

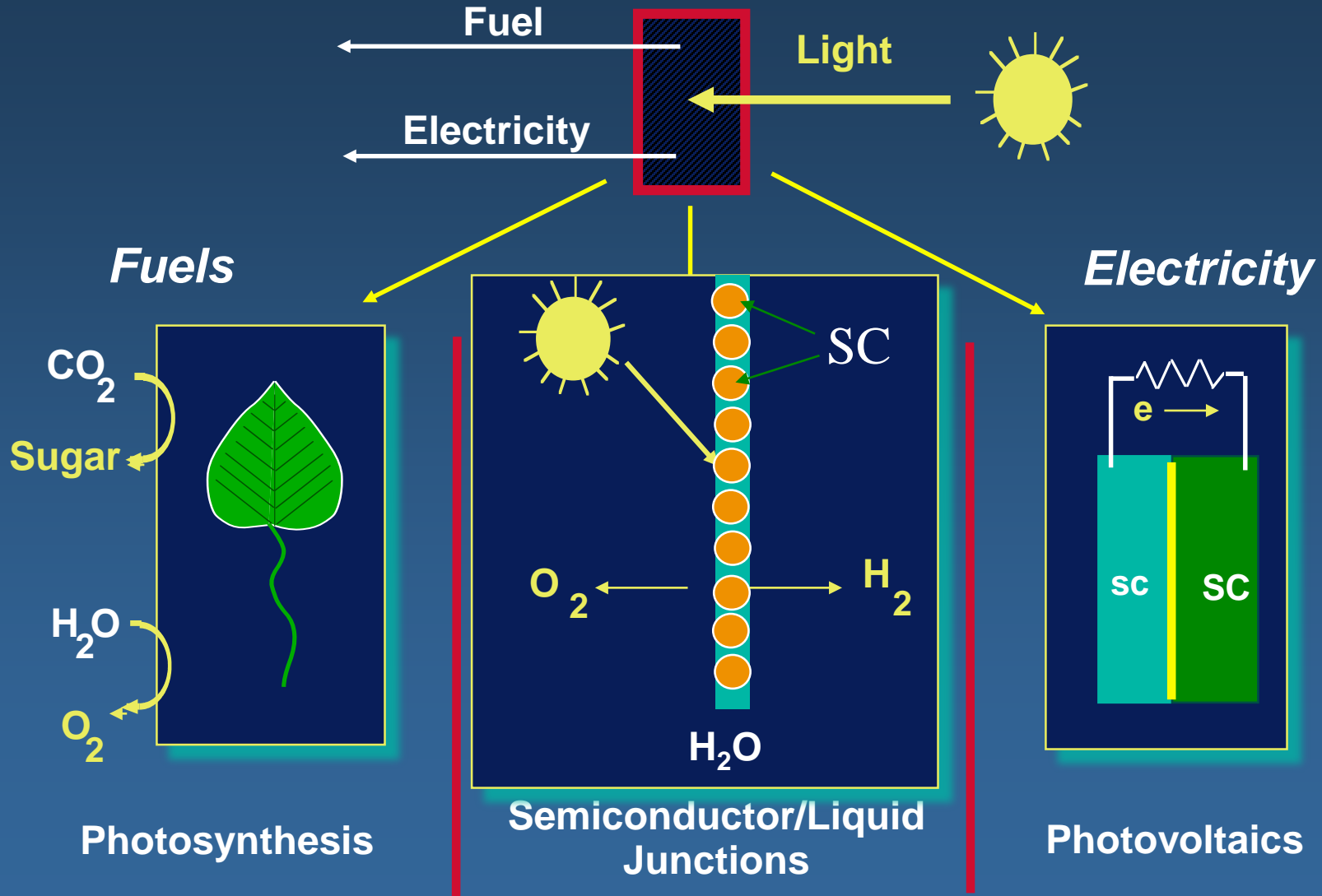
***ARTIFICIAL PHOTOSYNTHESIS:
DIRECT PRODUCTION OF FUELS
FROM SUNLIGHT***

(NO BUGS, NO WIRES)

