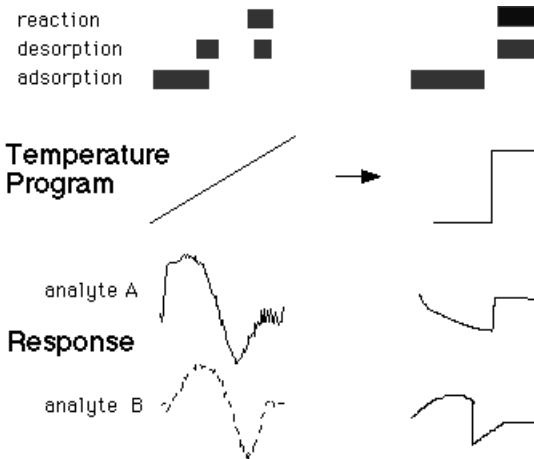
Approach: Trace Detection in Varied Backgrounds

tunable microsensor arrays

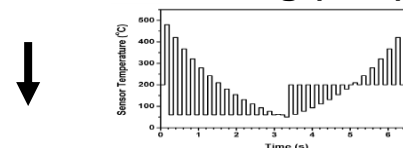
- MEMS platforms
- robust chemiresistive nanomaterials
- surface and sensor science
- advanced signal processing



independent chemiresistive
MEMS microhotplate array elements



increased analytical content per unit time with
Temperature Programmed Sensing (TPS)



large $T(t)$ databases from
“virtual sensors”

Signal Processing

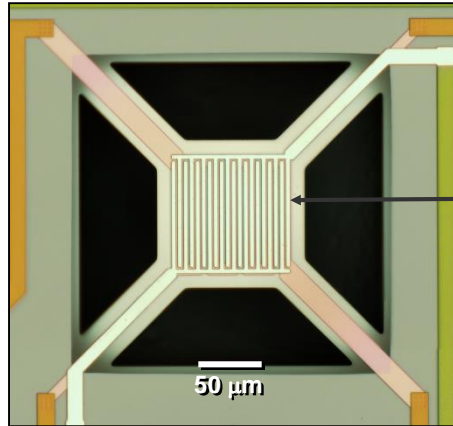
target detection/recognition/monitoring

Microhotplate Platforms for Chemiresistive Sensing

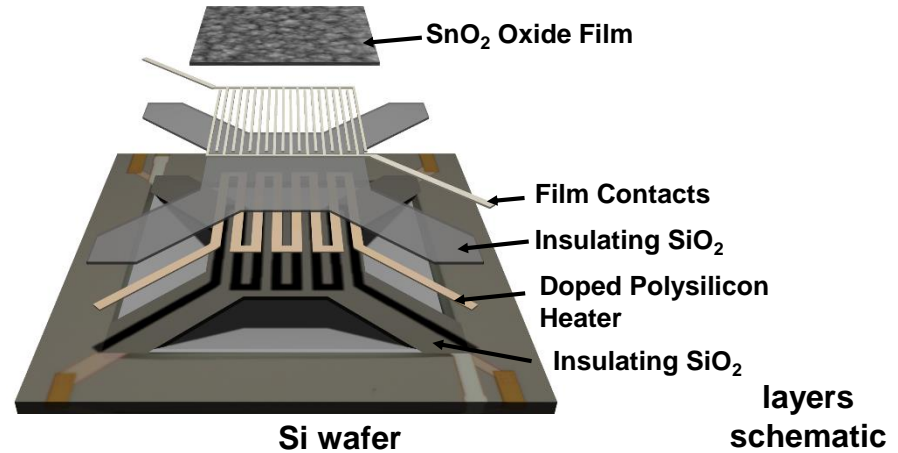
single
microhotplate
element

(mass ~ 250 ng;

heating/cooling $t_c < 5$ ms)



Suspended
Structure
(micromachining
of Si)



layers
schematic

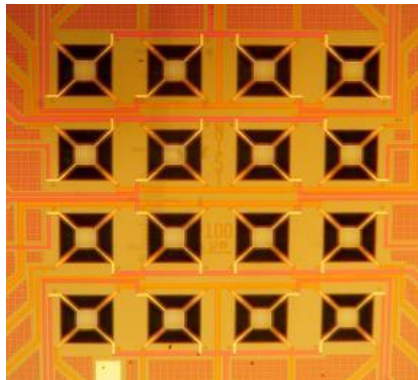
Functionality

- T measurement and control
- electrical characterization

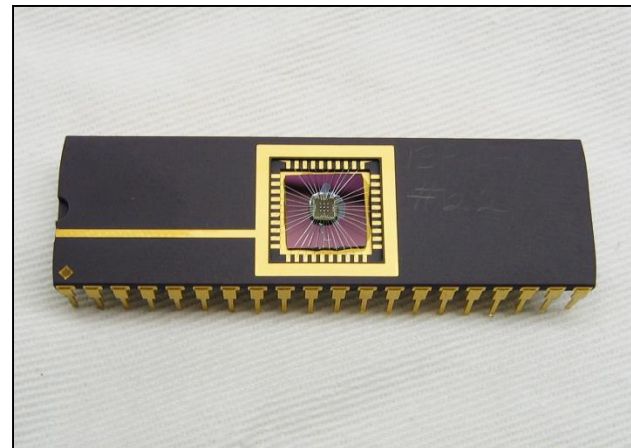
Features

- 20 °C to 500 °C; (~ 20 °C per mW)
- capable of heating rates of 10⁵ - 10⁶ °C/s
- CMOS design rules

replication to
16-element
array



multi-channel electronics used
for operation & data collection



packaged
device

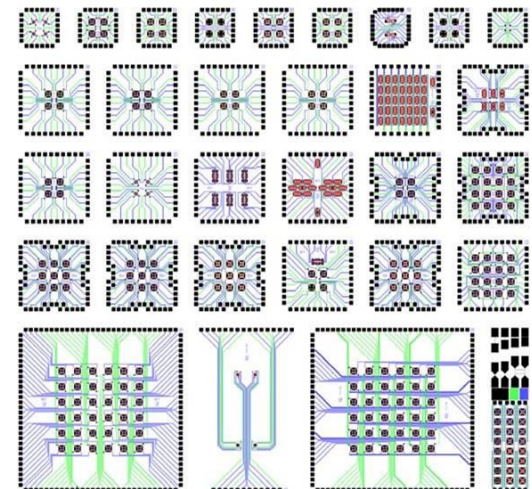
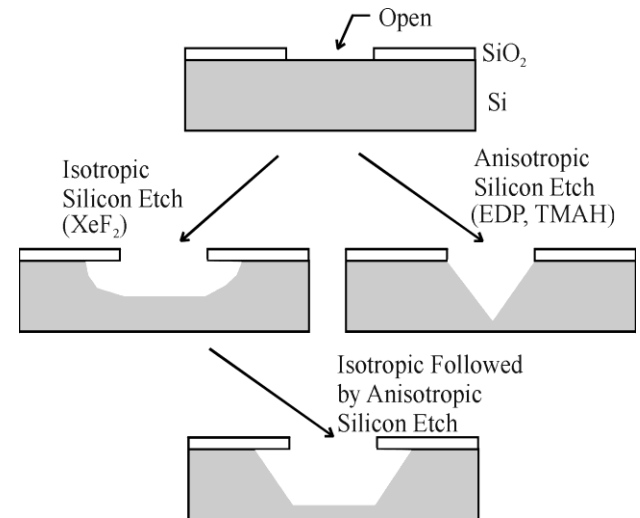
MEMS Platform Fabrication

Processing Steps

- Design of multilevel mask set*
- Fab run at Si foundry
- Dicing of wafer and/or die
- Si etch “micromachining”
- Post-processing (e.g. - contacts)
- Mounting/wirebonding

* *CMOS design rules*

Top-Surface Si Micromachining



15 mm x 15 mm die with multiple micohotplate-related designs

*75 die on 6" wafer
(~ 1500 array devices)*

Film Deposition and Processing Methods

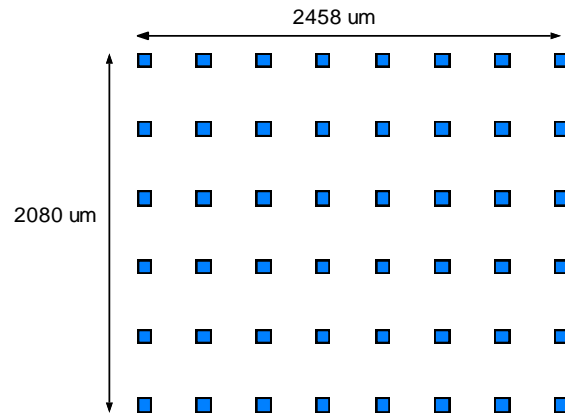
Local Heating

- Thermally-activated CVD
- Sol-gel, suspension drying
- Annealing
- Thermal lithography
- Thermally-assisted imprinting

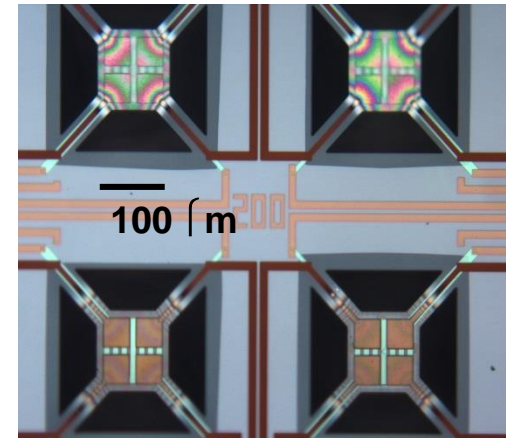
Local Potential Control

- Addressable electrochemical or electrophoretic deposition

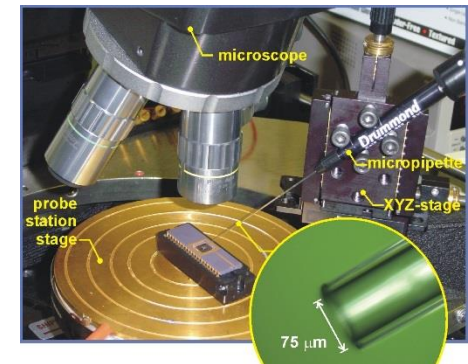
Masking



litho-defined shadow mask



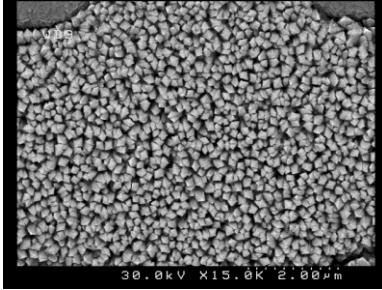
Micro-Dispensing (pipetting)



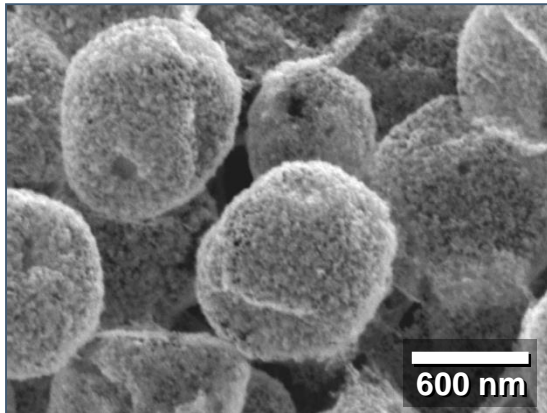
all processing done after etching and packaging (to avoid etchants and to use electrical contacts)

Examples: Integrated Nanostructured Materials

High Area and Nanoparticle Materials

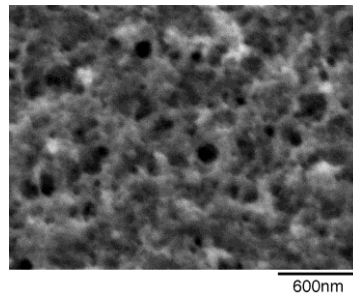


SnO₂ thin film

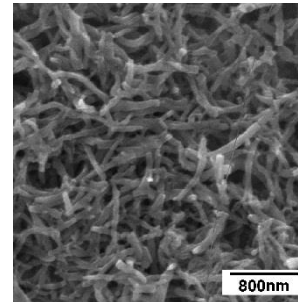


Sb:SnO₂ nanoparticle microshells

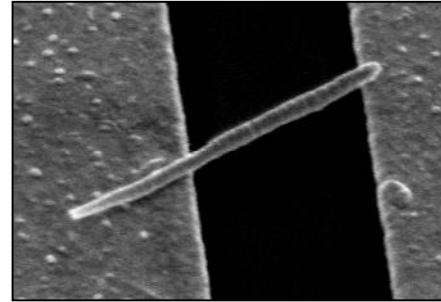
“Fast” Nanostructured Polymers



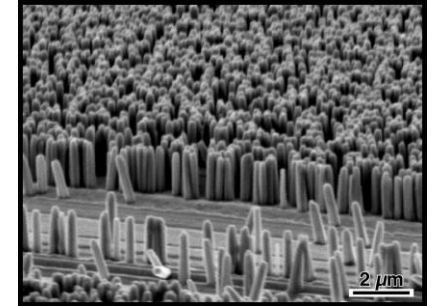
spongy PANI/nafion



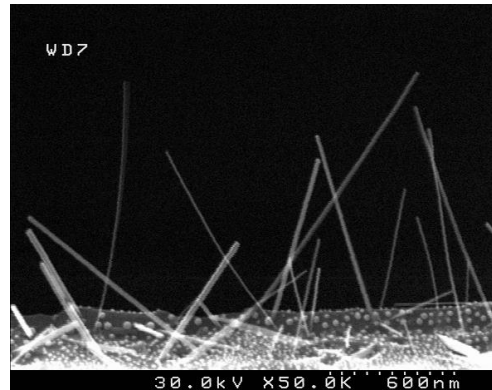
Nanowires, Nanorods and Nanotubes



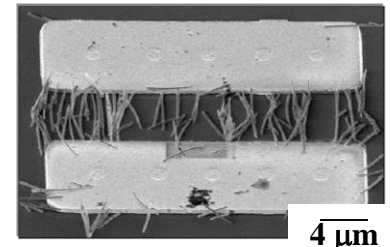
WO₃ wires



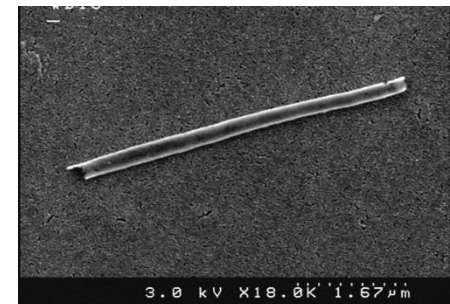
ZnO rods



CNTs

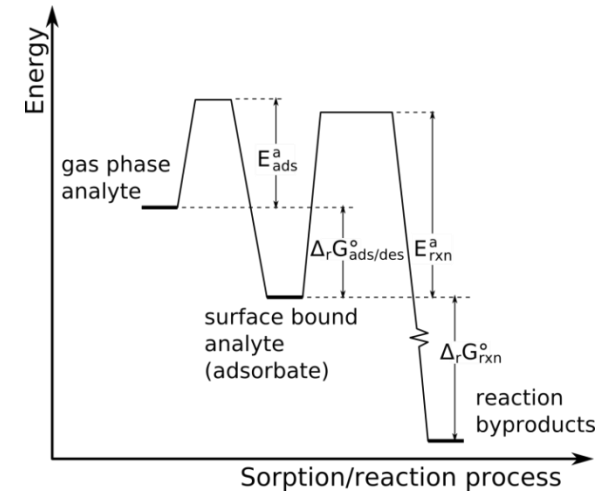
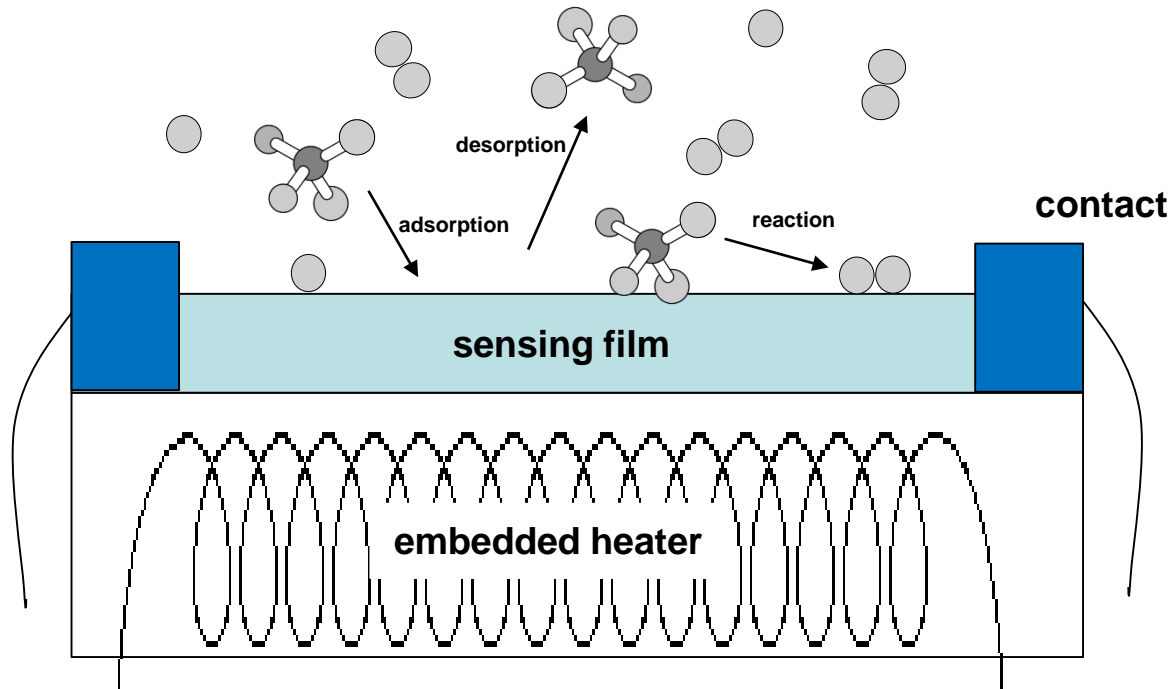