Nathan S. Lewis

***Joint Center for Artificial Photosynthesis
California Institute of Technology
Pasadena, CA 91125***

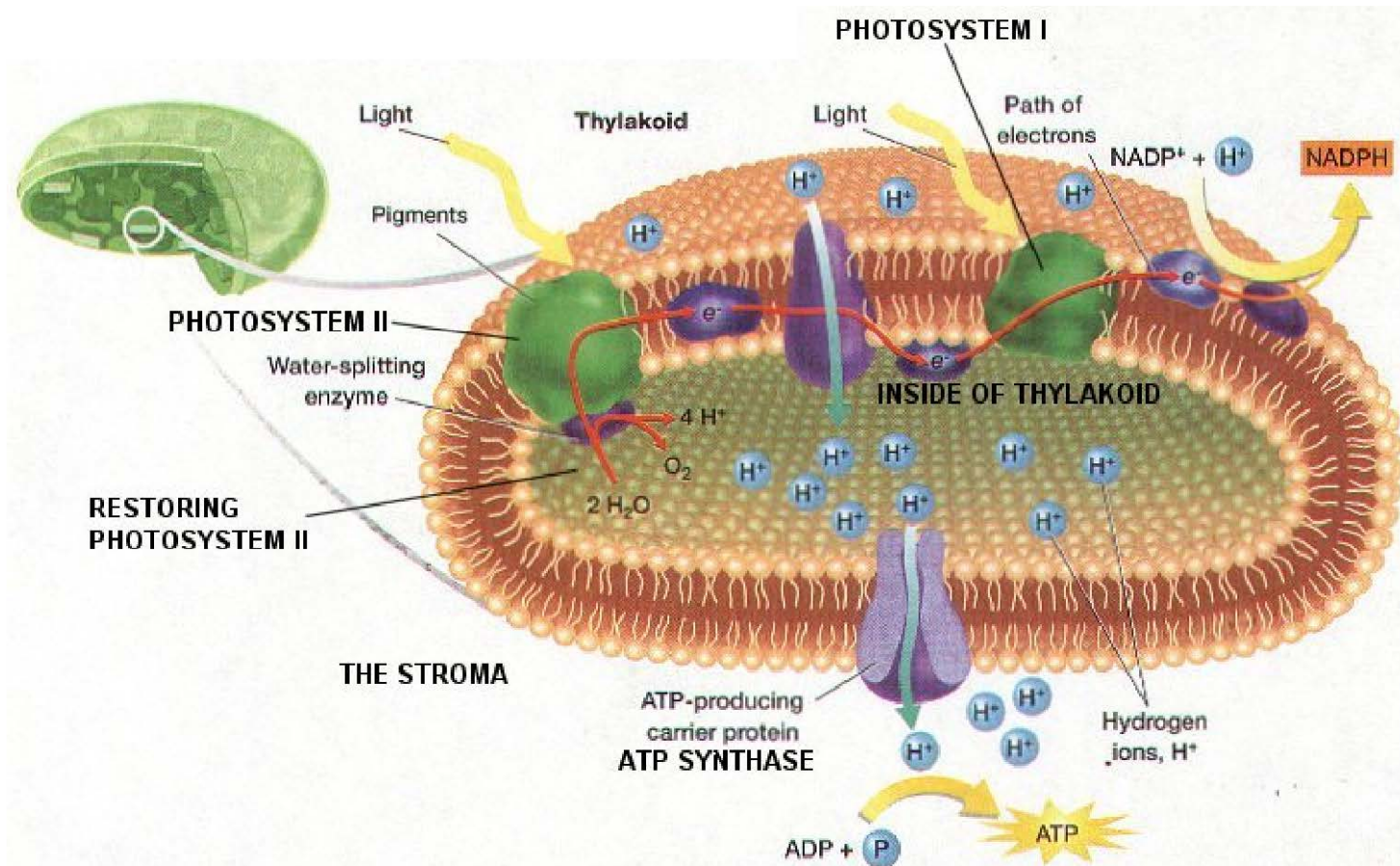
Energy Conversion Strategies



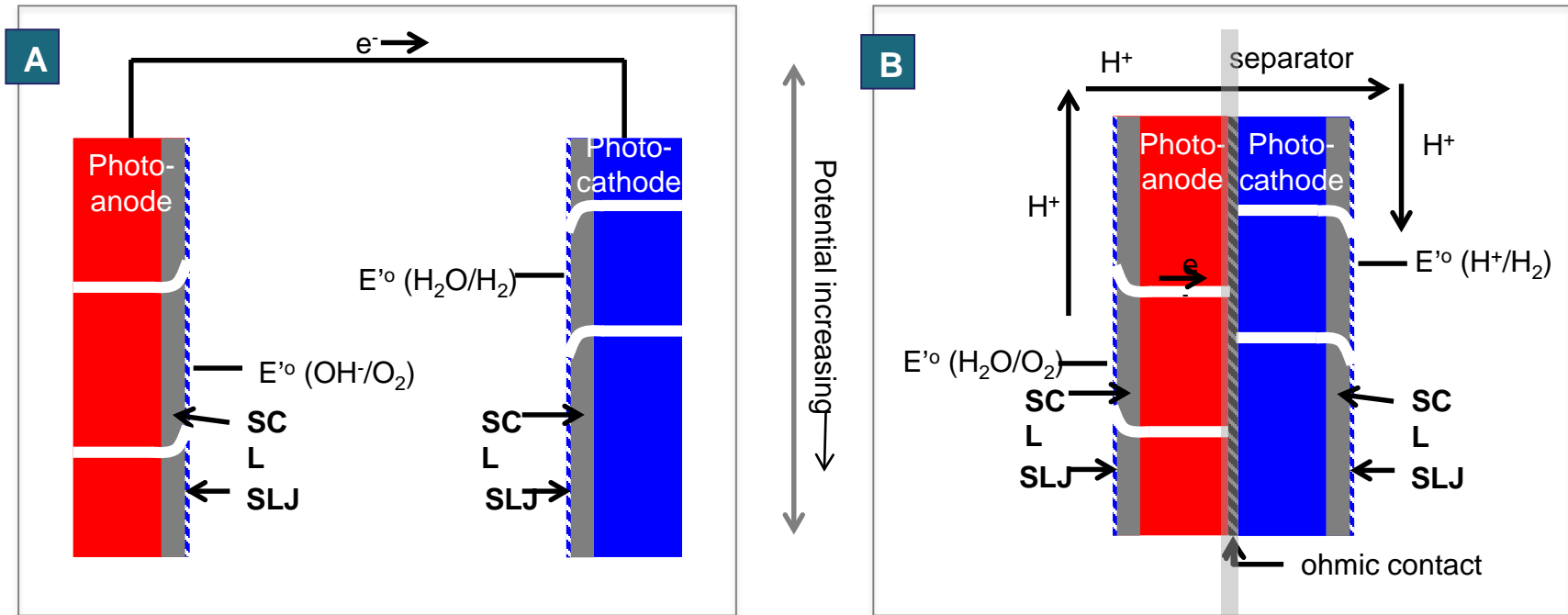
Fuel from Sunlight



Lessons from Photosynthesis

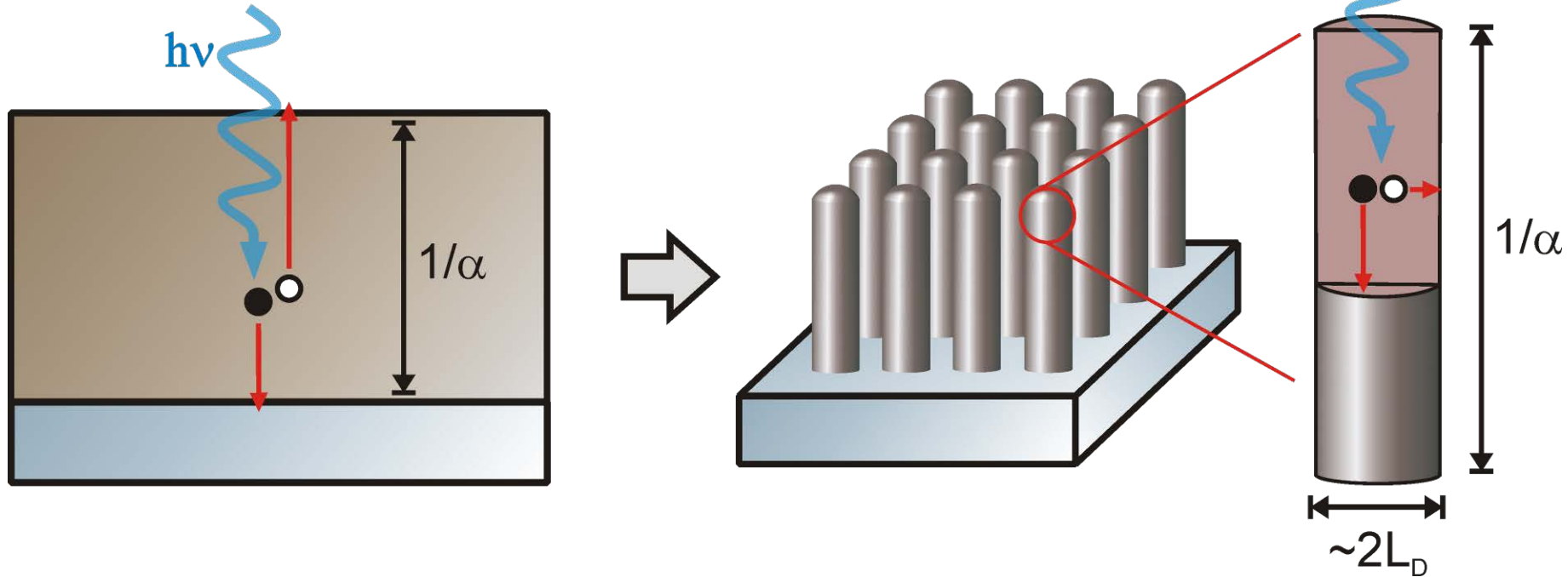


• DUAL AND TANDEM CELLS WITH SEMICONDUCTOR/LIQUID JUNCTIONS



- A) Dual photoelectrodes – Each photoelectrode provides a portion of the total photovoltage produced by the system and required for splitting water.
- B) Tandem photoelectrodes – The photoanode is in direct electrical contact with the photocathode. because the potential of the bottom of the conduction band of the photoelectrode is more positive than $E^\circ(\text{H}_2\text{O}/\text{H}_2)$, so photogenerated electrons can not produce the desired fuel (in this case H_2) without some additional energy input.

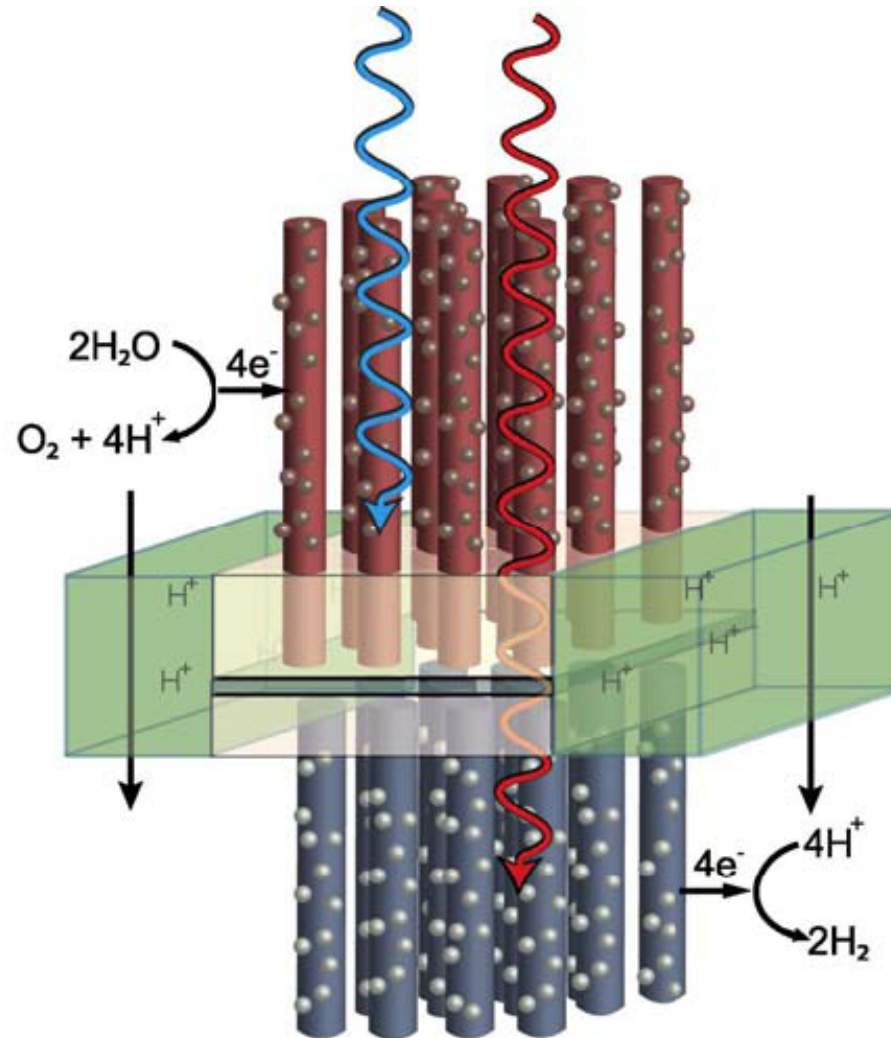
Structure – Radial Advantage



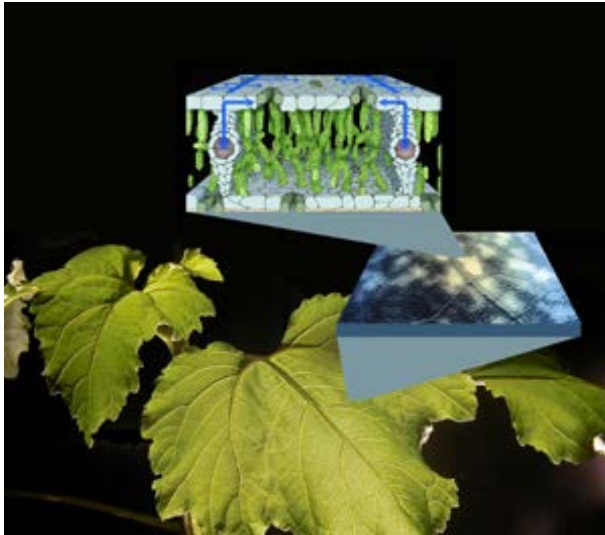
$L_D \propto \text{purity} \propto \text{materials cost}$

Impure material but high performance

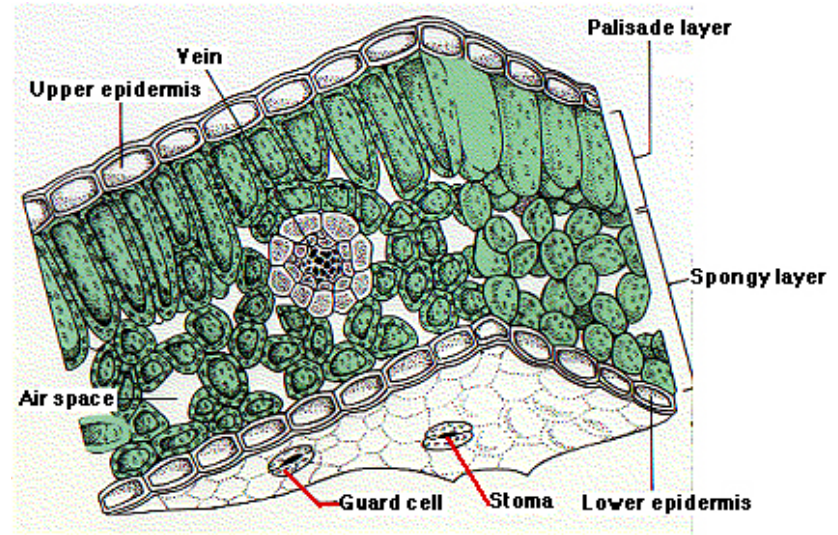
Constructing the Pieces of a Solar H_2 Fuel Generator



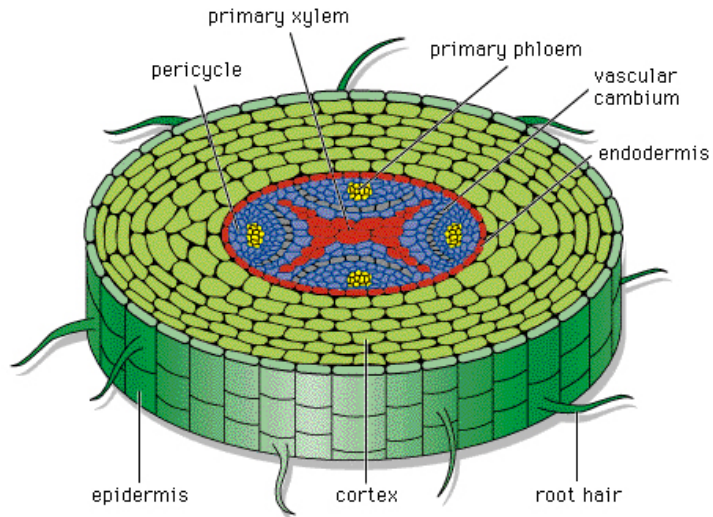
Structural Organization in Nature



Stomata



Leaf

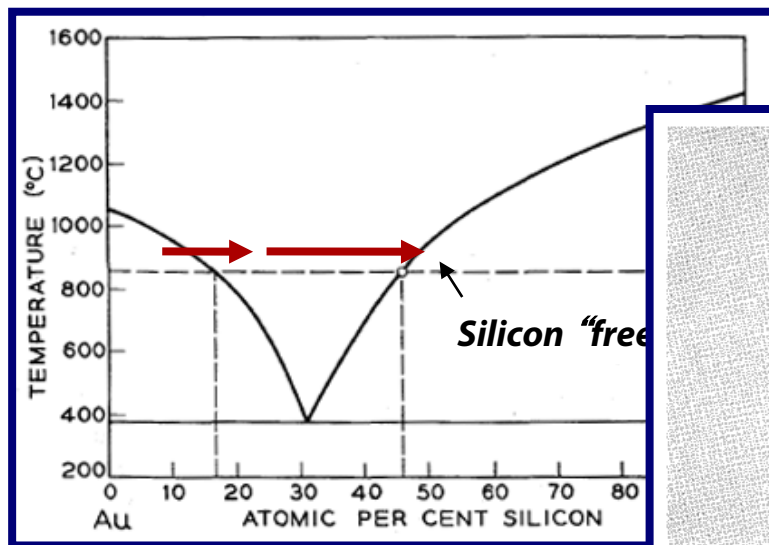
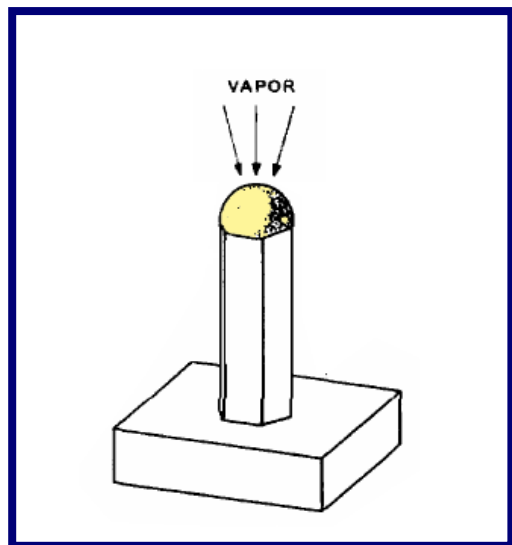


Root



Forest

Si Rods by Vapor-Liquid-Solid (VLS) Growth



- **Single crystals**
- **Growth direction controlled by substrate**
- **High growth rates (up to microns/s)**
- **Inexpensive gas phase precursors**

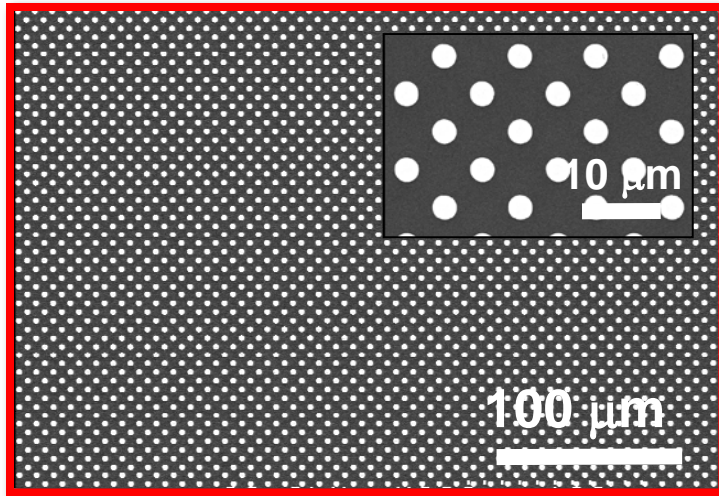
B. M. Kayes, M. A. Filler et al., *App. Phys. Lett.* **91**, 103110 (2007)

R. S. Wagner and W. C. Ellis, *App. Phys. Lett.* **4**, 89 (1964)

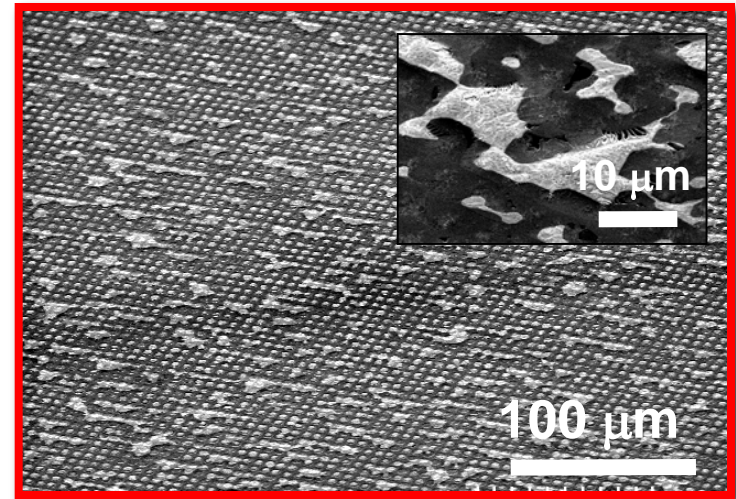
Oxide Buffer Layer - Pattern Fidelity



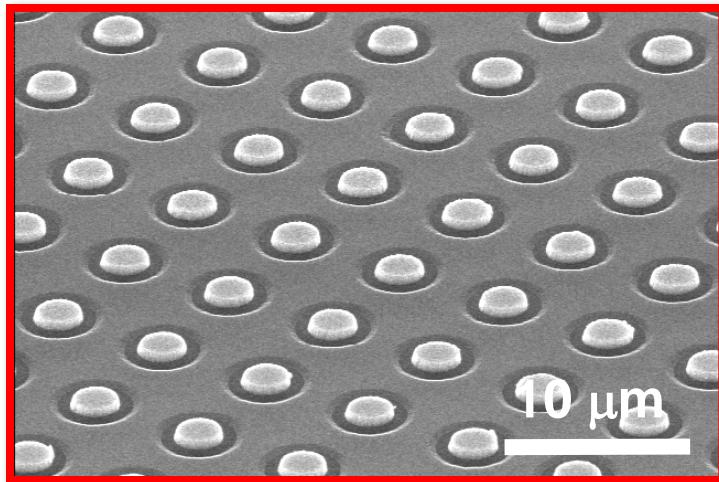
No oxide buffer layer



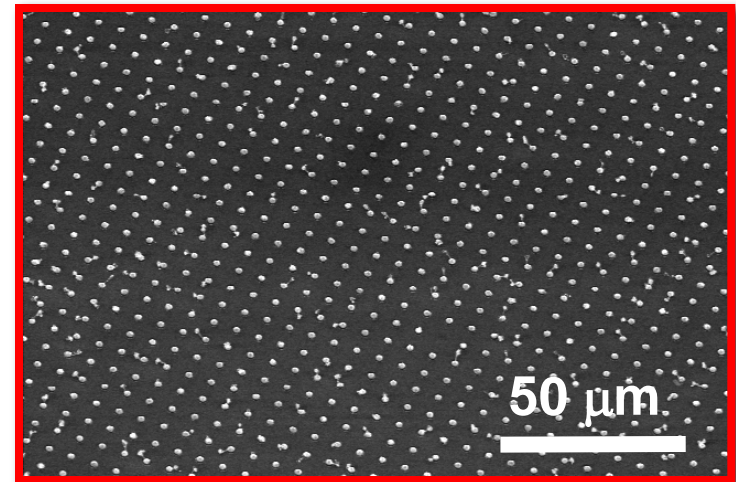
H₂ anneal
1000°C



Oxide buffer layer



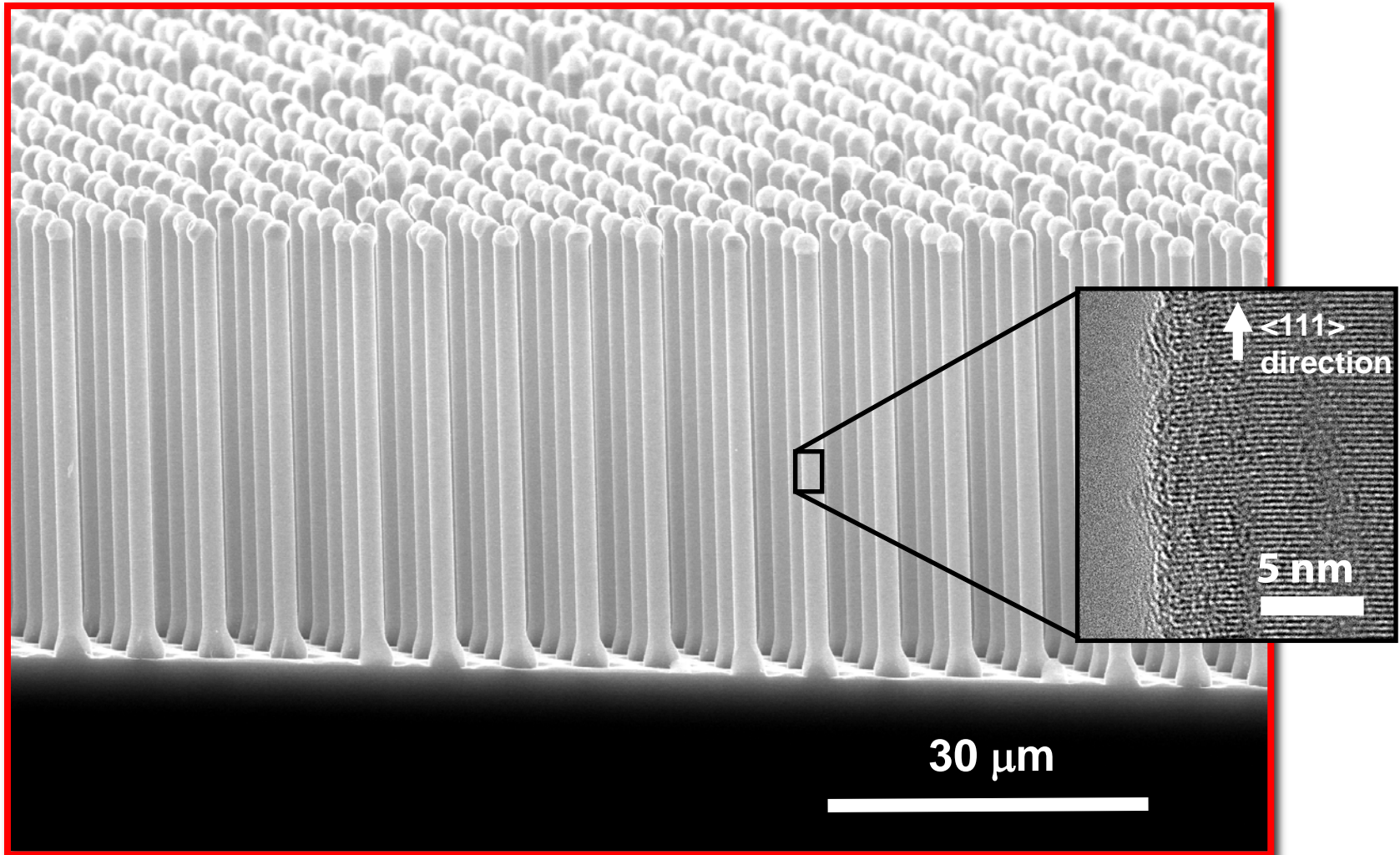
H₂ anneal
1000°C



3 μm array, 500 nm Au, $T_{\text{growth}} = 1000^\circ\text{C}$, $P_{\text{growth}} = 760$ Torr

An oxide buffer layer is critical for maintaining pattern fidelity during growth.