SnO_x nanowires



TiO₂ nanotube

Temperature in Gas-Phase Chemical Sensing



$$s = f(T, \Theta)$$

$$R_{ad} = \frac{d\Theta}{dt} = \frac{s_0}{N_{\max}} \cdot f(\Theta) \cdot \frac{P}{\sqrt{2\pi mkT}} \cdot e^{-\frac{E_{ad}^*}{kT}}$$

$$R_{des} = -\frac{d\Theta}{dt} = v_x \cdot \Theta^x \cdot N_{\max}^{x-1} \cdot e^{-\frac{E_{des}}{kT}}$$

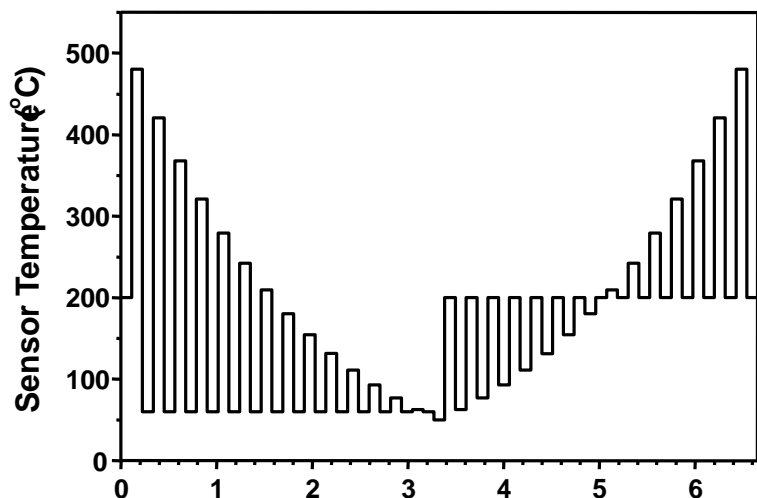
Klaus Christmann

Sticking coefficient, adsorption and desorption rates - enable interesting operating modes when platforms have *time constants in ms range*

(Chemiresistive Sensing)

Higher Dimensionality: Temperature Programmed Sensing

2-Base, Exponential TPS Cycle



Time (s)

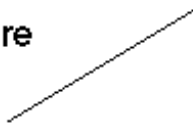
reaction
desorption
adsorption



diffusion & charge
exchange also

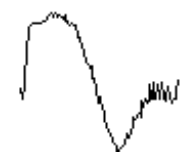


Temperature
Program



Response

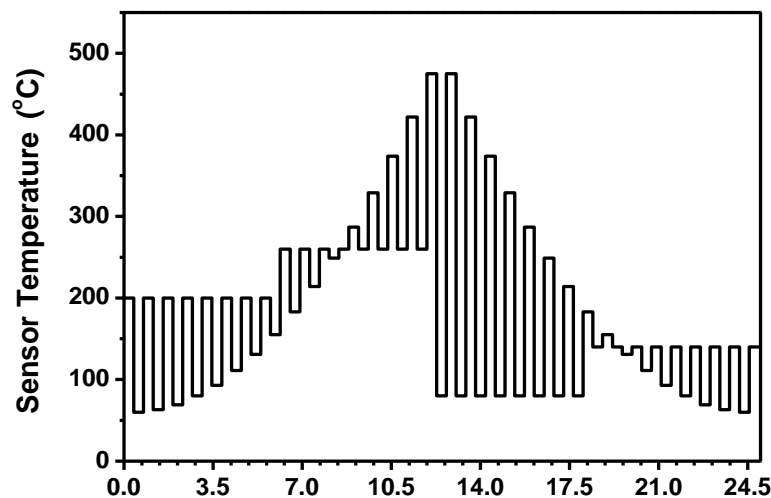
analyte A



analyte B



4-Base, Quadratic TPS Cycle



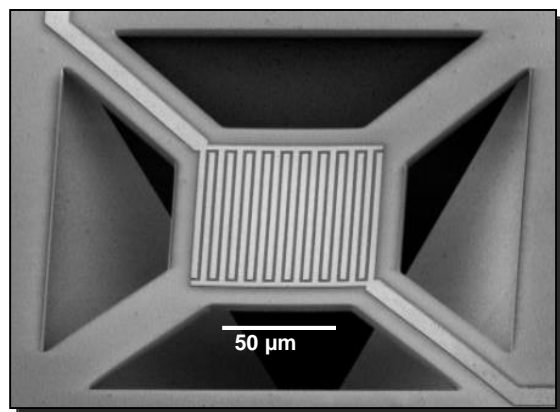
Sensor Temperature (°C)

Time (s)

- gas-film interactions alter film's electronic transport
- purposeful temperature excursions as a function of time drive surface phenomena that produce an analytically-rich temporal signal
- TPS cycles are run repeatedly to collect environmental information

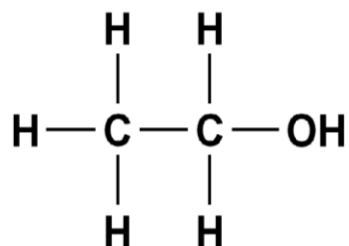
Single Microdevice Sensing

With fast temperature modulation, what discrimination can be accomplished?

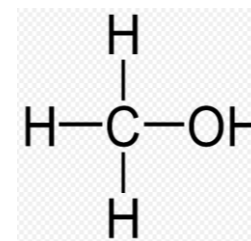


TPS-based operation of a single sensor
(SnO₂ film + ~20A annealed Pd)

Challenge – discriminating 2 similar chemicals



ethanol



methanol

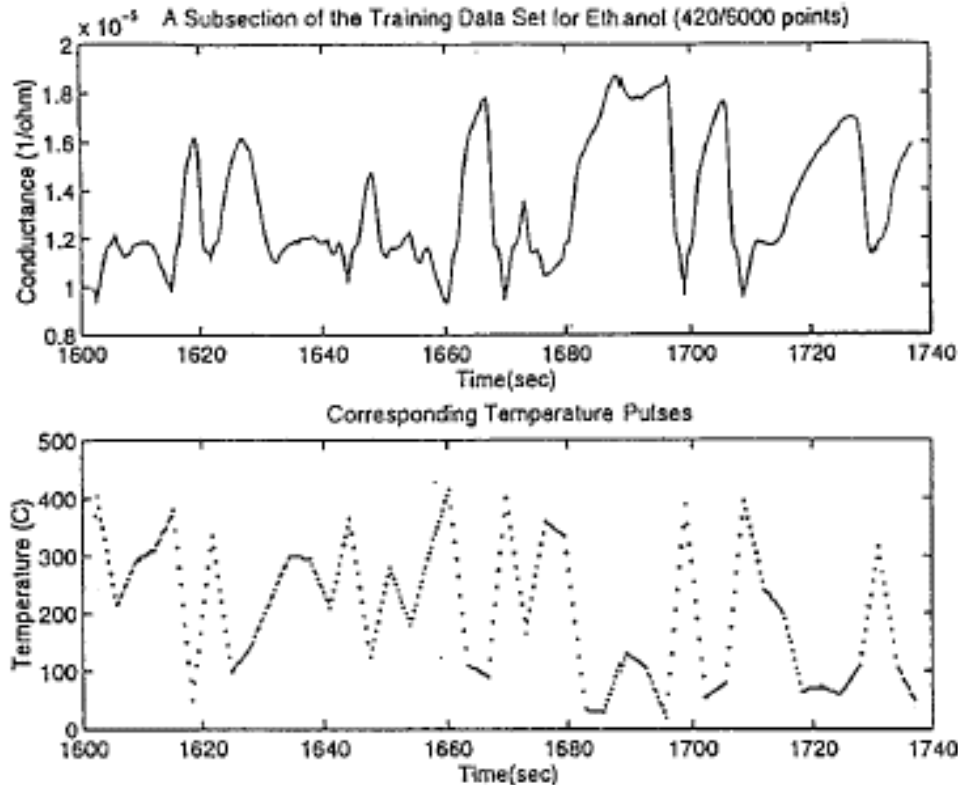
alcohol vapors in air

How should the temperature of the microsensor be varied to best discriminate between the two alcohols?

Let the sensor LEARN the answer

Learning about the Sensor's Response vs T

using semi-random response information to aid discrimination



Measured response (for ethanol)

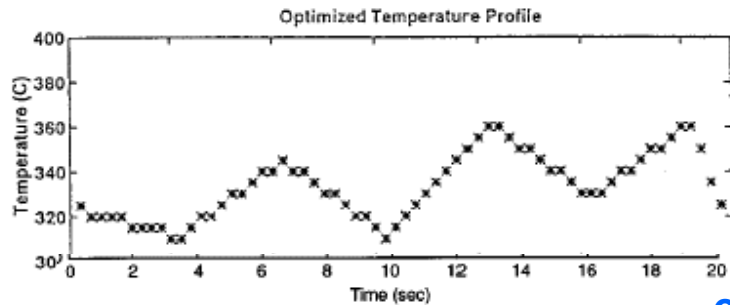
Temperature can be flexibly programmed to vary between $\sim 20^\circ\text{C}$ and 450°C

**Random jumps were found to be too erratic -
but 10 points along randomly generated T slopes worked well,
with T reset period ~ 300 ms**

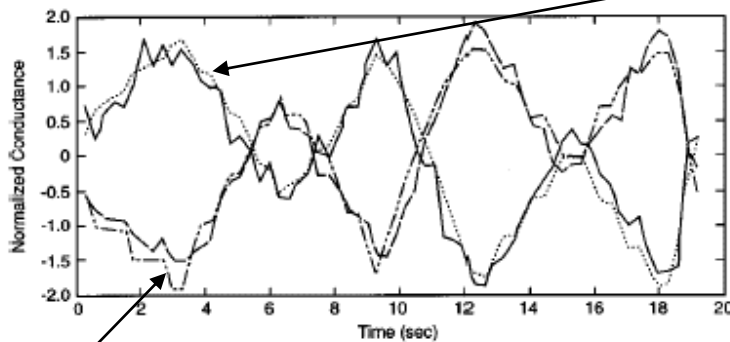
T. A. Kunt, T. J. McAvoy, R. E. Cavicchi and S. Semancik, Optimization of temperature programmed sensing for gas identification using micro-hotplate sensors, (Sensors and Actuators B 53, 24-43)

Predicted Response/Measured Response Comparisons

T. A. Kunt, T. J. McAvoy, R. E. Cavicchi and S. Semancik, Optimization of temperature programmed sensing for gas identification using micro-hotplate sensors, (Sensors and Actuators B 53, 24-43)



**Predicted optimal discrimination
T program**
(by maximizing learned response separation)



ethanol prediction
ethanol data

**Optimized operation predicts
“polarity reversals”
(out-of-phase) responses
for ethanol and methanol
*which greatly aids discrimination***

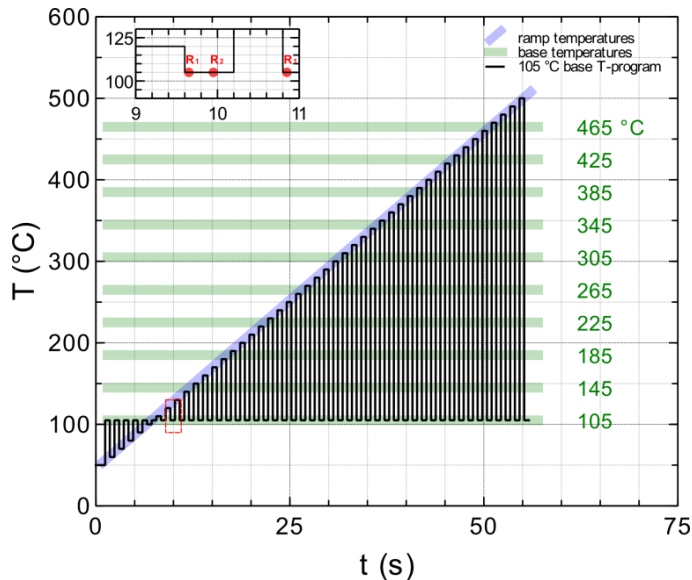
methanol prediction
methanol data

(Minimum discrimination time for these alcohols: ~ 2.2 sec)

[with rich data and PCA/LDA can do mixtures]

Systematic Data Acquisition

for development of application-optimized T programs

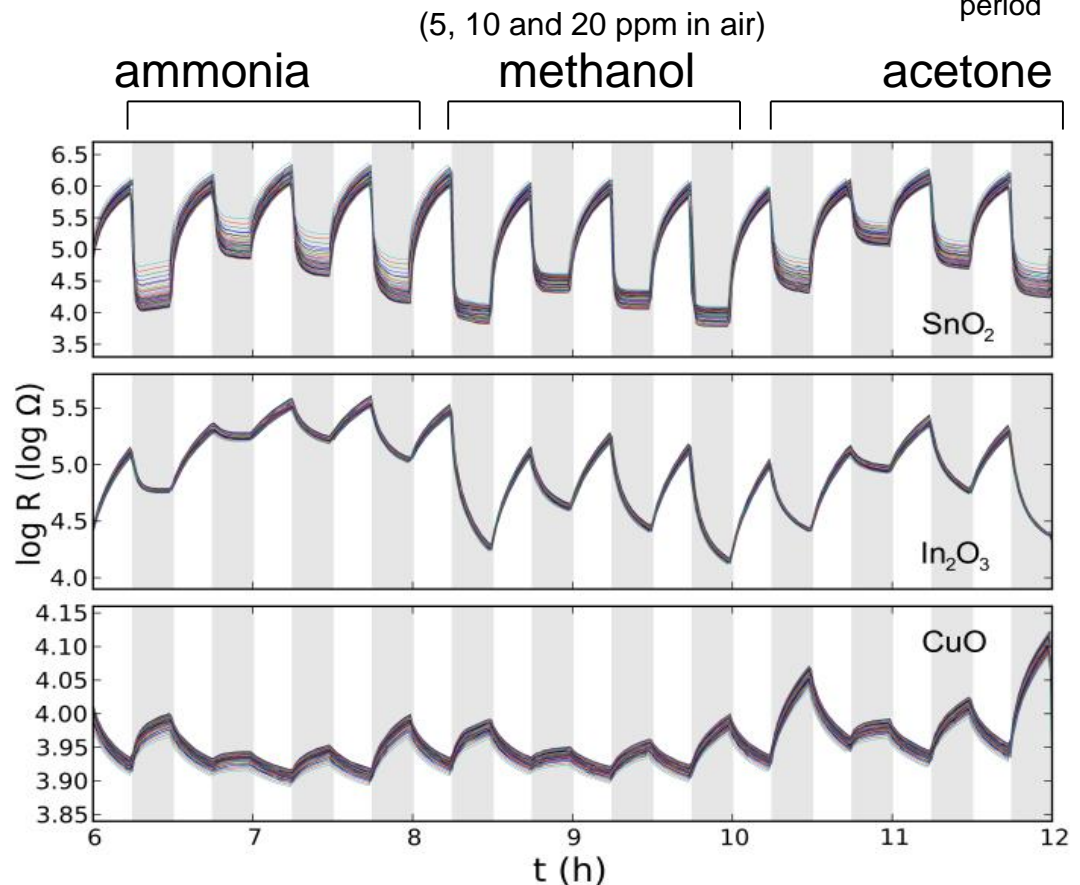


parametric pulsed-ramp programs employed for elements with 3 types of sensing films

locate high-value portions of TPS datasets via linear discriminant analysis (LDA) cluster separability and biplots