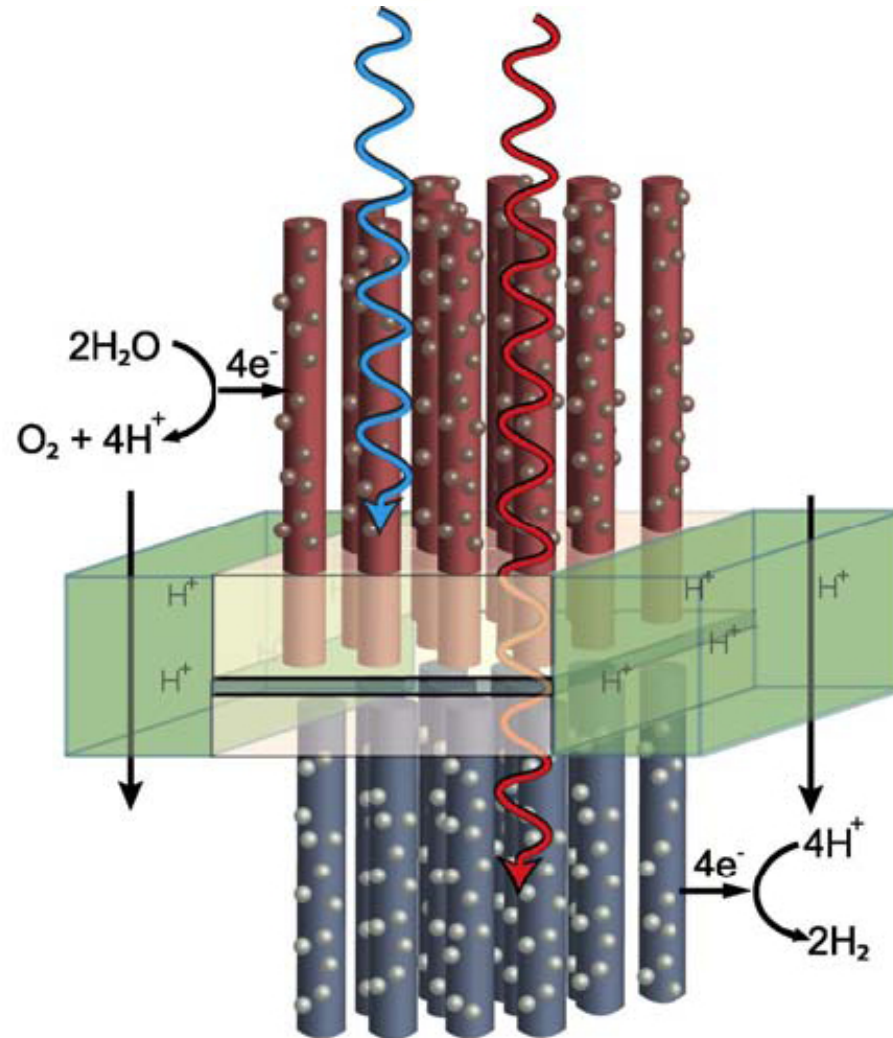
Large Area Au-Catalyzed Si Arrays



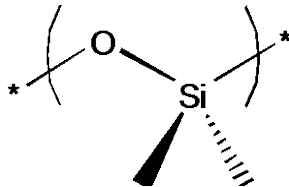
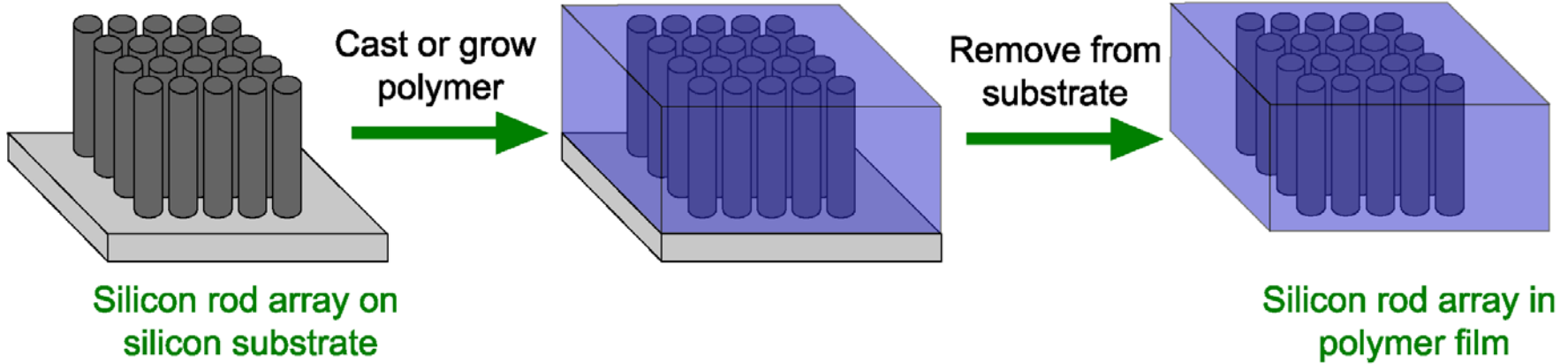
3 μm array, 500 nm Au, $T_{\text{growth}} = 1000^\circ\text{C}$, $P_{\text{growth}} = 760$ Torr, 30 min growth, 2 mole % SiCl_4 in H_2

Nearly 100% vertically aligned, 75 μm length microwire arrays over areas > 1 cm^2 .

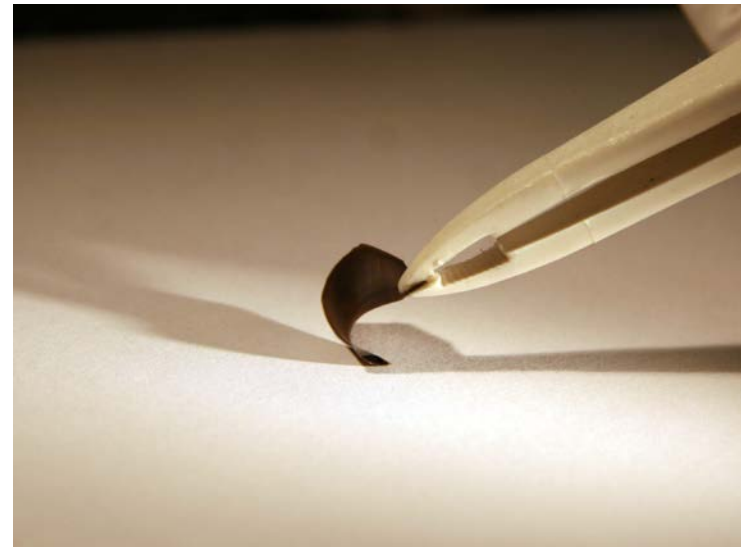
Constructing the Pieces of a Solar H_2 Fuel Generator



Polymer Embedding of Si Rod Arrays

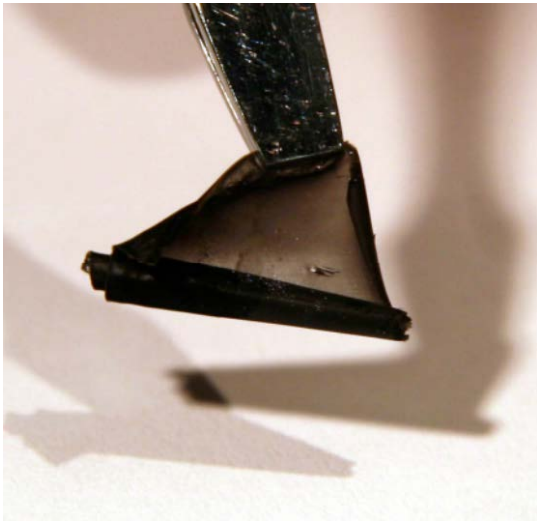


PDMS
(polydimethylsiloxane)

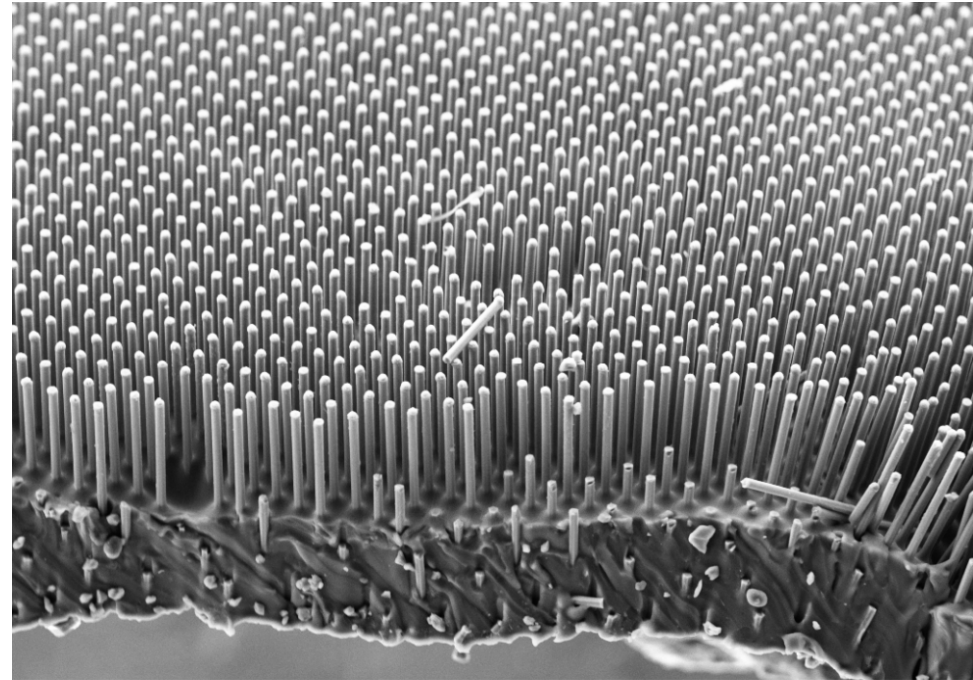


Large Area Rod Array Removal

Top-down view



Side view



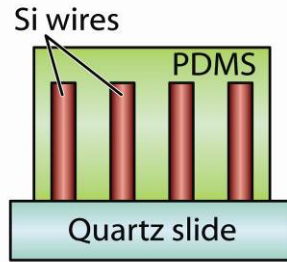
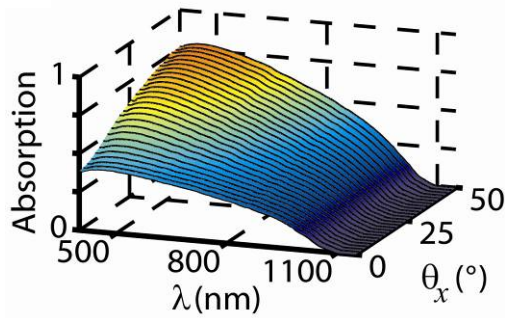
Caltech 10μm Mag = 1.51 K X EHT = 10.00 kV WD = 5 mm Signal A = SE2 Date :6 Mar 2008

- Large area arrays ($> 1 \text{ cm}^2$) transferred in one piece.
- Conformal coating from top to bottom of rods

Maximizing Si wire absorption

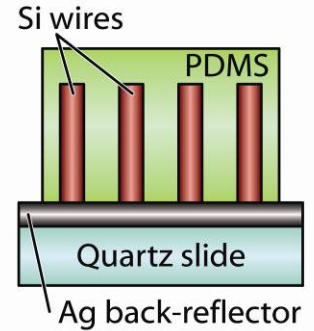
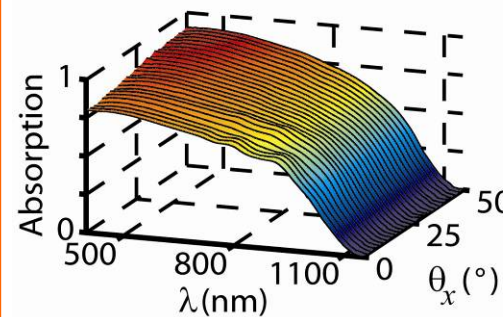


Polymer-embedded Si wire array



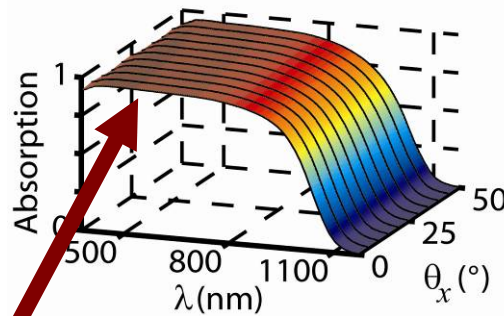
$$A(\theta_x, \lambda) = 1 - R(\theta_x, \lambda) - T(\theta_x, \lambda)$$

Polymer-embedded Si wire array w/ back-reflector

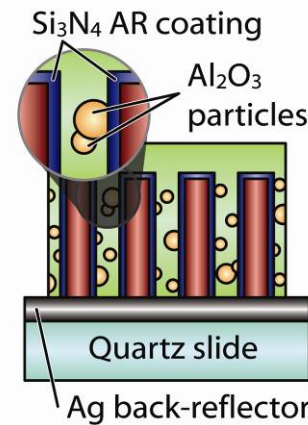


$$A(\theta_x, \lambda) = 1 - R(\theta_x, \lambda)$$

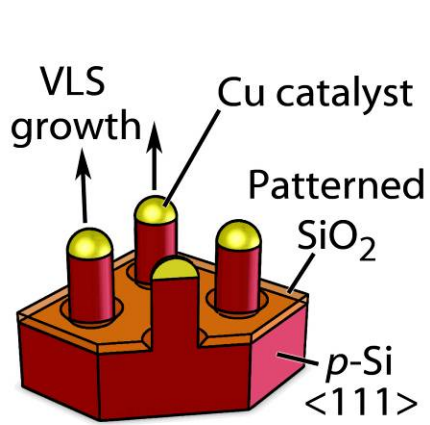
AR-coated array w/ light-scatterers and back-reflector



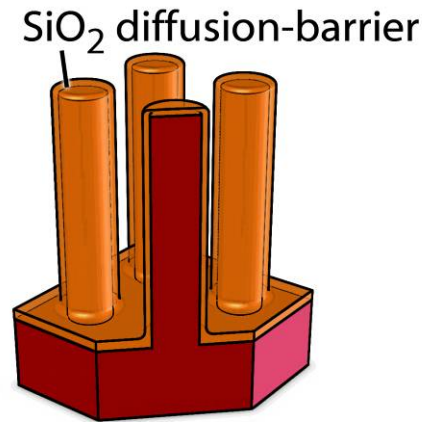
96% peak absorption



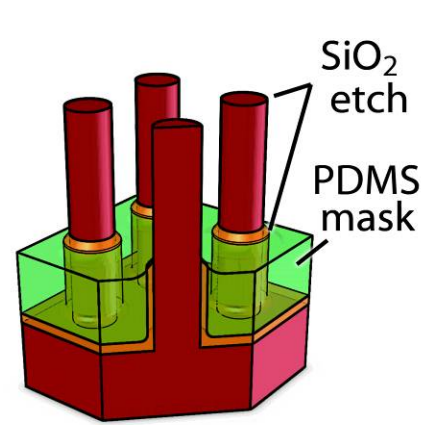
Si Microwire Solar Cell Fabrication Process