46 pseudo-isotherms (for each sensing material below) following all ramp-to-base transitions during the $T_{\text{base}} = 145 \text{ }^\circ\text{C}$ 6-hour exposure cycle [1/10 of systematic database]

trace
target
exposure period

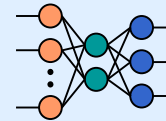


Signal Processing Methods (AI/ML)

methodology for extracting best analytical information

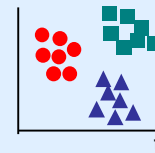
Recognition Classifiers

Artificial Neural Network (ANN)
K-nearest neighbors



Dimensionality Reduction and Data Preprocessing

Principal component analysis (PCA)
Linear discriminant analysis (LDA)
Baseline correction



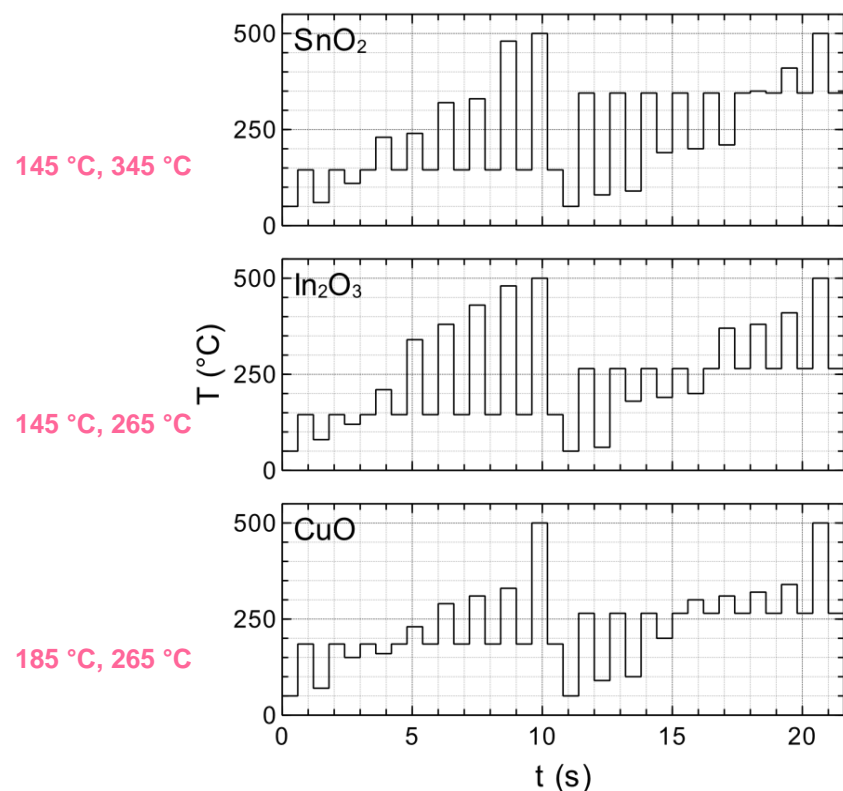
New Techniques for Improving Robustness

Bio-inspired signal processing
– Hierarchical Recognition Method
Event Detection

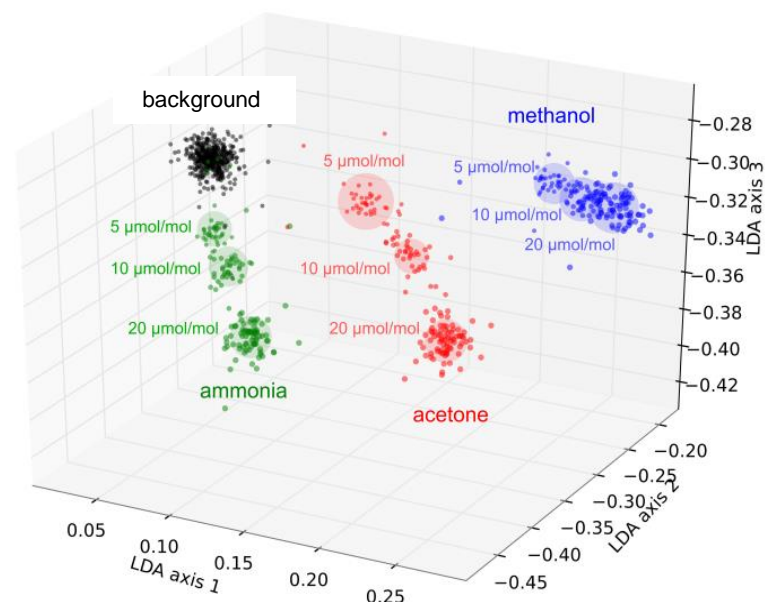
Pruned Optimization Programs/Resulting Performance

sensing material-dependent condensed
T program assembled using 2 T_{base} with
9 T_{ramp} inputs each

resulting separability of target
analytes at different concentrations



*22-second programs covering
2 bases and a variety of pulsed
excursions*



linear, unoptimized TPS

each 55 sec

0.47 cluster separability

optimized TPS (at left)

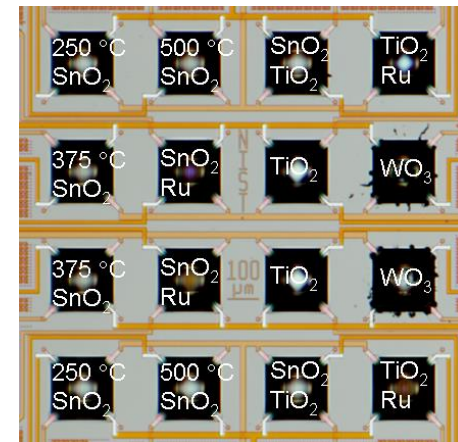
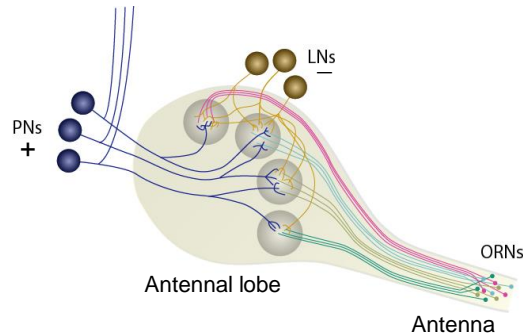
each 22 sec

0.73 cluster separability

Recognition and Classification

Taking things a bit further -

toward a hierarchical/bioinspired method for dealing with different types of chemicals



16 materials (2 copies of 8 films)

temperature ramped from 50 to 500 °C (in 1 °C increments)

16 materials x 450 temperatures = 7200 pseudo-sensors

For detecting compounds trained on, and “classifying” the class of unknowns

Recognize Trained Analytes/Classify Unknowns

Approach Examined:
Hierarchical Method

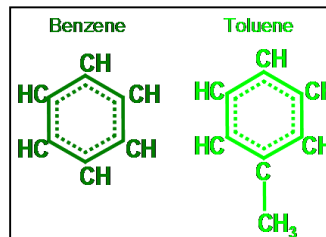
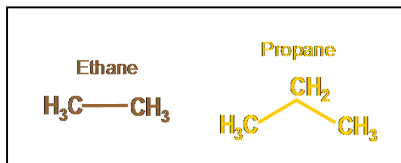
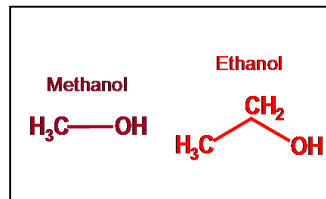
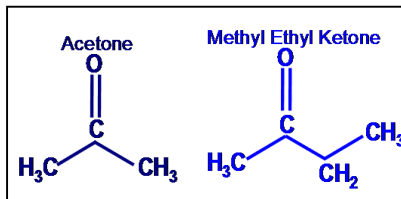
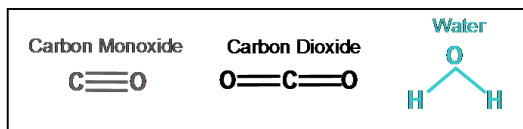
Training



Age the Sensor



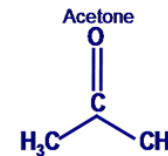
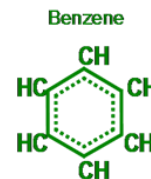
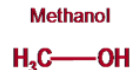
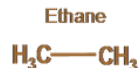
Test



Chemical Classes

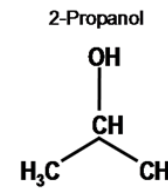
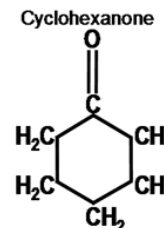
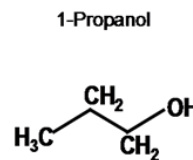
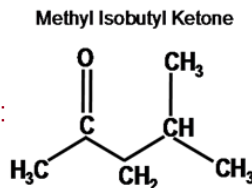
Simple Oxides
Ketones
Alcohols
Alkane
Aromatics

Repeated from training run:



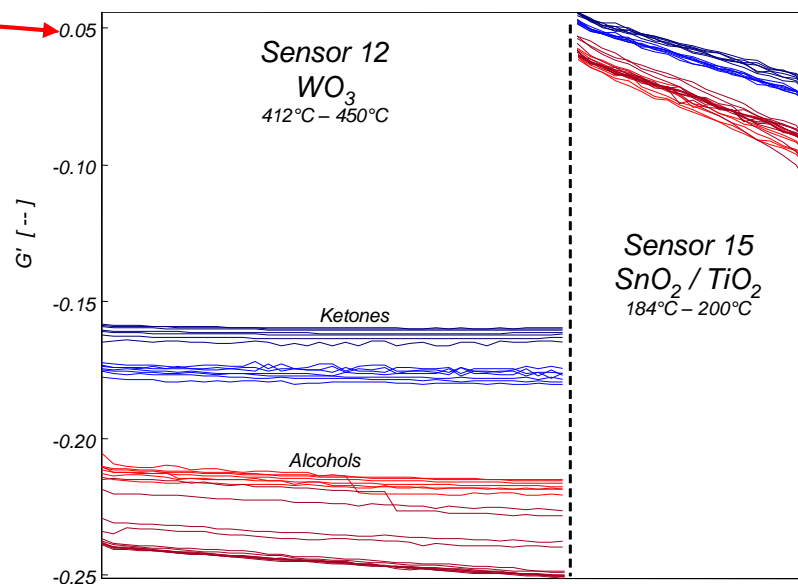
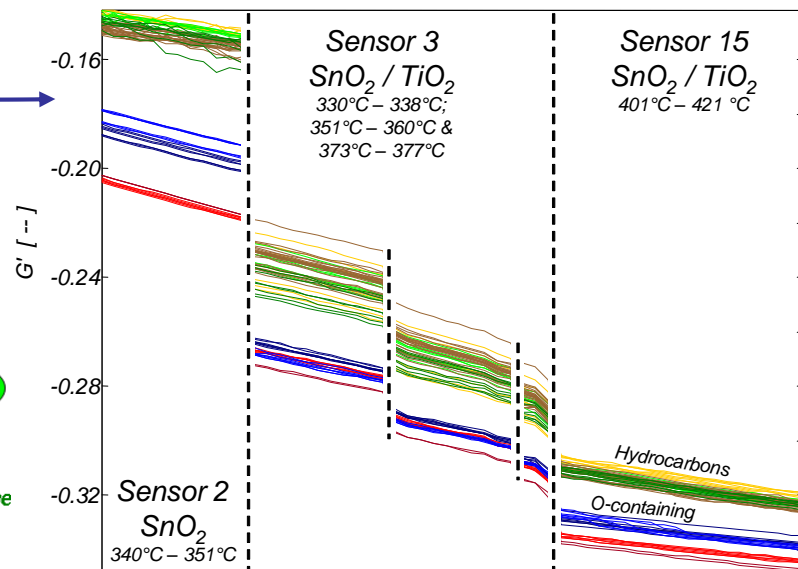
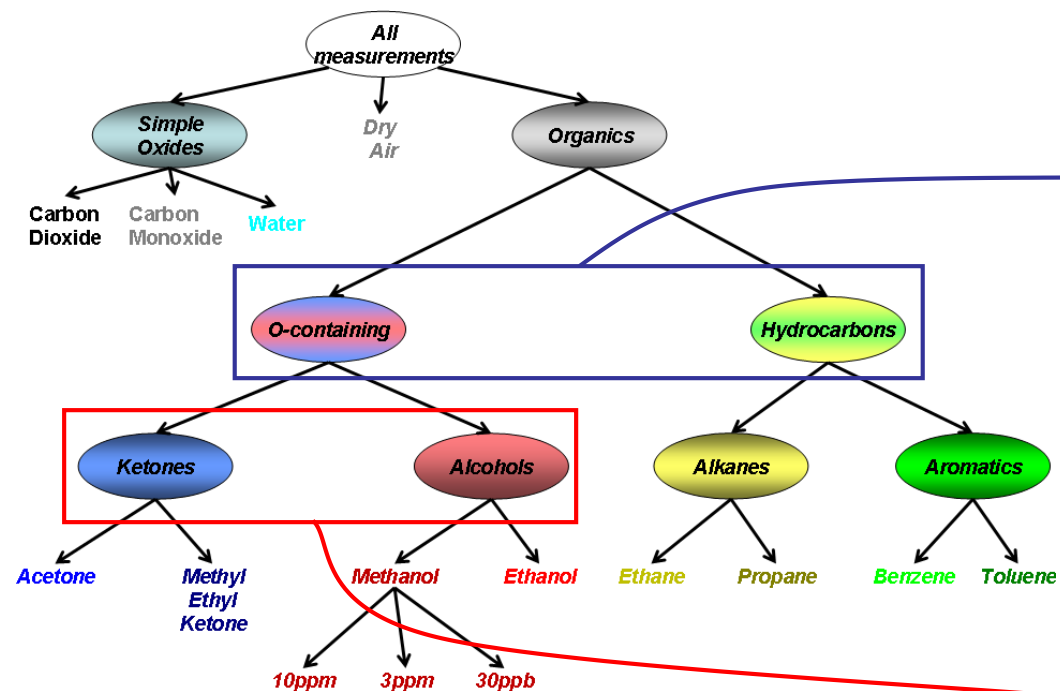
Recognition Task

New analytes at ~1-10 ppm:
"unknowns"



Classification Task

Optimized Analytical Answers from Rich Database

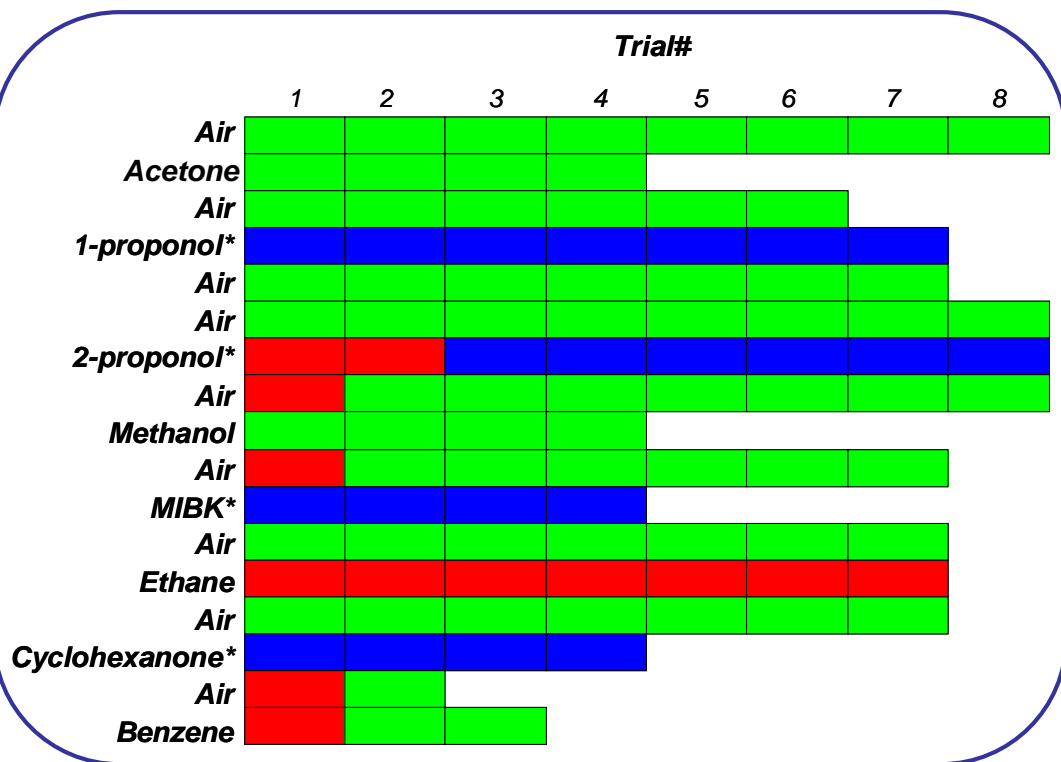


- Employ (different) optimized regions from the measured materials-dependent and T-dependent training set to address each successive recognition/classification problem

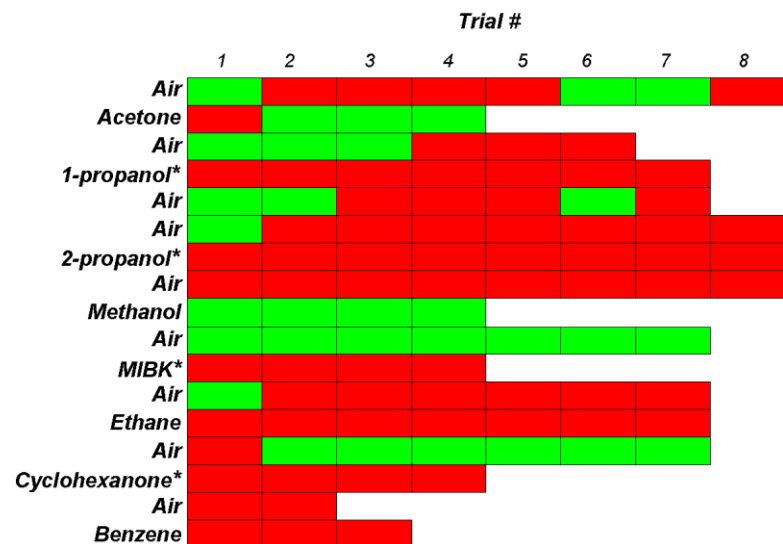
Bio-Inspired Methodology for Artificial Olfaction
 B. Raman, J. Hertz, K. Benkstein, S. Semancik,
 Analytical Chemistry 80, 8364-8371

New Signal Processing Approaches - Hierarchy

Divide-and-Conquer Approach



Old, 'All-at-Once' Approach



* Indicates chemicals not used in the training set

Mis-Classification



Correct Recognition

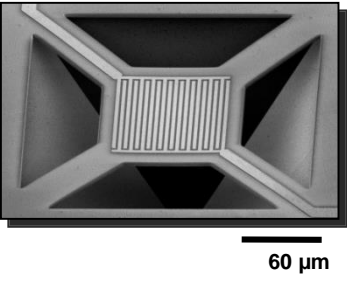


Correct Classification



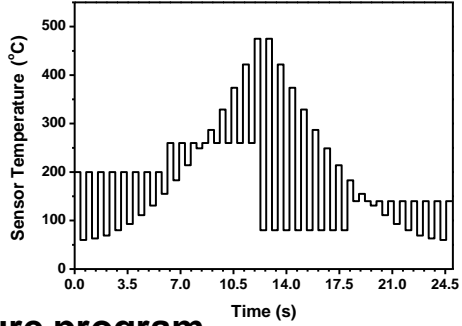
- good recognition of specific compounds
- chemical classification of (untrained) unknowns
- method automatically avoids data regions where significant drift occurs

Adaptability and Tunability with Common MEMS Platform

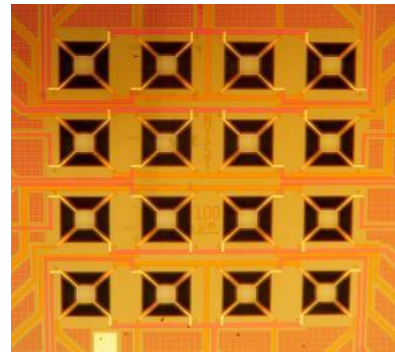


MEMS microhotplate

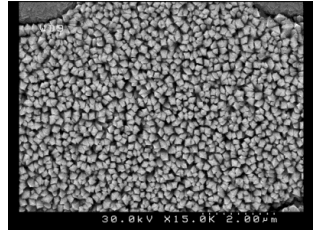
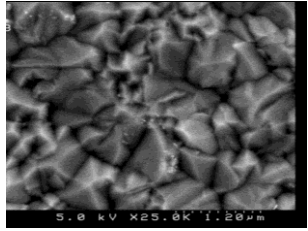
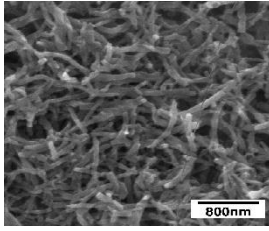
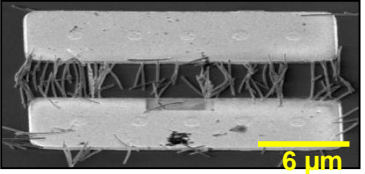
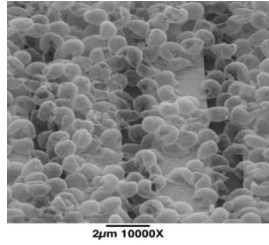
common platform e-nose for application-specific tuning



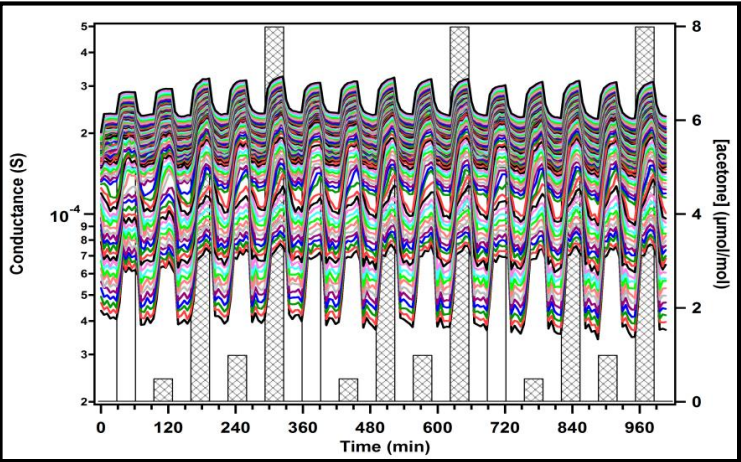
temperature program



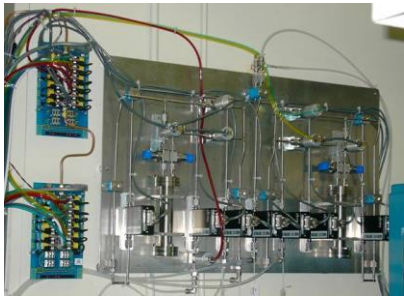
array platform with individually addressable elements



nanomaterials with varied compositions and morphologies

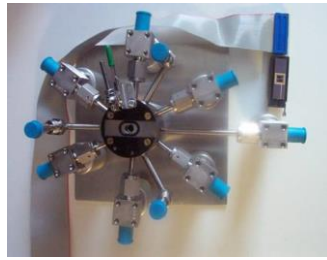


rich data streams



testing facilities

TICs
phosphonates
volatile organics
breath biomarkers



-
-
-

Some Chemicals and Interferences Tested with NIST Conductometric Microsensor Technology

Volatile Organic Compounds

Acetone	C_3H_6O
Benzene	C_6H_6
Carbon Monoxide	CO
Carbon Tetrachloride	CCl_4
Chloroheptane	$C_7H_{15}Cl$
Dichloromethane	CH_2Cl_2
Ethanol	CH_3CH_2OH
Hydrogen	H_2
Methanol	CH_3OH
Naphthalene	$C_{10}H_8$
Trichloroethylene	$CHClCCl_2$

and others

Chemical Warfare Simulants & Agents

Vap.Press @25°C(torr)

CEES	C_4H_9ClS	3.4
DFP	$C_6H_{14}FO_3P$	0.579
DMMP	$C_3H_9O_3P$	1.2
GA	$C_5H_{11}N_2O_2P$	0.07
GB	$C_4H_{10}FO_2P$	2.9
HD	$C_4H_8Cl_2S$	0.11

Toxic Industrial Chemicals

Acrylonitrile	C_3H_3N	107.8
Ammonia	NH_3	>760
Arsine	AsH_3	>760
Formaldehyde	CH_2O	>760
Hydrogen Cyanide	HCN	≈760
Methyl Isocyanate	C_2H_3NO	348
Parathion	$C_{10}H_{14}NO_5PS$	7.4

and others

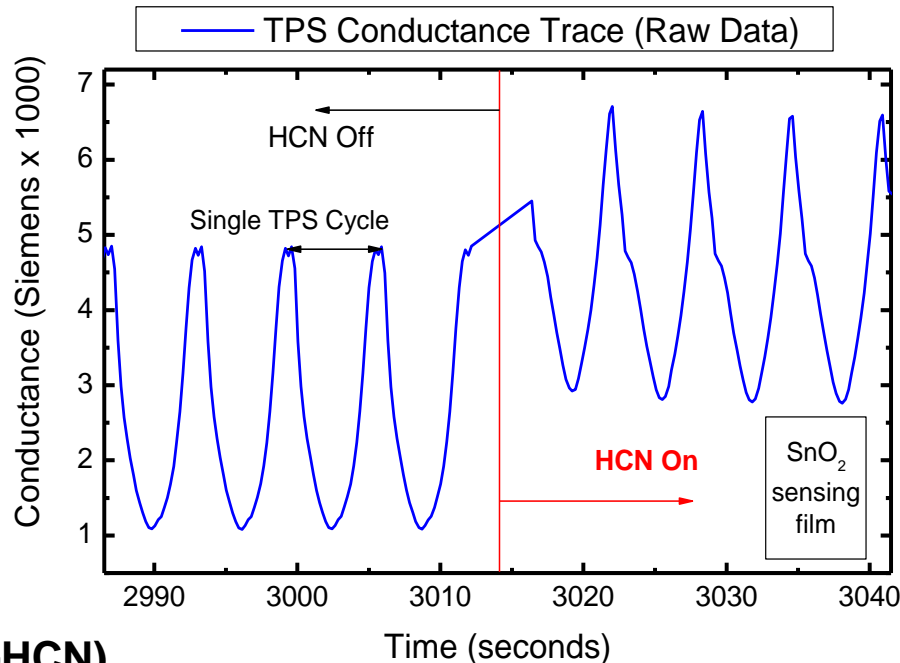
Detection at ppm concentrations has been demonstrated for *ALL* analytes; ppb (and high ppt) sensitivity has been demonstrated for some compounds.

Detecting Chemical Hazards in Challenging Backgrounds

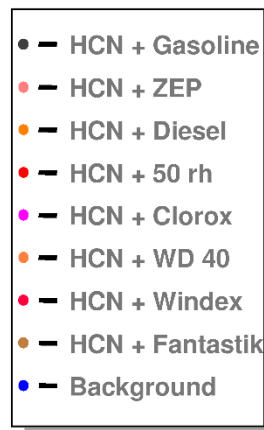
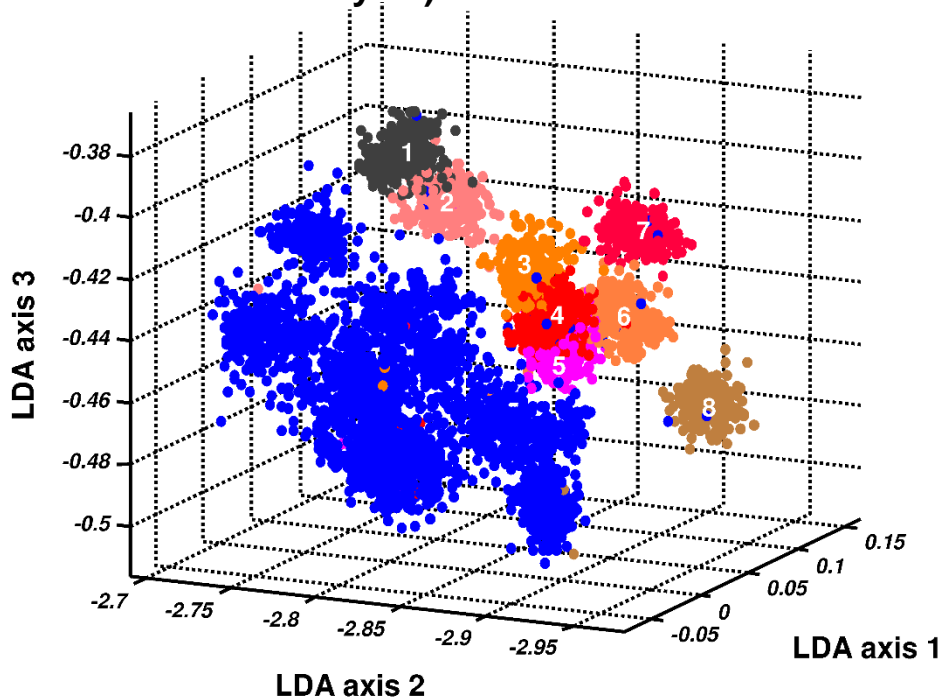
Raw Data (HCN - 1 condition)

50 ppm HCN, 50% RH

data from single T-programmed
SnO₂ element in array (32 T's)



Fully-Processed Data (all HCN and non-HCN) (Linear Discriminant Analysis)

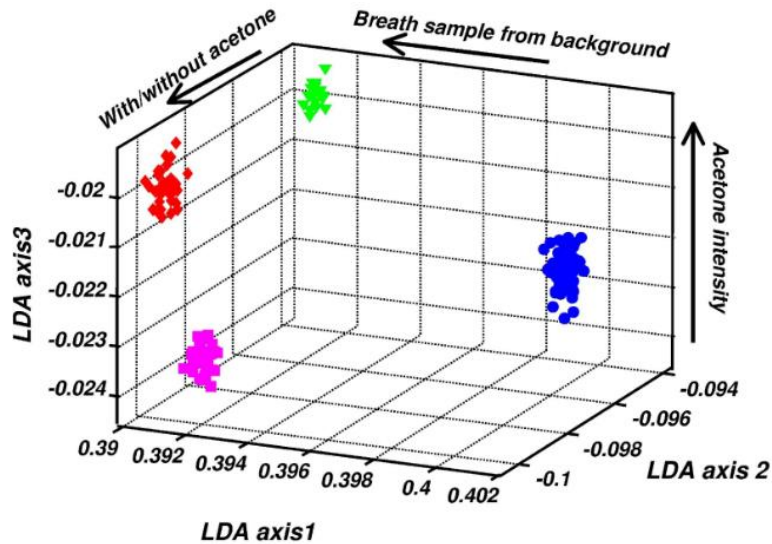


Sensors and Actuators B 137, 617–629
Annual Rev. Anal. Chem. 2:463–84

Visualization of Detection Capability

Preliminary Efforts: Medical Breath Analysis

acetone (*diabetes*) in synthetic breath



- Background air
- ▼ Exhaled breath - clean
- Exhaled breath - acetone low (1 ppm)
- ◆ Exhaled breath - acetone high (5 ppm)



IEEE Sensors Journal 10, 137-144

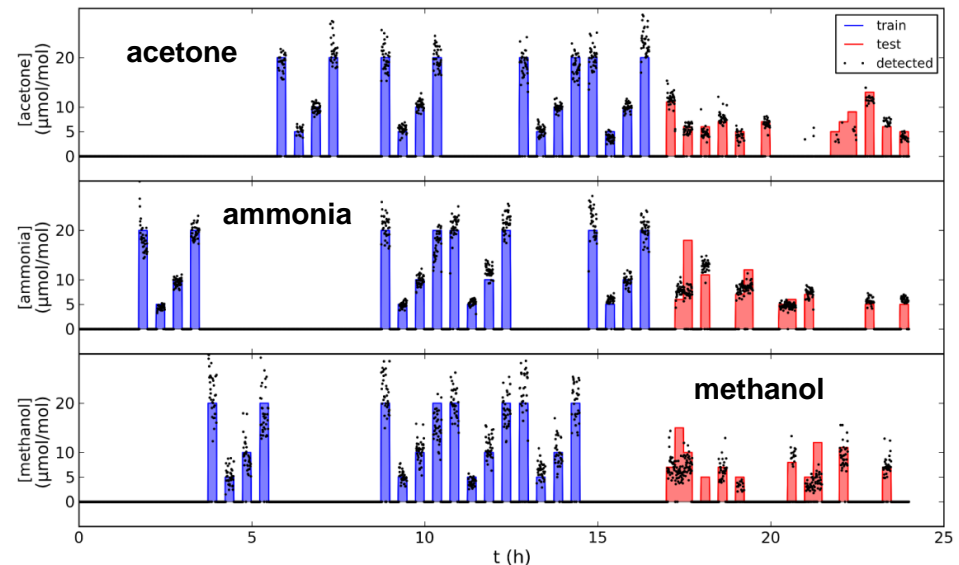
SnO₂ film + Sb:SnO₂ microshell film in TPS mode

multielement TPS with training and testing for 0, 1, 2 or 3 target analytes (*biomarkers for diabetes, renal issues*) in synthetic breath