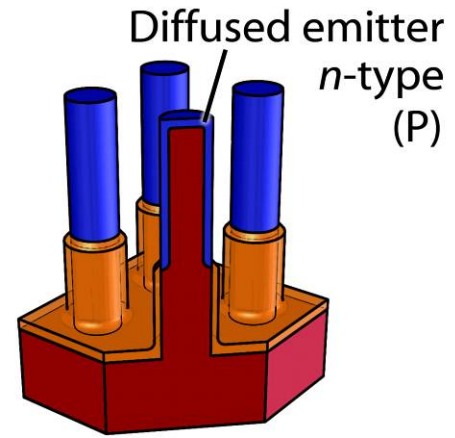
Growth



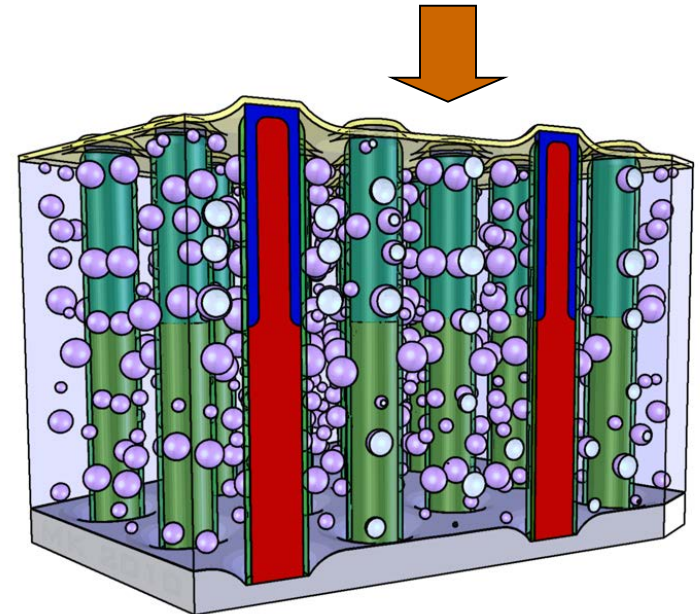
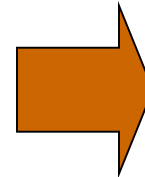
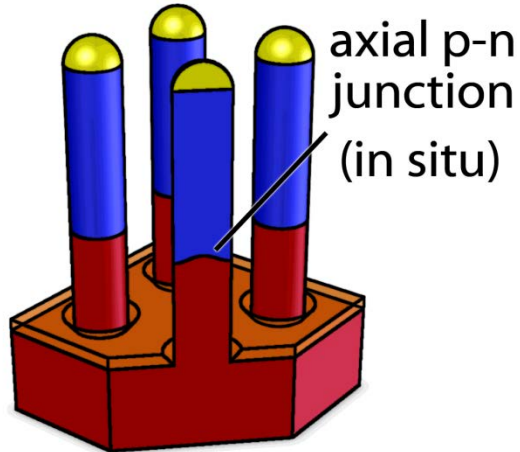
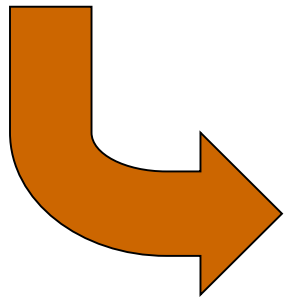
Oxidation



Chemical processing



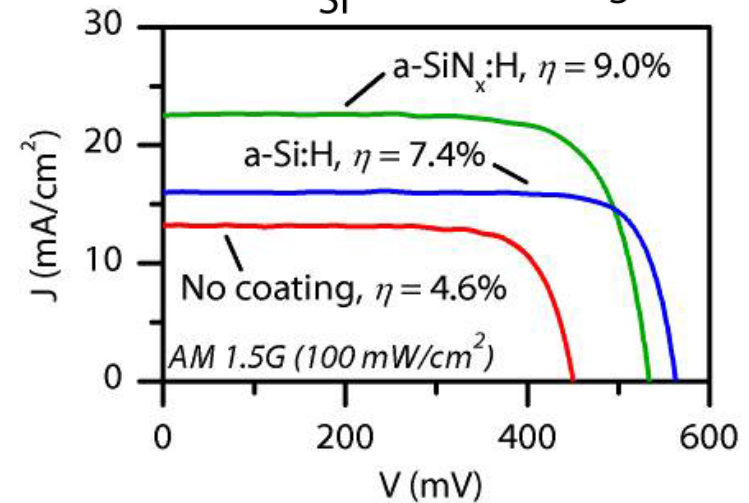
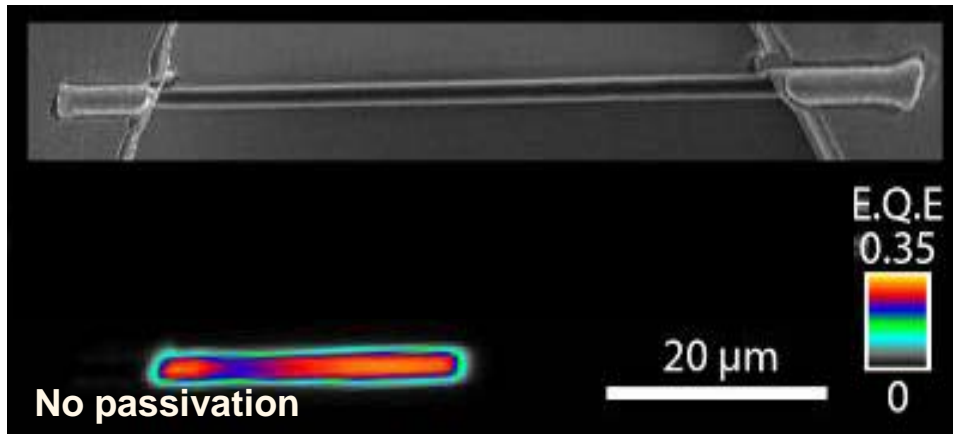
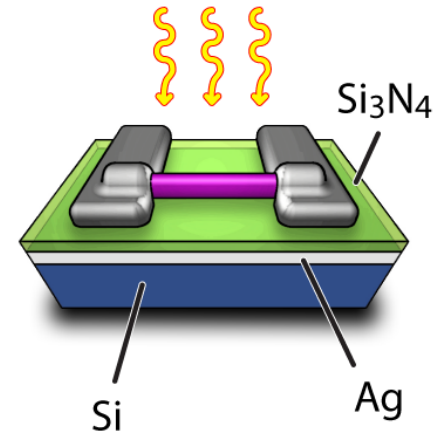
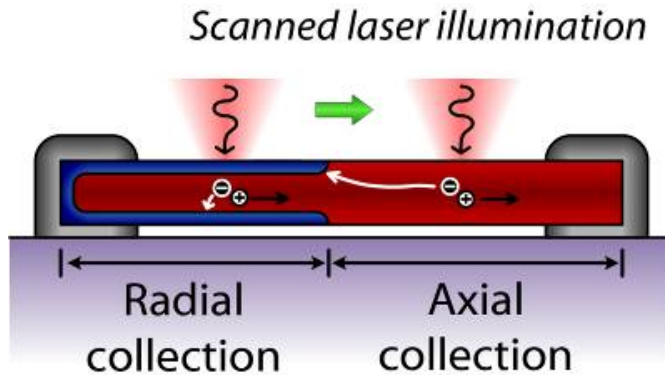
Diffusion



Insight from Single-Wires



AM 1.5G ($100 \text{ mW}\cdot\text{cm}^{-2}$)



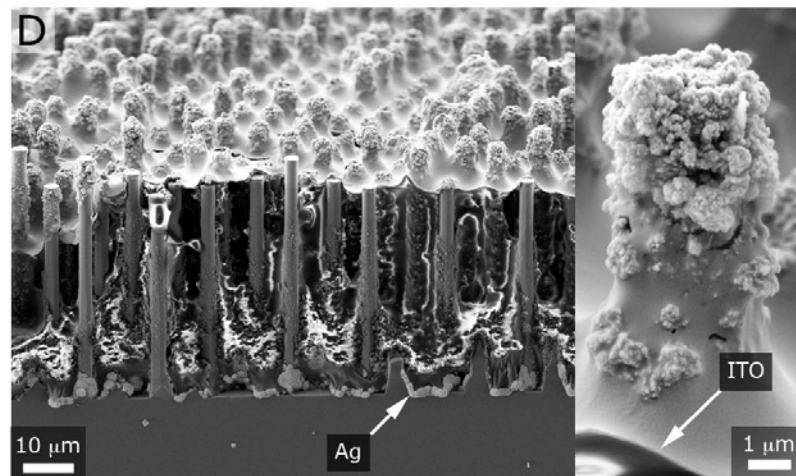
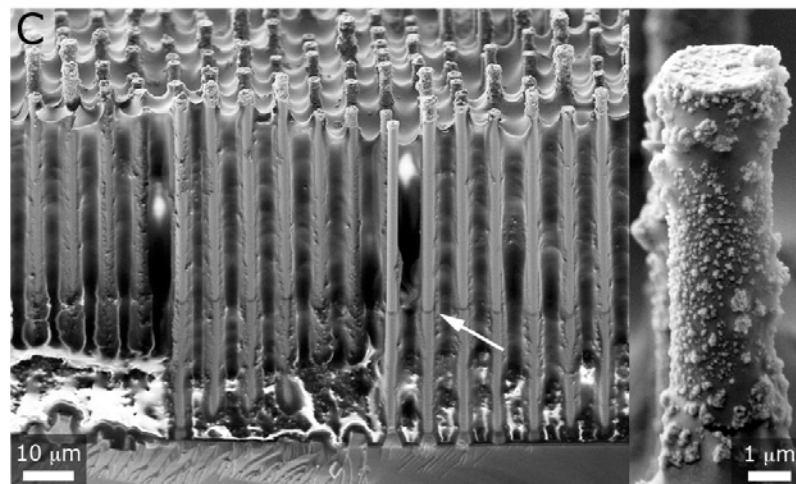
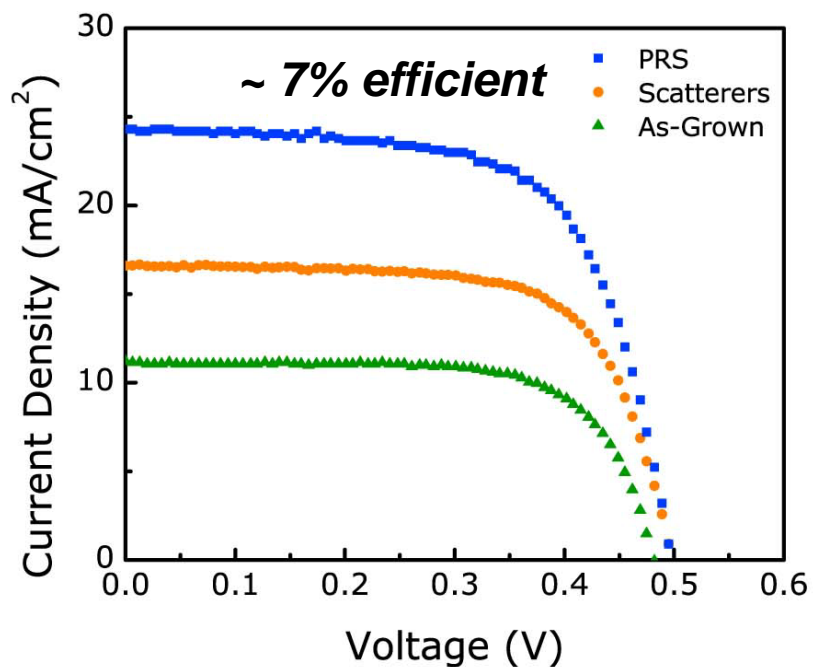
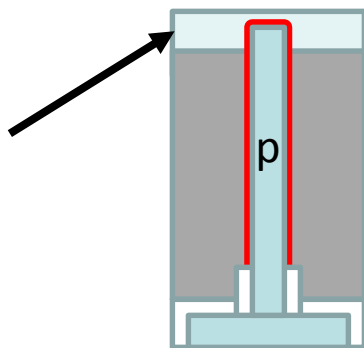
**Minority carrier diffusion length $\gg 20 \mu\text{m}$
Confirms high quality VLS p-Si**

600 mV V_{OC} – competitive with commercial wafer-based Si.

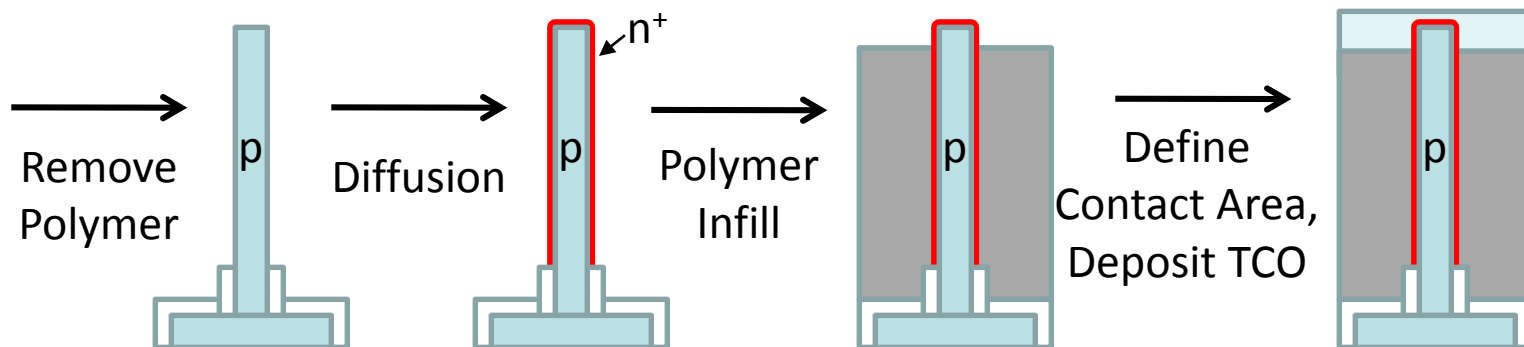
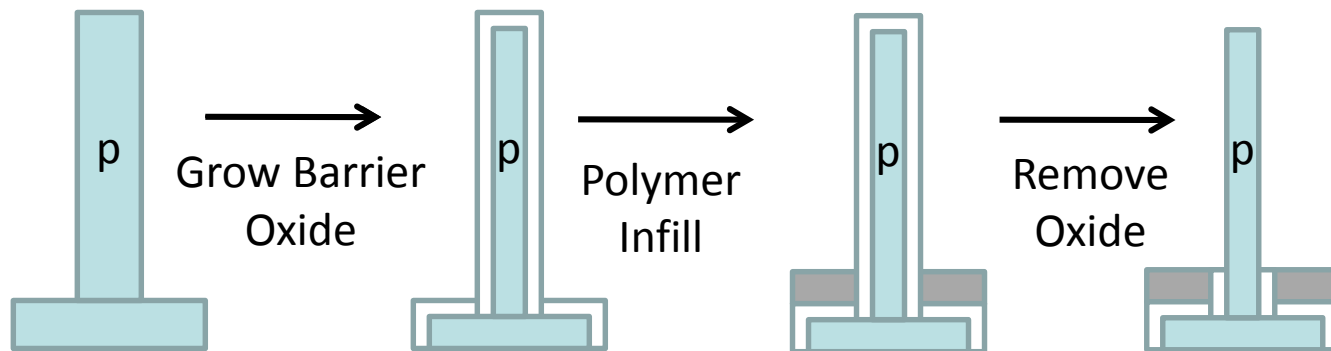
Solid-State PV Si Microwire Array Devices



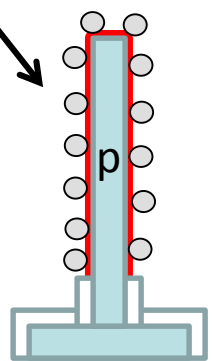
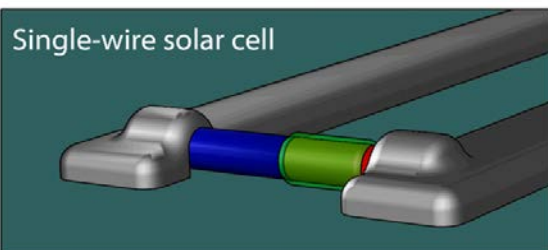
Add transparent top contact:



Radial pn-Junction Photocathode Process

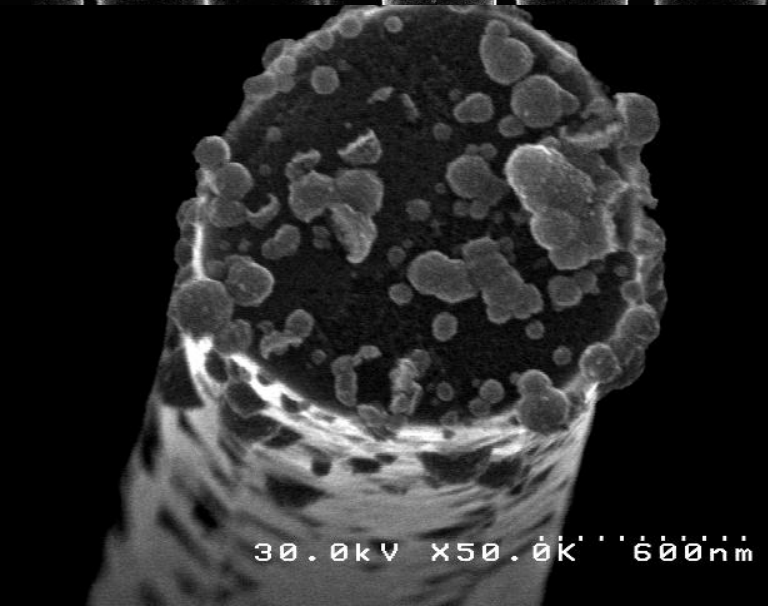
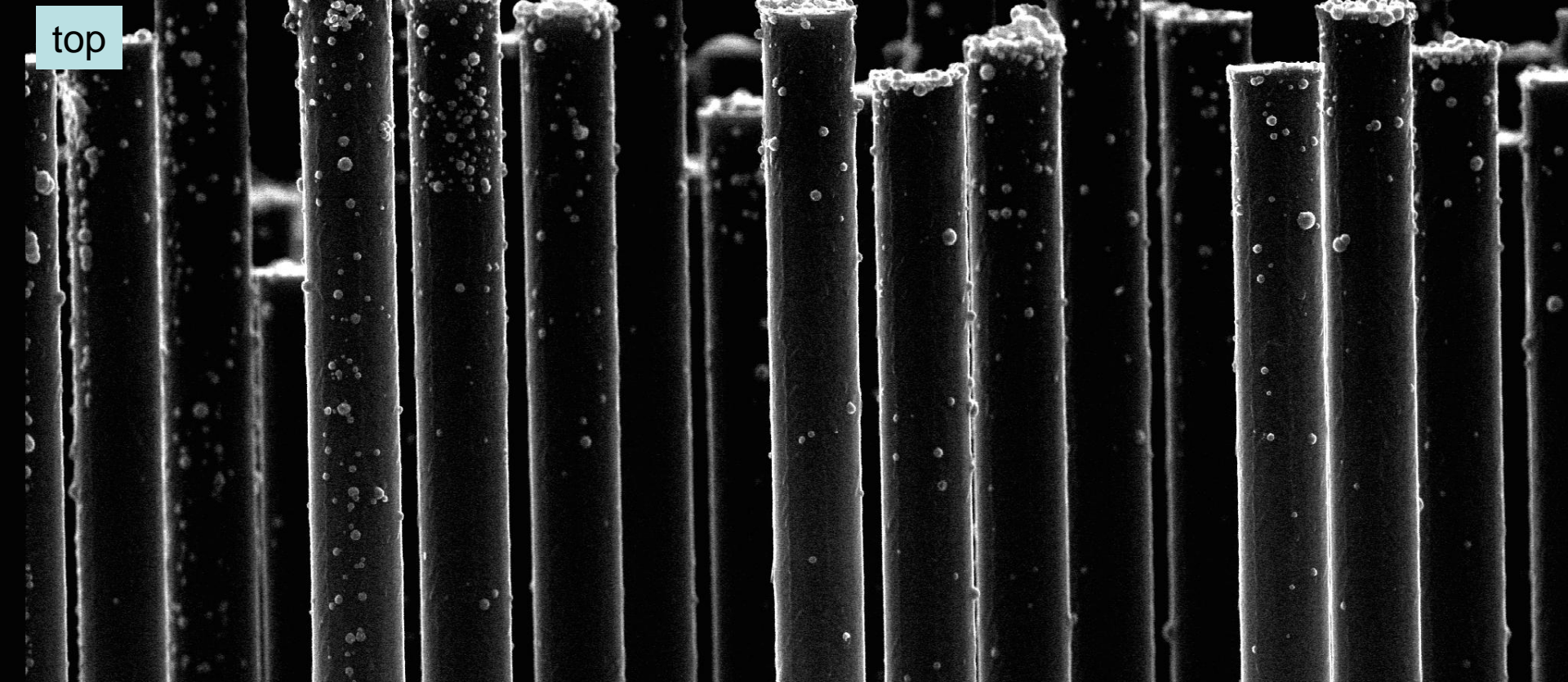


Solid-state wire-array PV



Incorporate metal catalyst for hydrogen evolution.

top

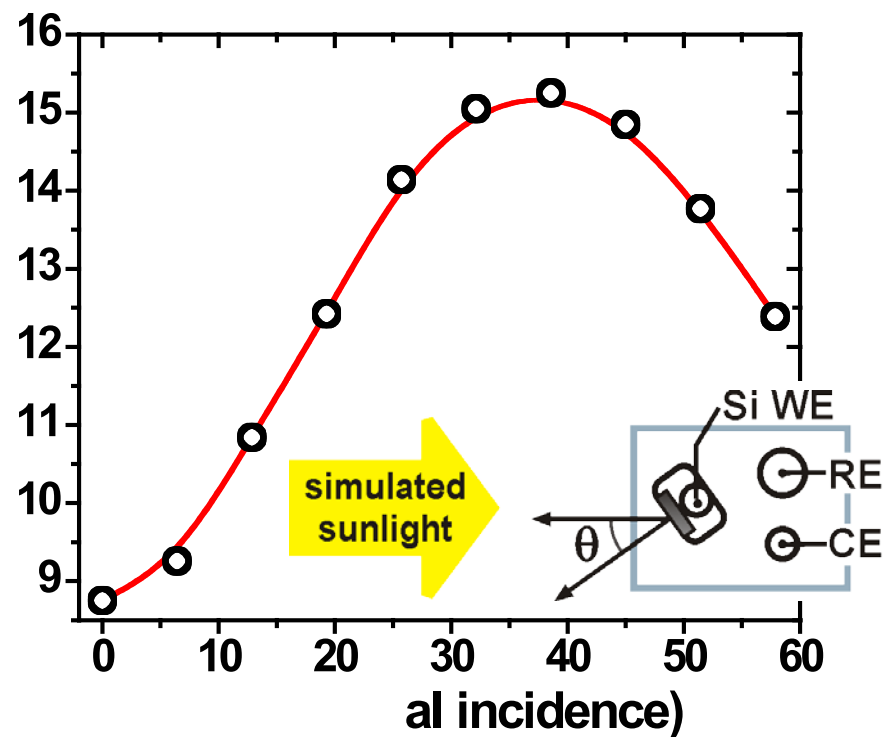
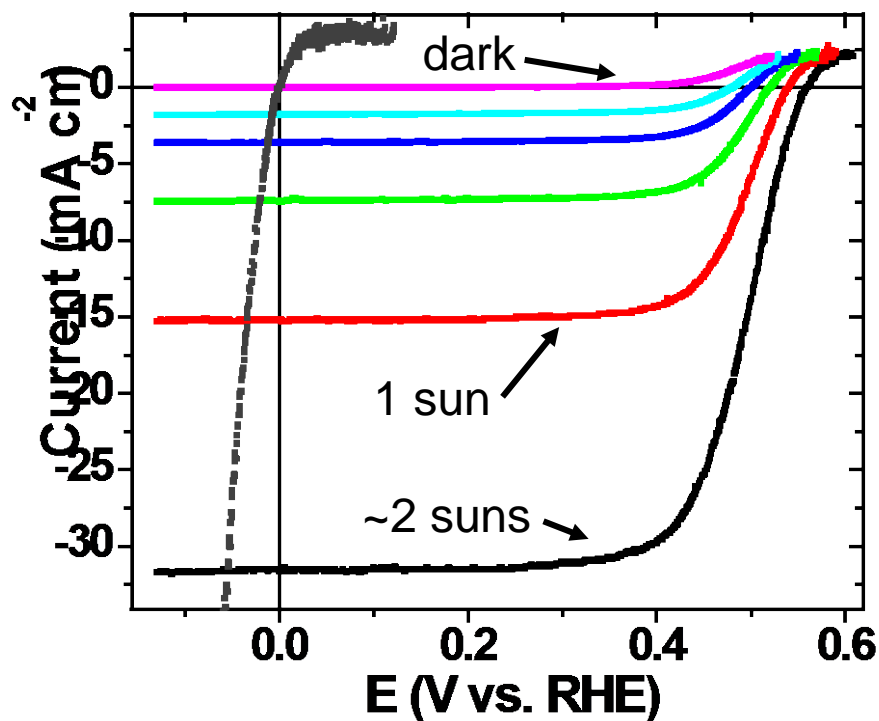