[3 materials, 18 ramp T, 2 base T]

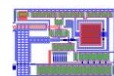
Rogers, Benkstein, Semancik, Analytical Chem. 84, 9774

(*machine learning*)

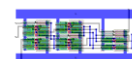


Other Enabling Technical Features

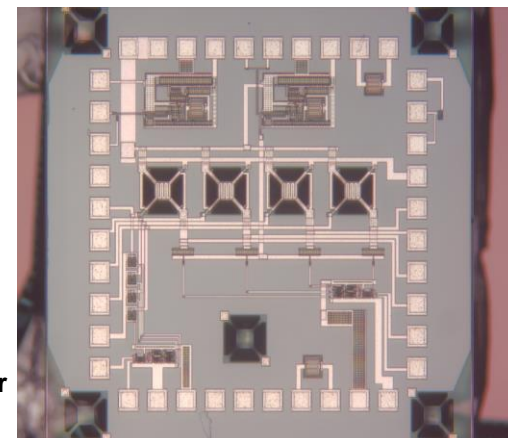
- **Efficient Materials Studies/Optimization**
combi-array processing/performance evaluations
- **Integration of Support Functions**
CMOS-based on chip electronics; coupled microfluidics
- **Reliability/Longevity**
improved materials & contacts, “delayed activation”
replacement within deployed devices



op-amp



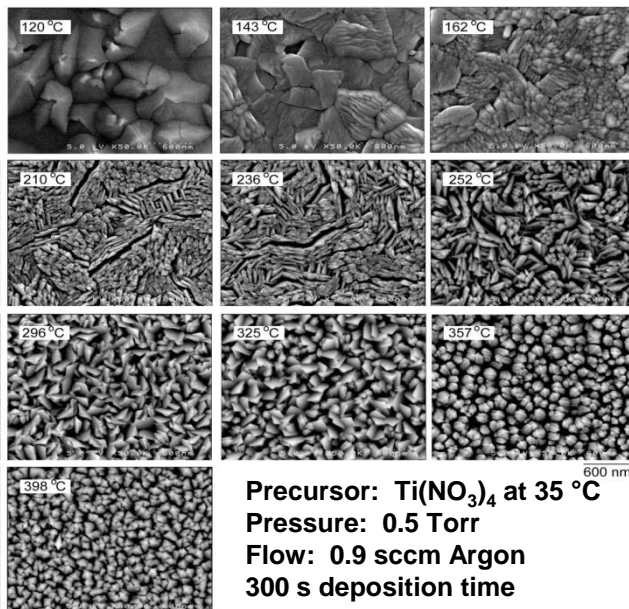
multiplexer



Monolithic Devices & Electronics

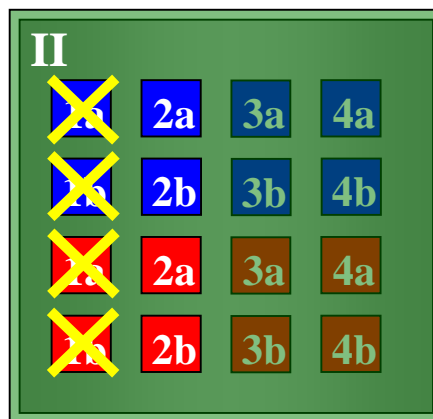
Array-Based Materials Studies

No deposition
(T too low)

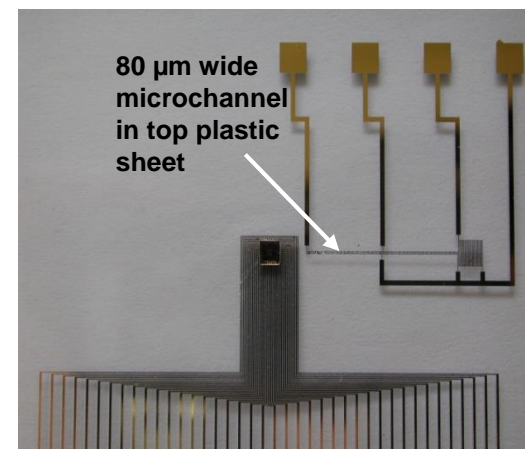


Precursor: $\text{Ti}(\text{NO}_3)_4$ at 35 °C
Pressure: 0.5 Torr
Flow: 0.9 sccm Argon
300 s deposition time

“Delayed Activation”



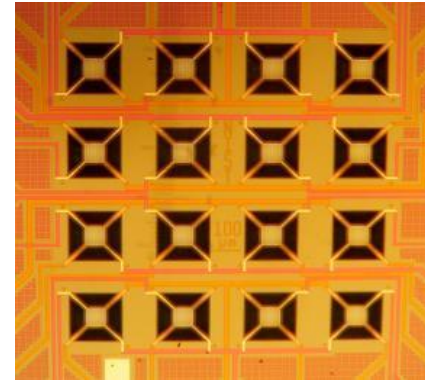
Microanalytical Systems



Outline

- **Chemiresistor Array**

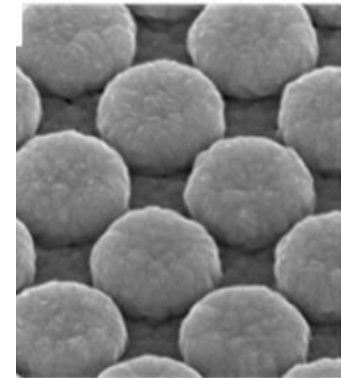
application-adaptable gas sensing



100 μm
elements

- **Plasmonic Optical Sensor Platform**

gas-phase and condensed-phase sensing

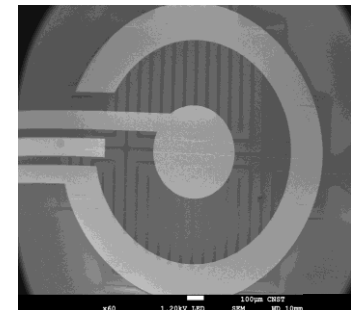
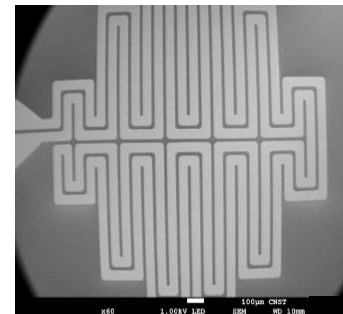


380 nm
features

- **Microscale Electrochemical Device**

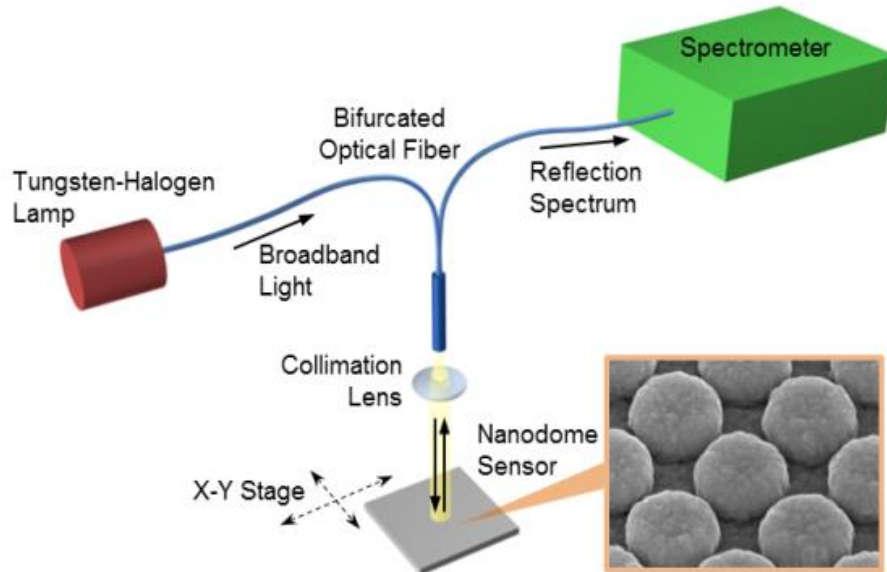
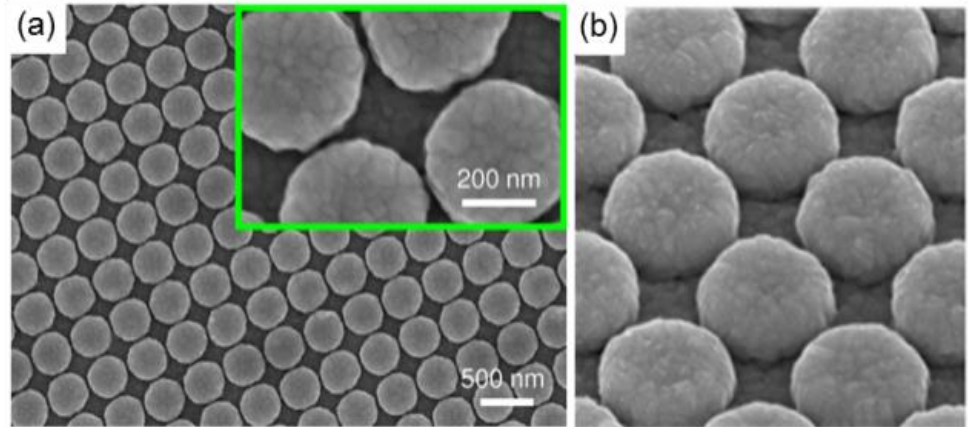
biochemical characterization

DNA on
500 μm
working
electrode



Nanoengineered Materials for Photonic Sensing

Ag Nanodome Arrays (380 nm diameters, 15-20 nm gaps)



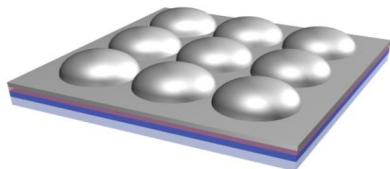
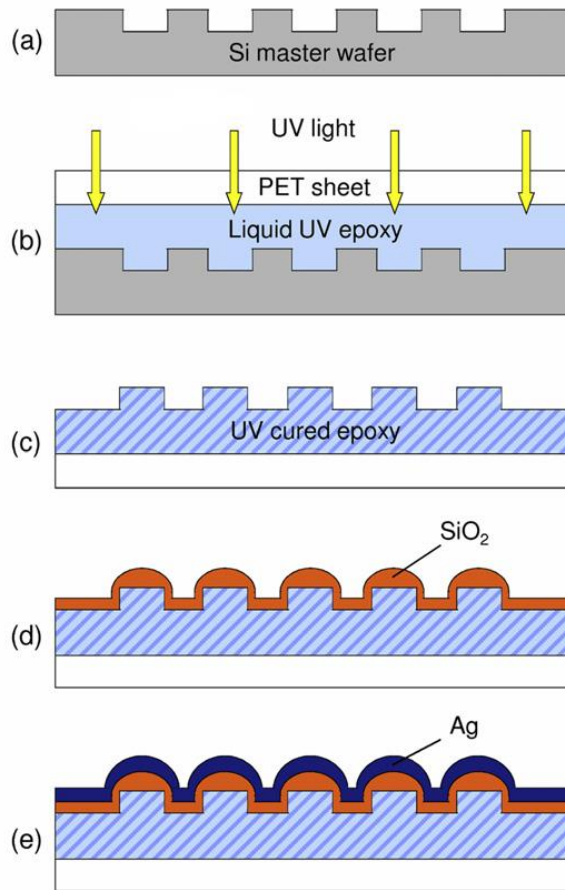
Sensing Signals

Surface-Enhanced Raman Scattering (SERS)

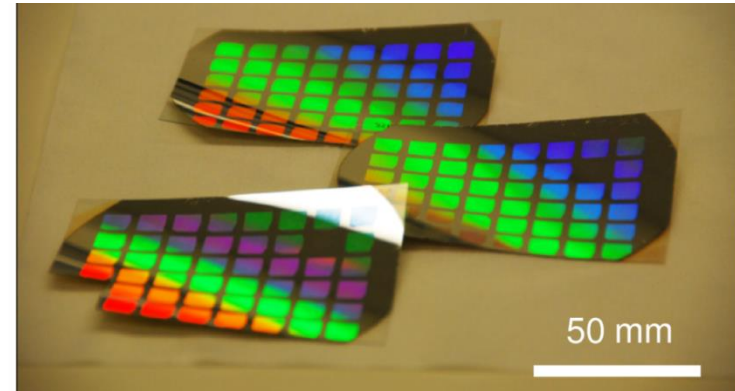
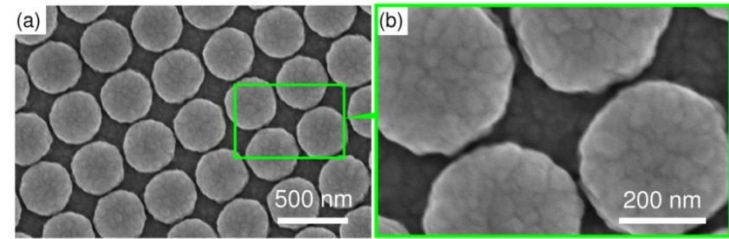
Localized Surface Plasmon Resonance (LSPR) – capture affinity

Nanodome Array Fabrication

Nanoreplica Molding Process



Fabrication at NIST NanoFab



array on flexible polyethylene terephthalate substrate (PET)

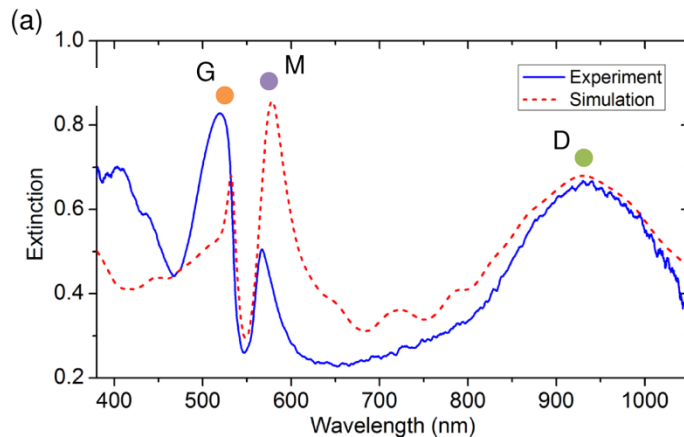
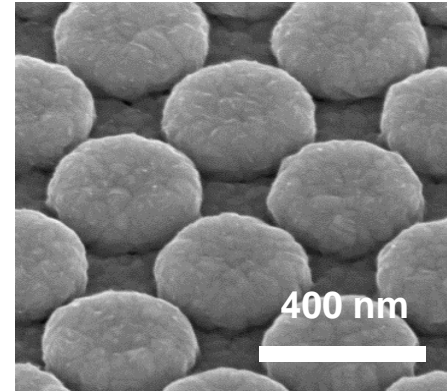
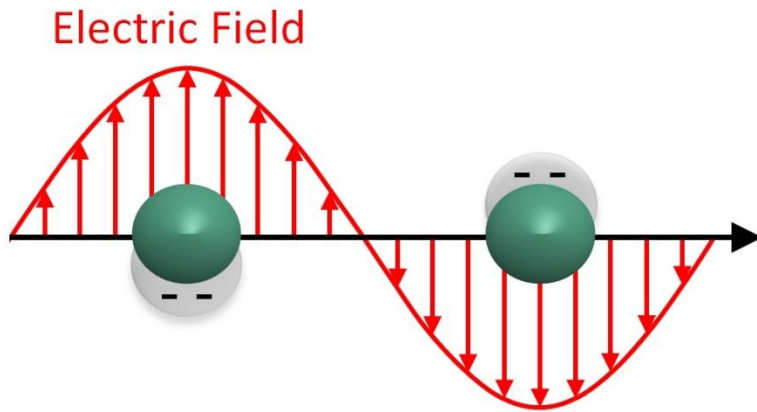
- Room temperature
- Low-force molding
- Plastic substrate



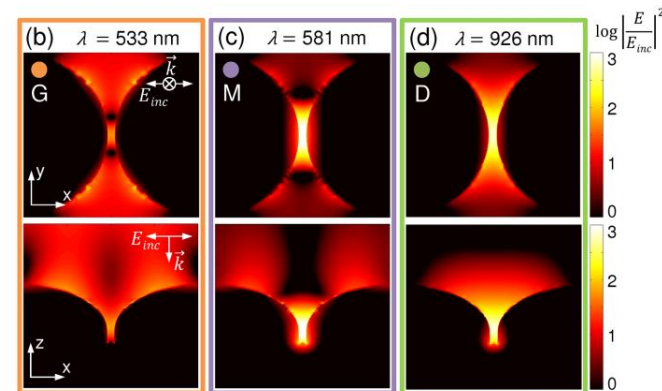
- Low-cost/disposable
- Large-area fabrication
- Uniform, reproducible structure
- Simple to control gap dimension