30.0kV X50.00K 6.00um

30.0kV X50.00K 600nm

Platinized 6 min, 0.25 M HF 0.5 mM K_2PtCl_4

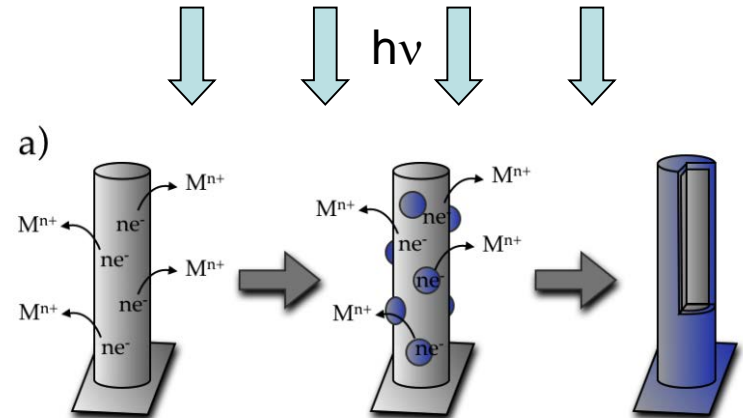
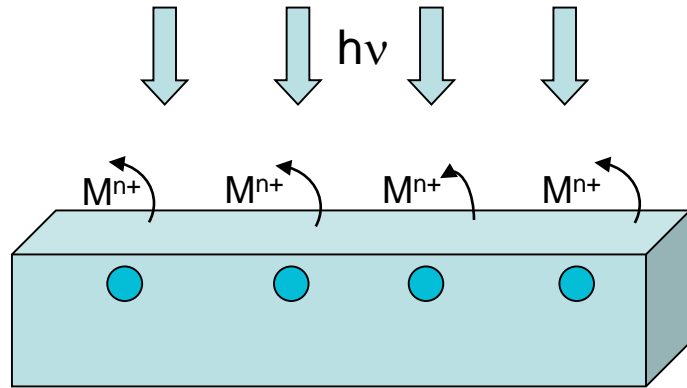
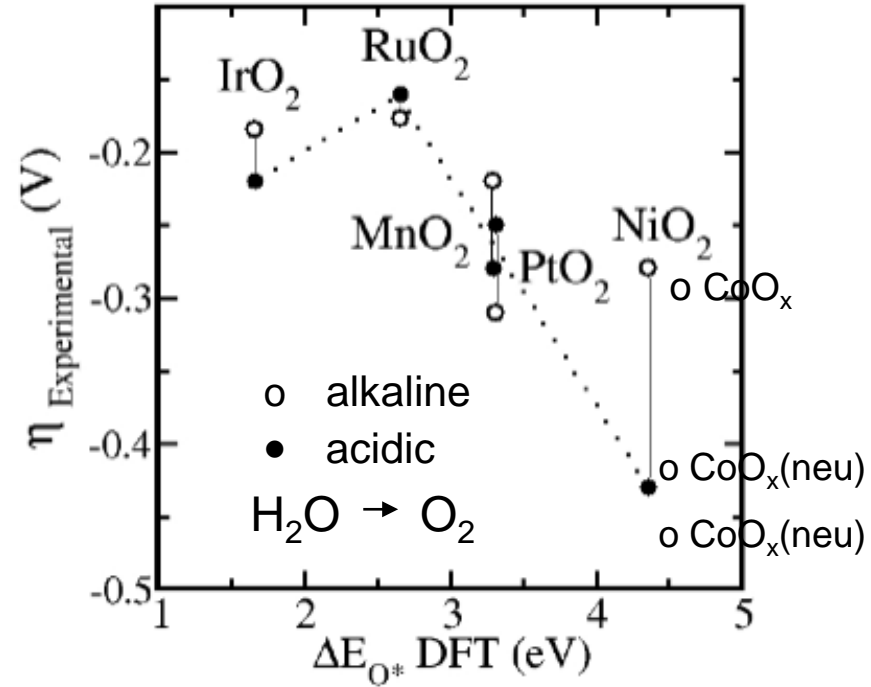
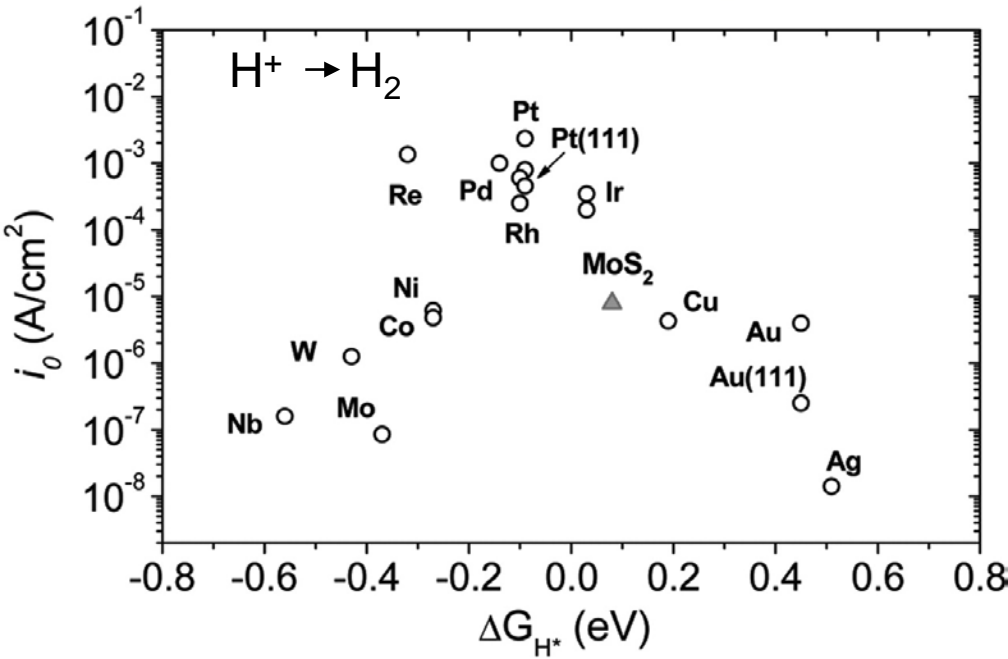
H_2 Evolution from Pt/pn-Si Rods



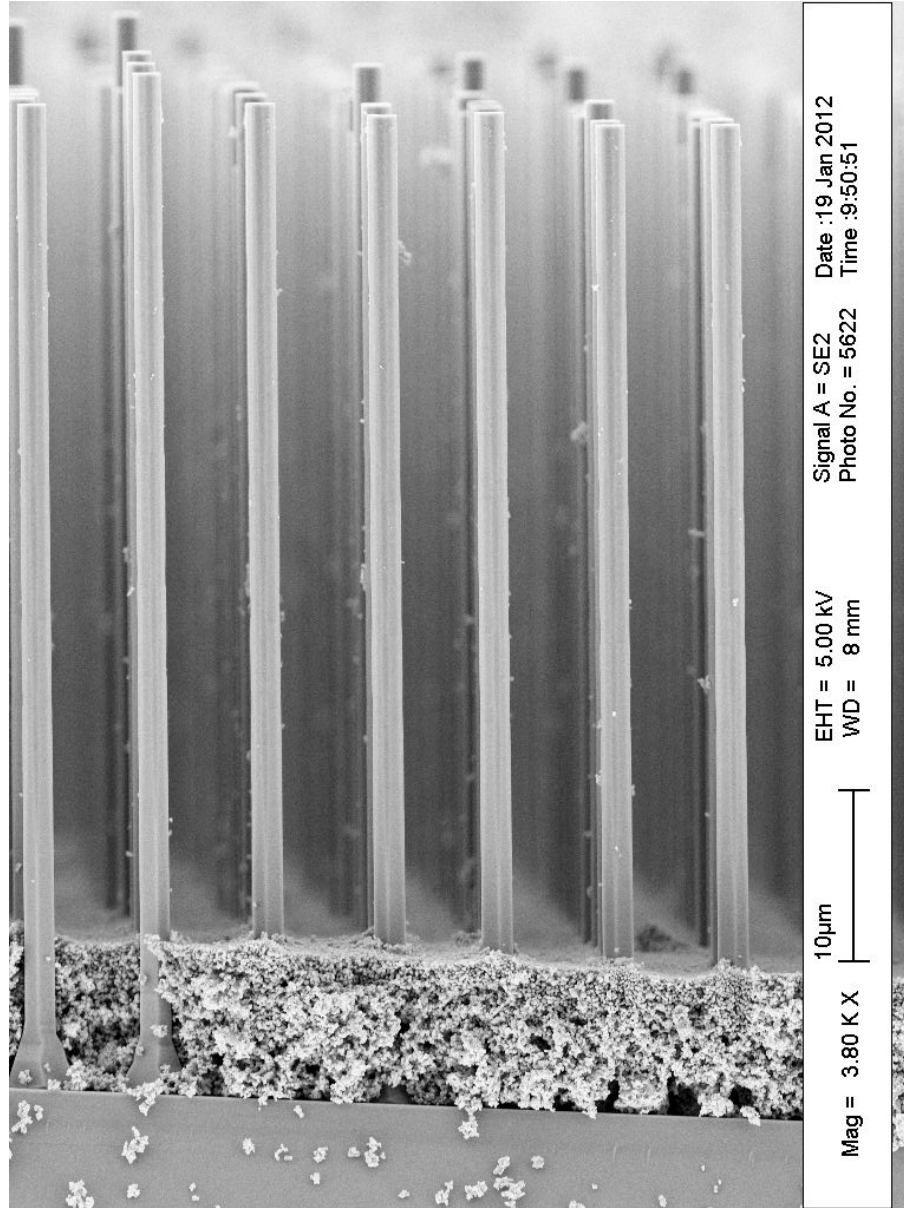
$V_{OC} = 539$ mV, $FF = 0.71$, $J_{SC} = 15.3$ mA cm^{-2} , $\eta = 5.8$ %

Near ideal fill factor, high V_{OC} , current low.