Localized Surface Plasmon Resonance (LSPR)

Plasmonic "Nanodome" Array



Modes
 Grating Diffraction (G)
 Multipole (M)
 Dipole (D)

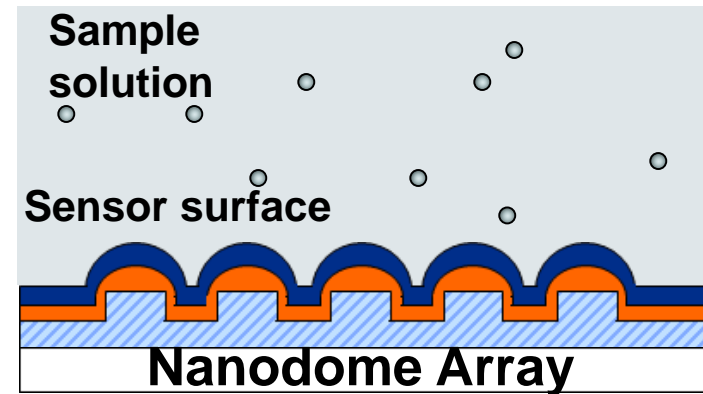
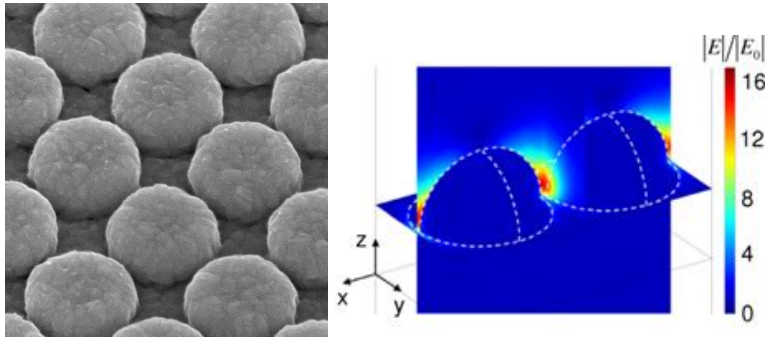


Finite Difference Time Domain (FDTD)
 modeling of photonic modes
 (spatial/size effects)

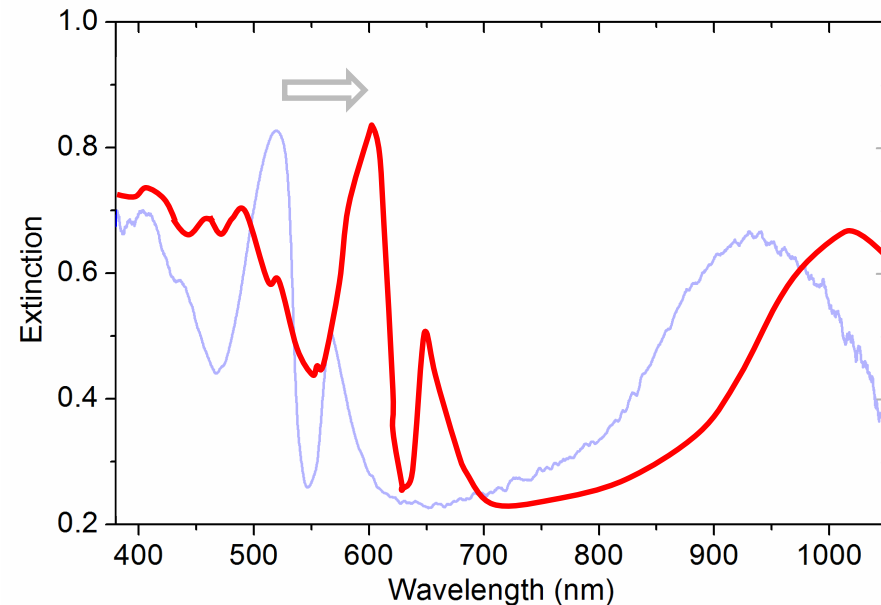
Label-Free Capture Affinity Biosensor

Plasmonic Nanodome Array

- Biomolecular attachment to sensor surface



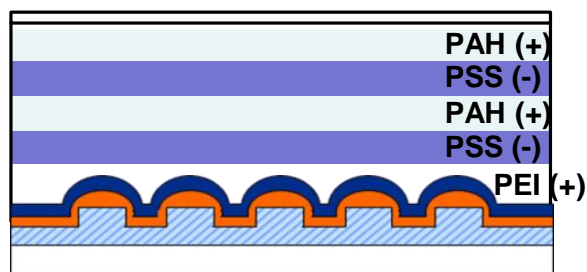
induces shift in
resonance
of localized modes



Surface Sensitivity

stacking alternate-charge 5 nm layers to explore surface range and compare to theory

Charged polyelectrolyte layers



Nanodome Sensor

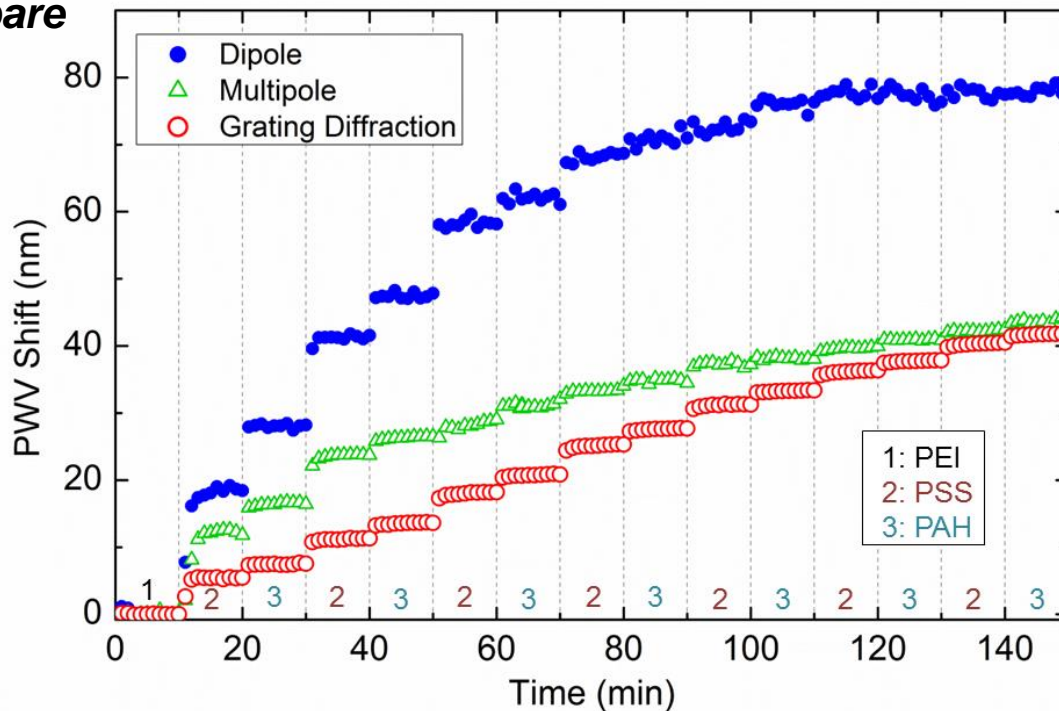
PEI: poly(ethylenimine)

PSS: poly(sodium 4-styrenesulfonate)

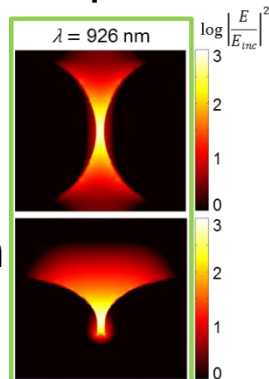
PAH: poly(allylamine hydrochloride)

Finite Difference Time Domain modeling of photonic modes

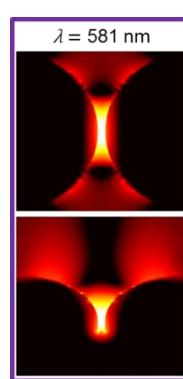
~ 60 nm



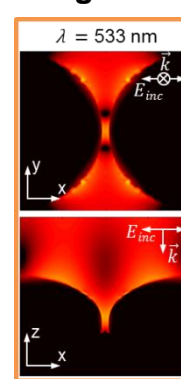
Dipole



Multipole



Grating Diffraction

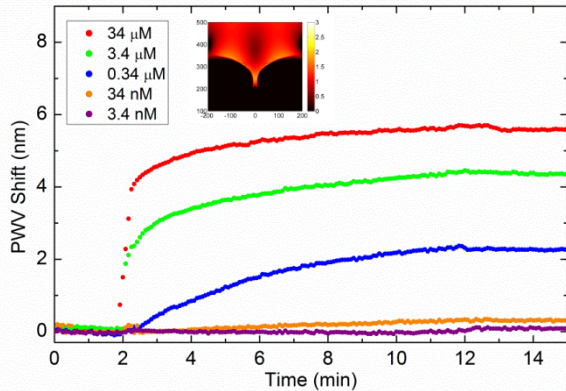


> 75 nm

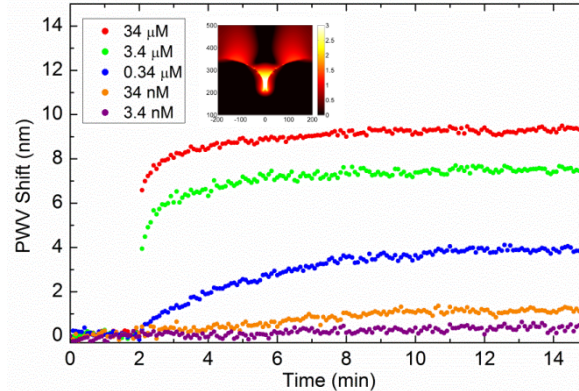
Protein – Protein Binding Assay

Protein A (0.5 mg/mL) adsorbed on nanodome array and exposed to IgG

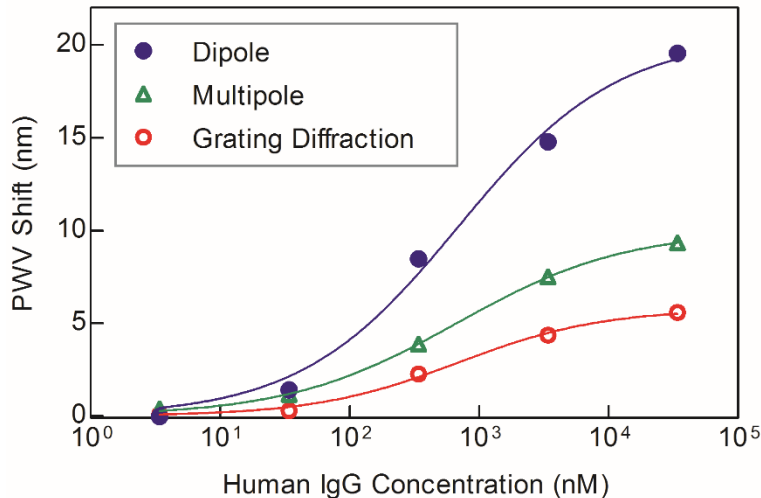
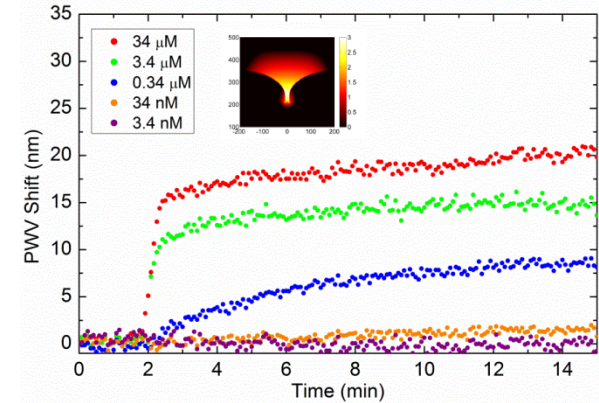
Grating Diffraction



Multipole



Dipole

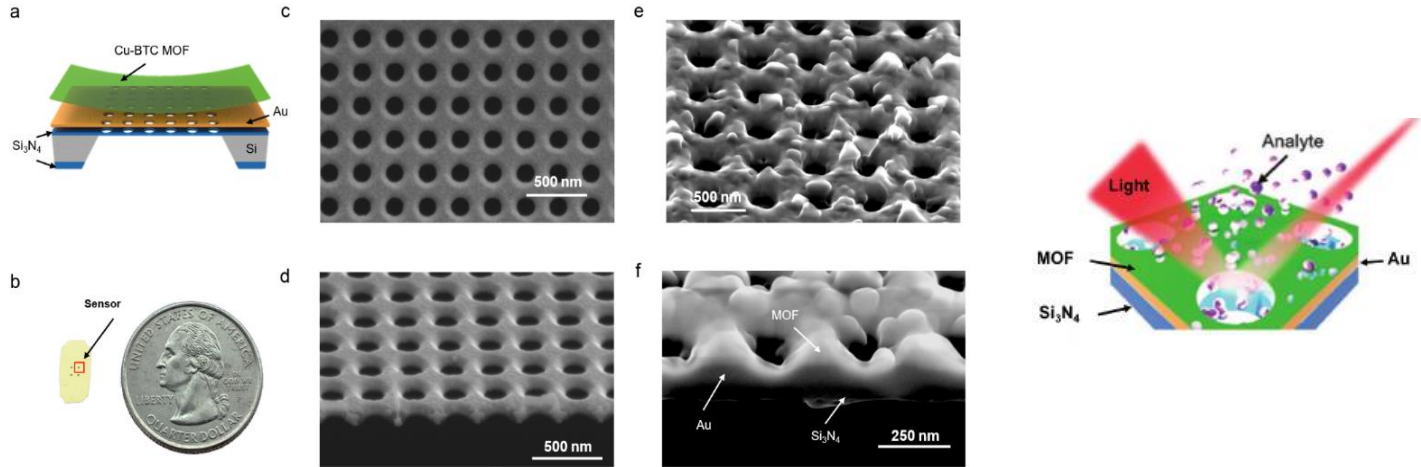


Resonance Mode	K_D (μM)	Detection Limit (nM)
Dipole	0.693	0.131
Multipole	0.672	0.767
Grating Diffraction	0.690	1.69

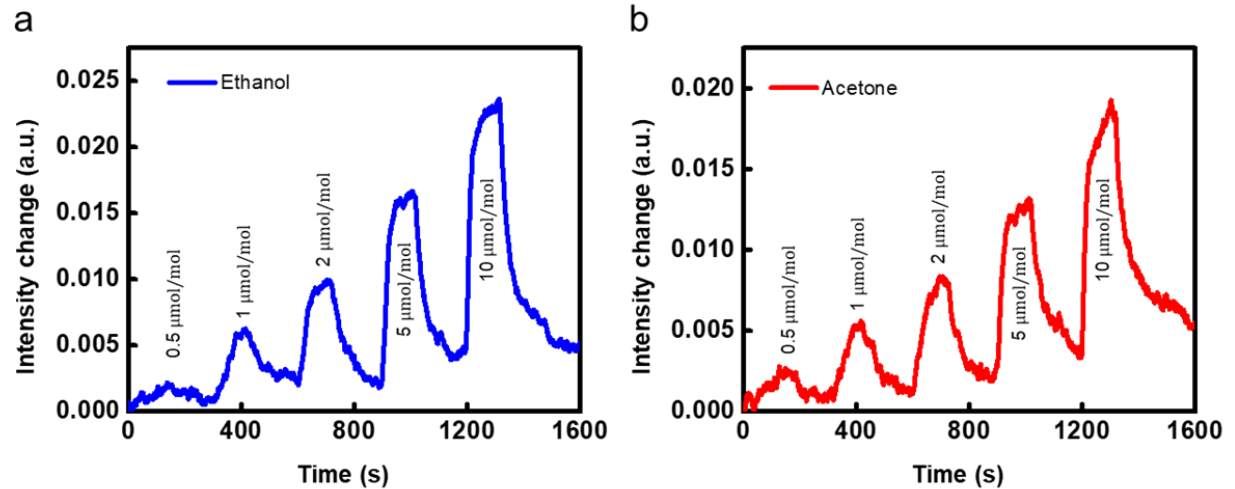
$K_D \sim 0.53 \mu\text{M}$ measured previously
 Lu *et al. Appl. Phys. Lett.* 93, 111113

Recent Plasmonics Studies: Gas Sensing with Holes and a Camera

Au nanohole array (NHA) coated with MOF (metal-organic framework)



detection of VOCs at MOF-nanohole interfaces

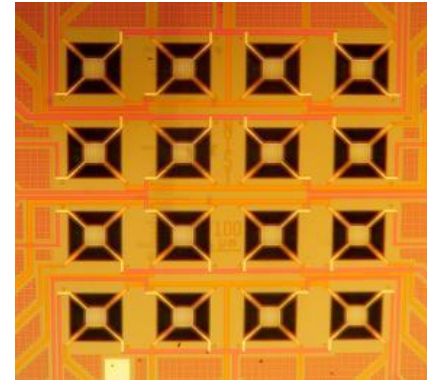


(with George Washington Univ.)

Outline

- **Chemiresistor Array**

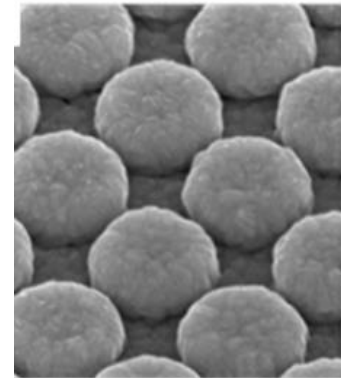
application-adaptable gas sensing



100 μm
elements

- **Plasmonic Optical Sensor Platform**

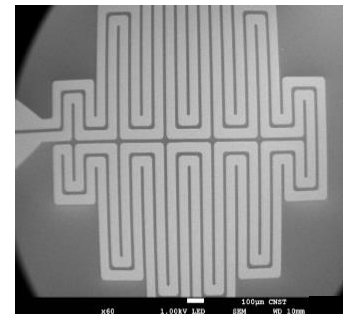
gas-phase and condensed-phase sensing



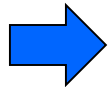
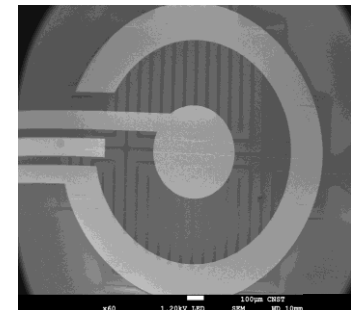
380 nm
features

- **Microscale Electrochemical Device**

biochemical characterization



DNA on
500 μm
working
electrode

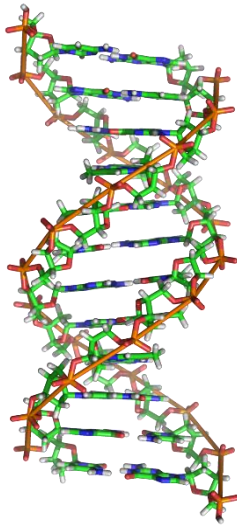


Electrochemical Biosensing

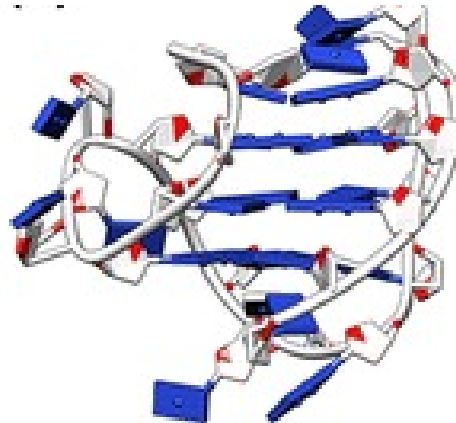
- Surface Immobilized biomolecules
- Microheater-Based T Control, but ~ 20 °C - 80 °C (not 20 °C – 480 °C)
- T_m Measurements as a Property Indicator (biomolecule stability or binding-based change in stability)

Drug discovery & quality control

Medical diagnostics



Duplex DNA



G-Quadruplex DNA

Electrochemical Sensing with Temperature Control

moving toward smaller-volume, electronic devices

Examples of prior electrochemical device research involving temperature:

- T-dependent hot wire electrochemistry

Flechsig, Peter, Hartwich, Wang, Grundler, *Langmuir* 21, 7848-7853

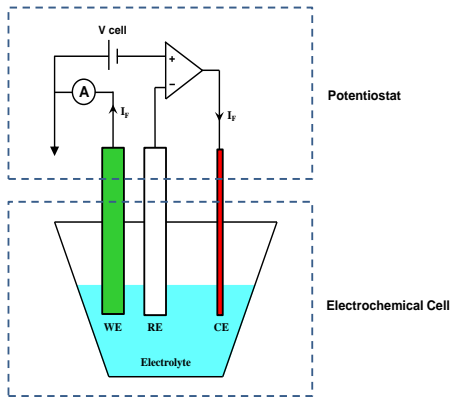
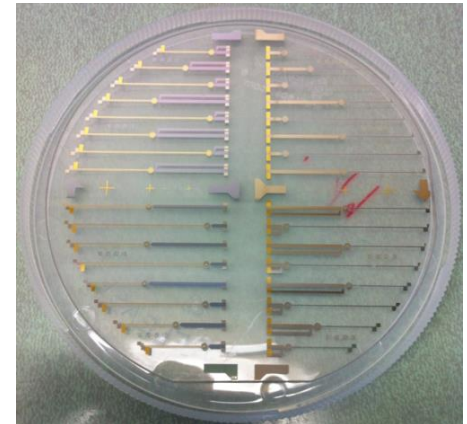
- Peltier heater-based electrochemical measurements

Yang, Hsieh, Patterson, Ferguson, Eisenstein, Plaxco, Soh *Angew. Chem. Int. Ed.* 53, 3163-3167

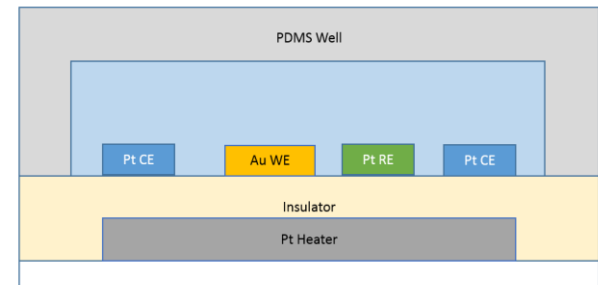
NIST

Present work:

- Wafer-based planar microscale fabrication
- Small-footprint elements - *small volume sample analyses*
- Integrated, localized heater (*dedicated to each element*)
- Potentially faster T sweeps and T programming
- Compatible with electronic microarray development

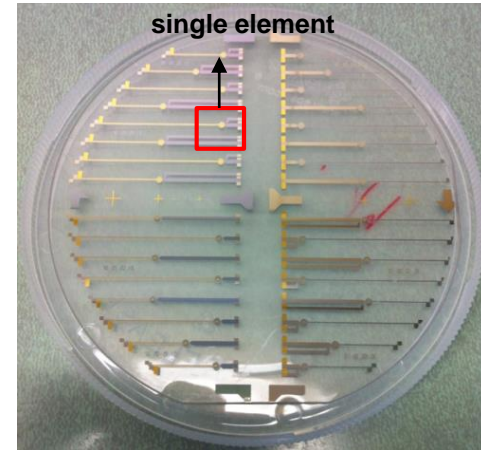
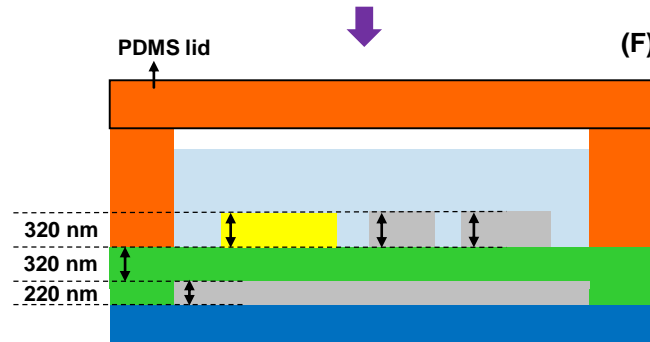
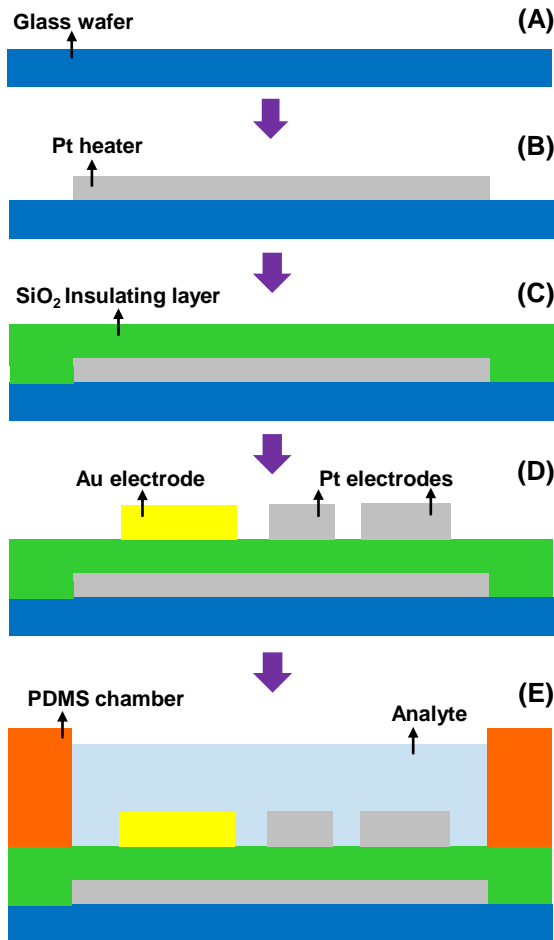


10 μ L (moving toward 100nL)



Miniature Temperature-Controlled Electrochemical Device

Wafer-Based Planar Fabrication 3-electrode platform with microheater



NIST NanoFab

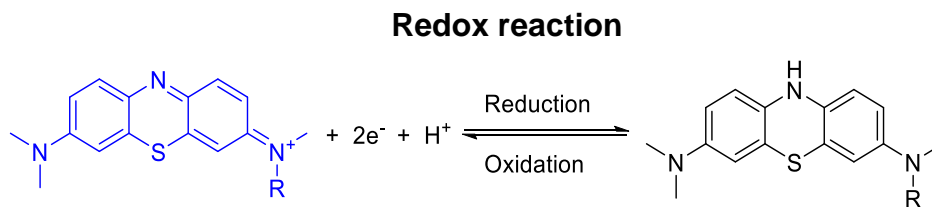
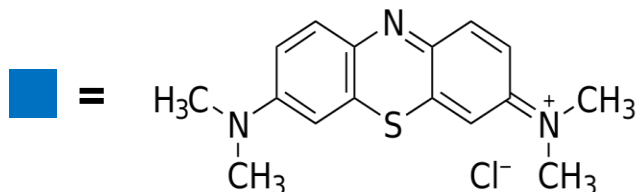


Sputter Cluster Deposition Tool
(Ion Milling Tool – not shown)

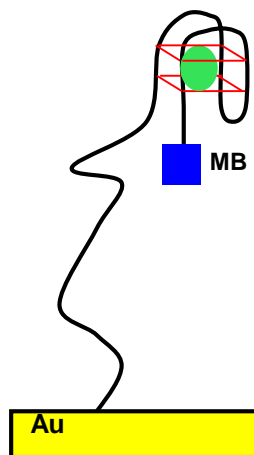
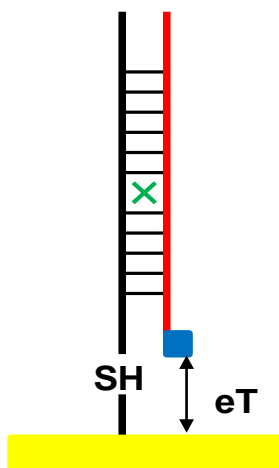
Feasibility for 2 Potential Application Areas

perform studies on hybridization and ligand binding interactions using microscale platforms and thiol-tethered & electroactively tagged biomolecular probes

methylene blue



- **Disease screening** - single nucleotide polymorphism (SNP) detection
- **Drug discovery** - detection of property changes associated with binding events for (small molecule) drugs



Monitor:

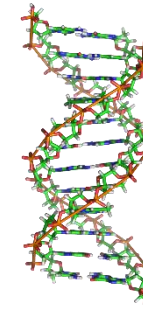
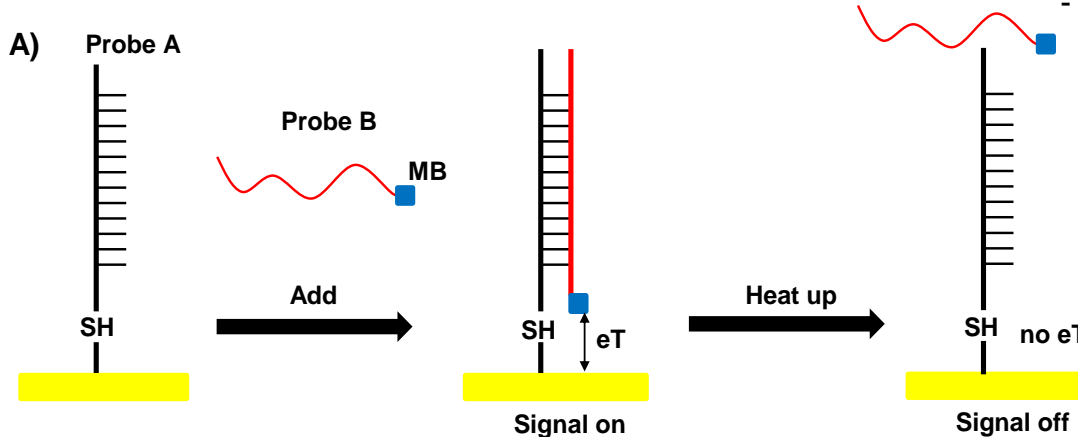
- binding-based electron transfer (hybridization)
- conformational/distance-based electron transfer
- thermally-induced dehybridization

Duplex Melting Curve Analysis

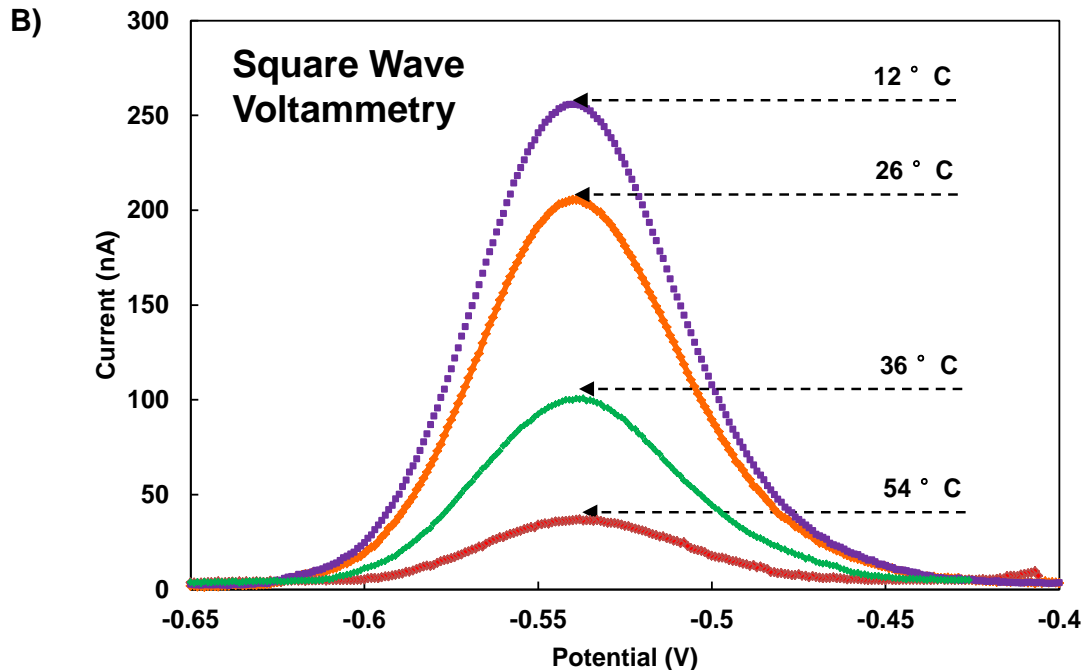
Feasibility: Model studies on DNA duplexes

Probe density must be viable:

- Dense enough to produce measurable signal
- Not so dense to create interaction hindrance



6-mercaptohexanol
backfill

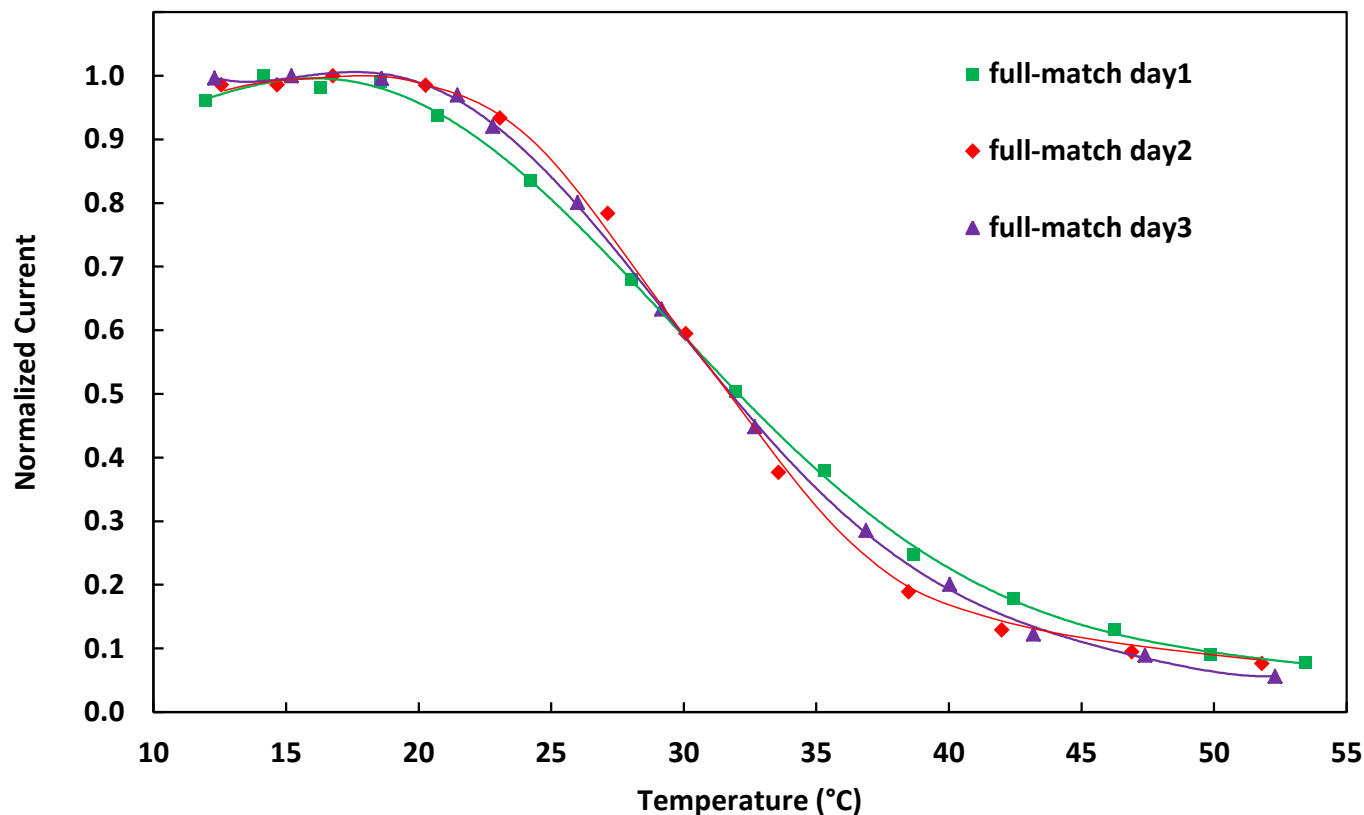
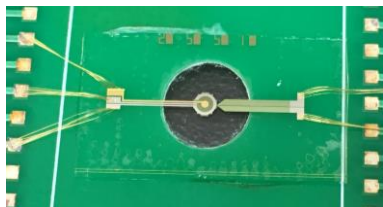


1. Increase the temperature by increasing the heater voltage.
2. Wait 5 seconds for equilibrium.
3. Take 15 seconds to take electrochemical measurement

Repeat step 1-3 to collect current at different temperature

The melting curve analysis can be done in ~ 15 mins using electrochemical detection

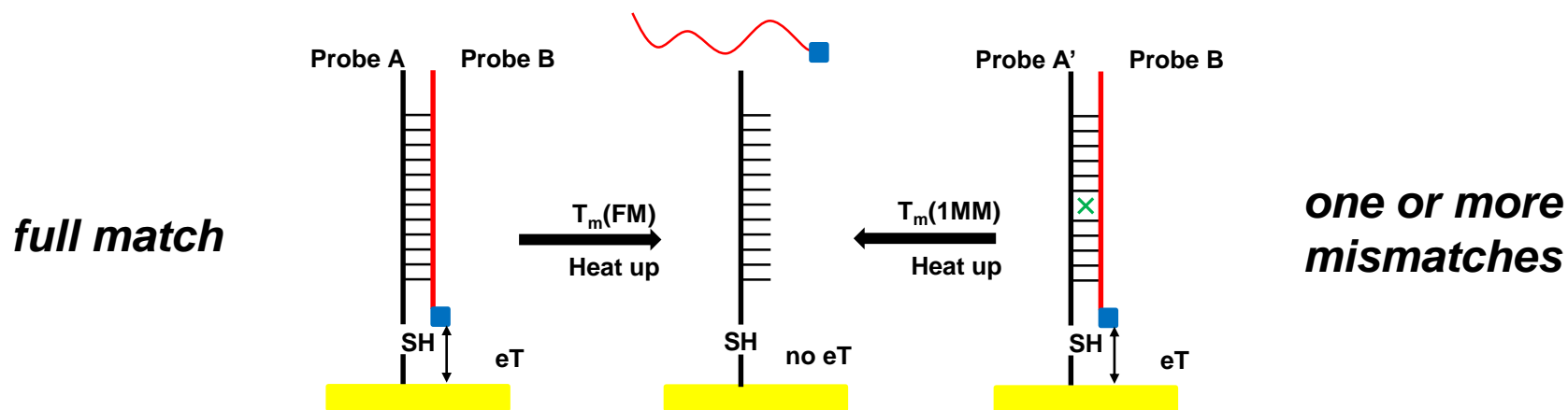
Reproducibility of Duplex Electrochemical Melting



- $T_m(\text{day1}) = 27.6 \text{ }^\circ\text{C}$, $T_m(\text{day2}) = 27.1 \text{ }^\circ\text{C}$ and $T_m(\text{day3}) = 28.0 \text{ }^\circ\text{C}$
- $T_m(\text{average}) = 27.6 \pm 0.4 \text{ }^\circ\text{C}$

T_m Detection of Single Nucleotide Polymorphism (SNP)

- SNP detection can be useful for predicting susceptibility to disease and drug metabolism
- Traditionally SNP detection employs optical melt-curve analysis
- Electrochemical measurements offer an electronic alternative



explore dehybridization with increased temperature for varied cases

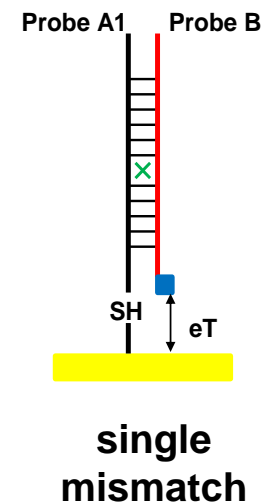
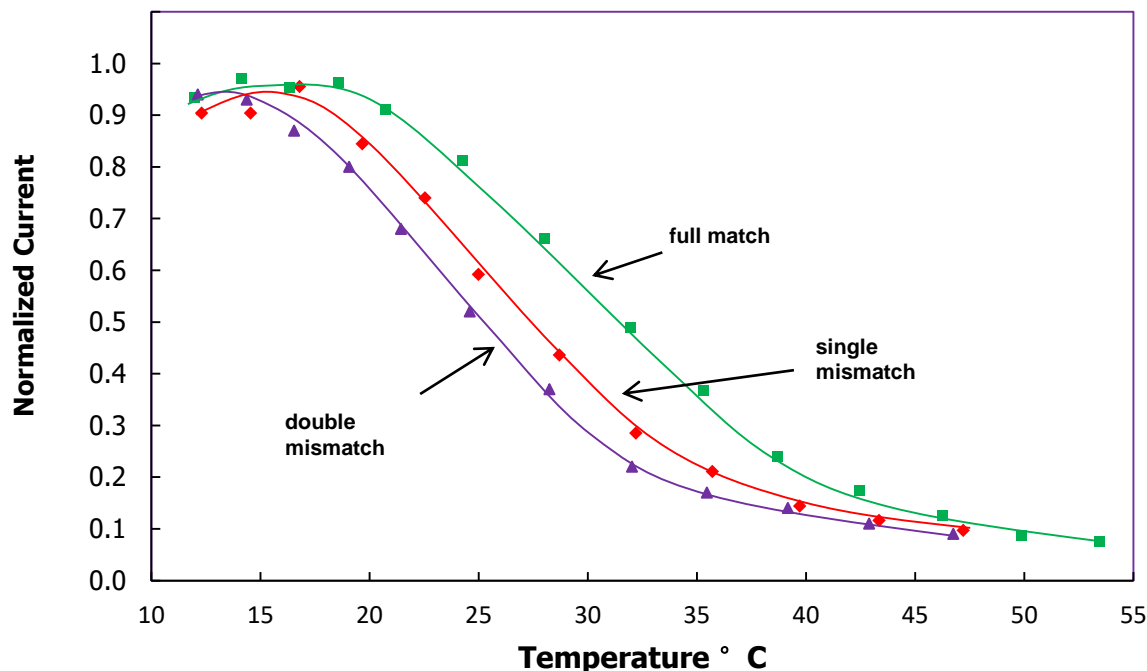
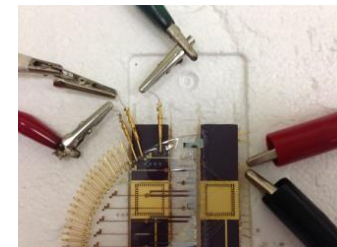
Single Nucleotide Polymorphism (SNP) Detection

Probe A (full match): 5'-SH-C6H12-**TTT ACC TTT ATT** -3'

Probe A1 (1 mismatch): 5'-SH-C6H12-**TTT ACG** TTT ATT -3'

Probe A2 (2 mismatch): 5'-SH-C6H12-**TTT AGG** TTT ATT -3'

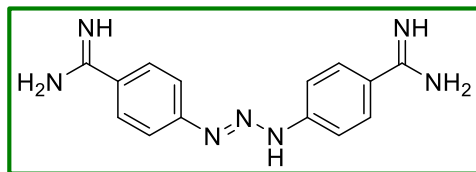
Probe B: 3'-**MB-AAA TGG AAA TAA CC**-5'



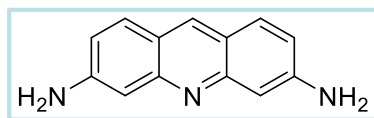
- $T_m(\text{FM}) = 27.6\text{ }^\circ\text{C}$, $T_m(\text{1MM}) = 22.5\text{ }^\circ\text{C}$ and $T_m(\text{2MM}) = 20.3\text{ }^\circ\text{C}$

Screening for Drug Stabilization Effects

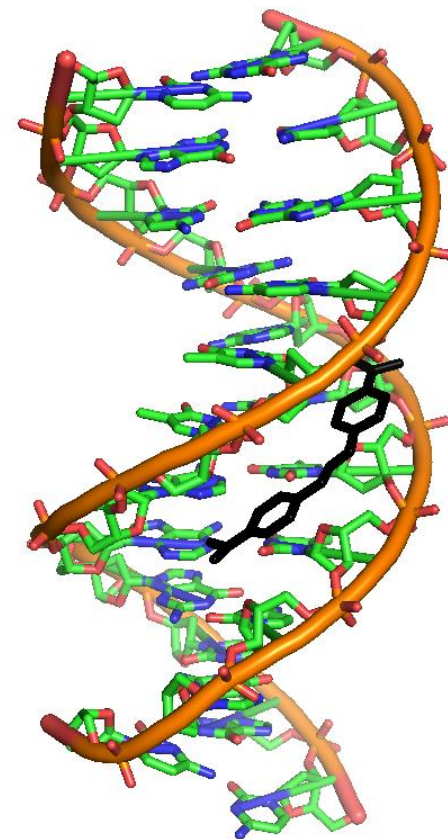
- Stability changes to detect *small molecule drug* binding
- Binding increases duplex DNA stability (higher T_m)



Diminazene aceturate (DMZ, Berenil)

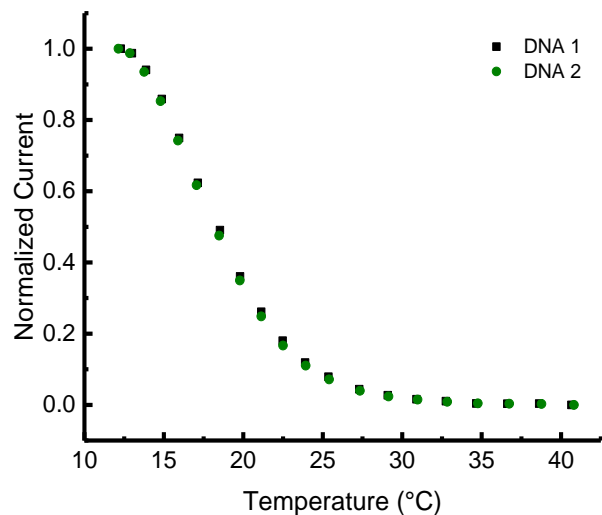


Proflavine

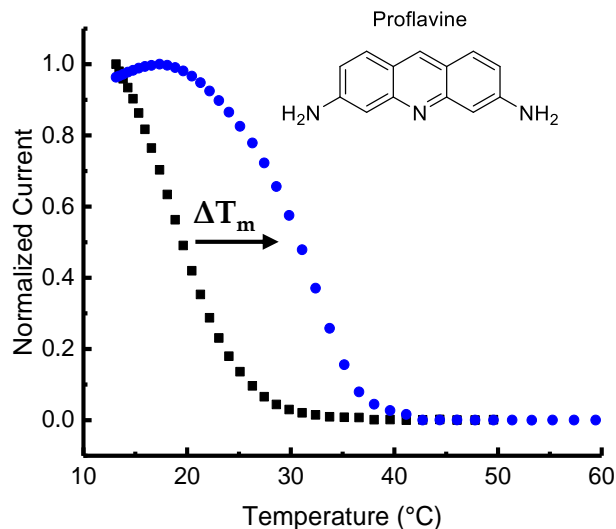


(with U MD and Purdue)

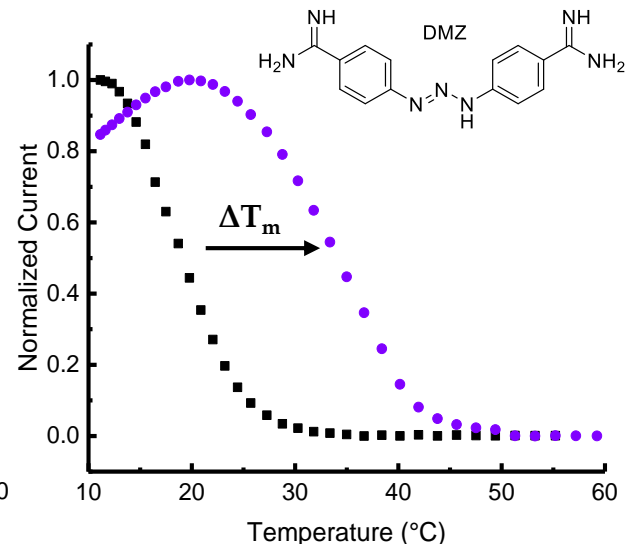
Ligand Stabilization of Duplex DNA



$$\Delta T_{m, \text{DNA}} = -0.3 \pm 0.4 \text{ } ^\circ\text{C}$$



$$\Delta T_{m, \text{proflavine}} = 11.8 \pm 0.3 \text{ } ^\circ\text{C}$$



$$\Delta T_{m, \text{DMZ}} = 14.6 \pm 0.6 \text{ } ^\circ\text{C}$$

Melting profiles in 10 μL of 2 $\mu\text{mol/L}$ cDNA and 13 $\mu\text{mol/L}$ ligand in 10 mmol/L PBS with 100 mmol/L NaCl pH 7.4.

(standard deviations given for three replicates)

Robinson, S. M.; Shen, Z.; Askim, J. R.; Montgomery, C. B.; Sintim, H. O.; Semancik, S. *Biosensors* 9, 54-67.

Conclusions

developing measurement concepts for microscale devices that report on chemistry/biochemistry

- **Multidisciplinary research**



- **Platform design and fabrication**
- **Nanomaterials**
- **Surface/interfacial phenomena**
- **Varied application areas**