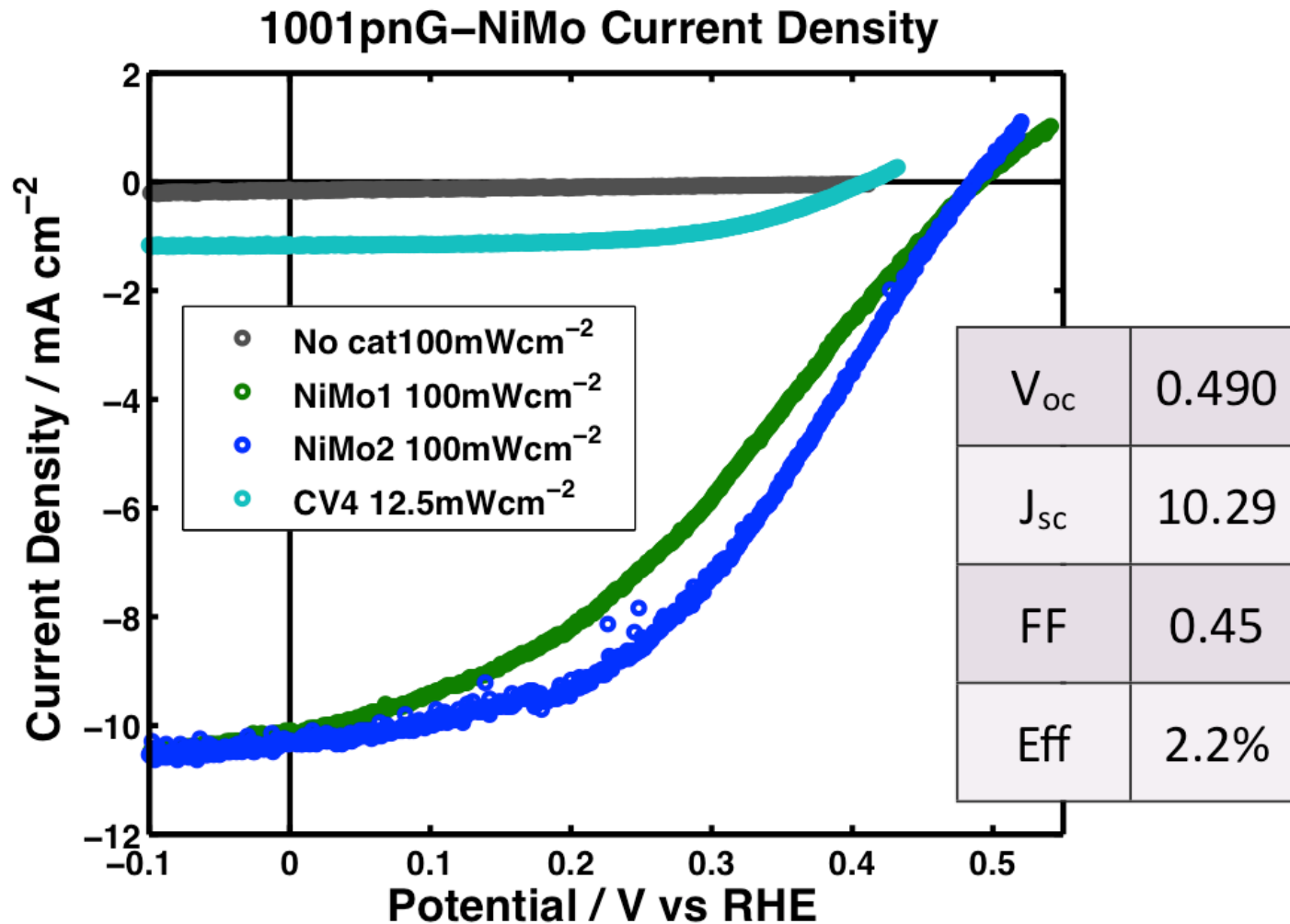
Relaxes Catalyst Activity Requirements



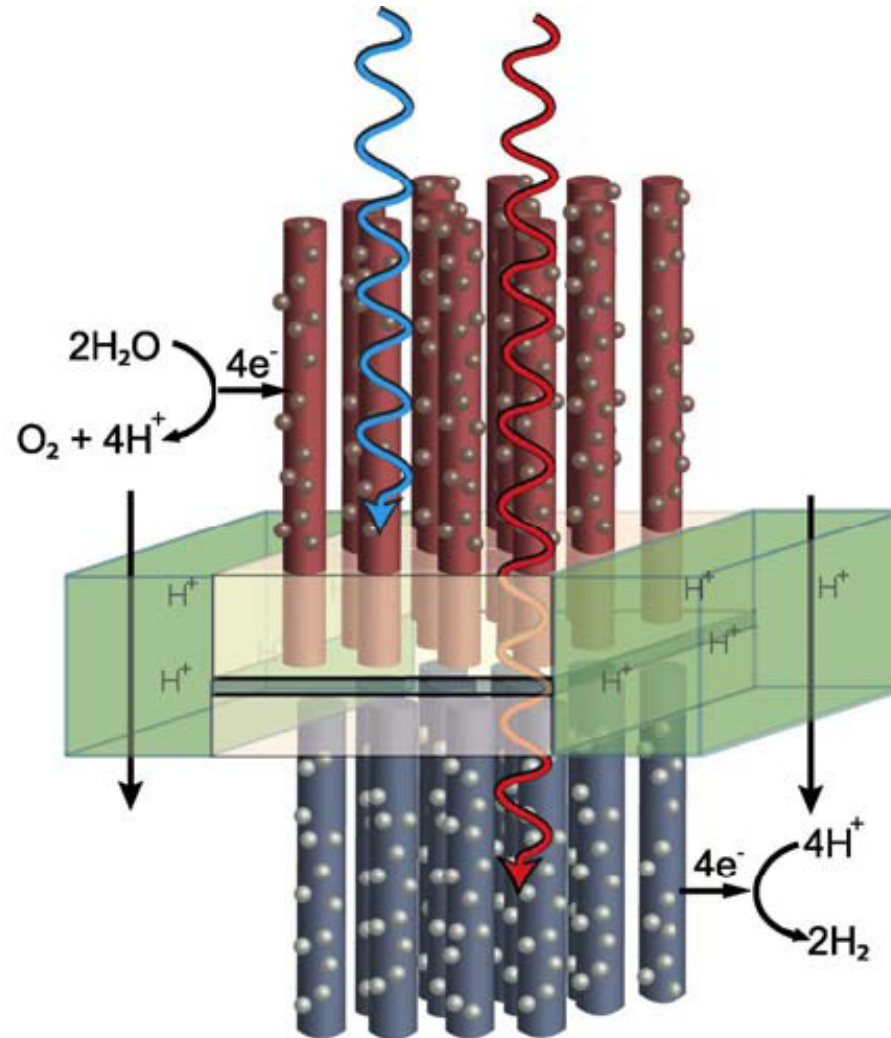
Ni-Mo HER Catalyst at Base of Si Wires



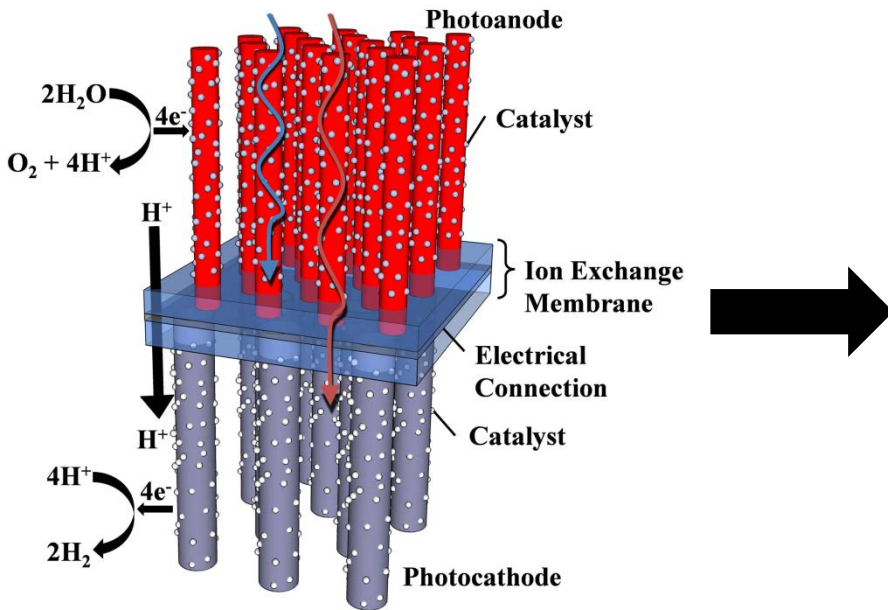
Ni-Mo on Si Microwires for H₂ Production



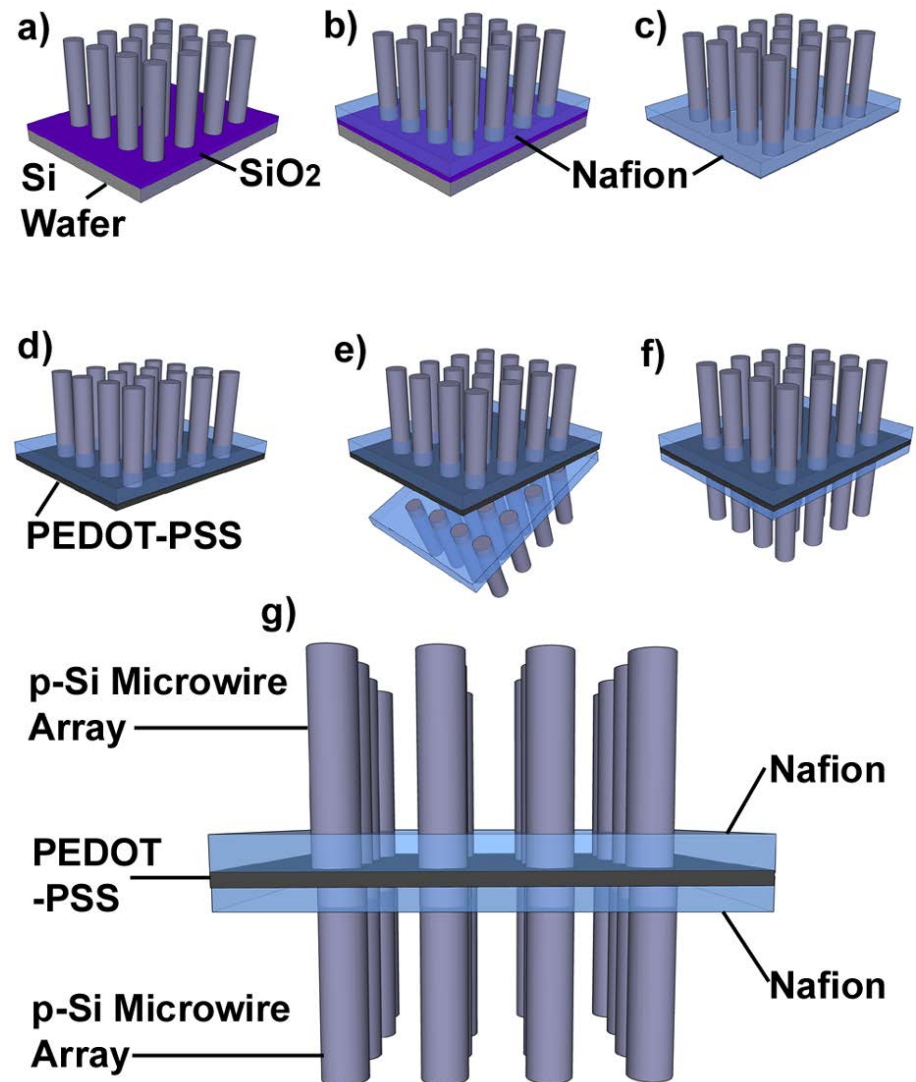
Constructing the Pieces of a Solar H_2 Fuel Generator



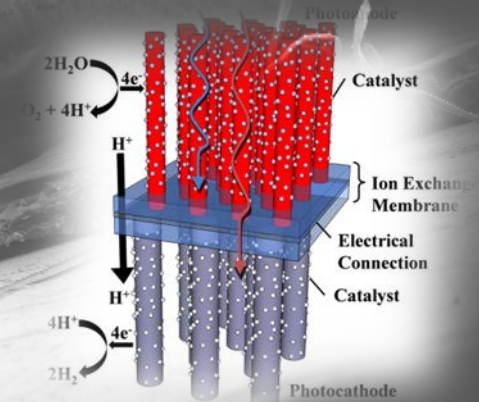
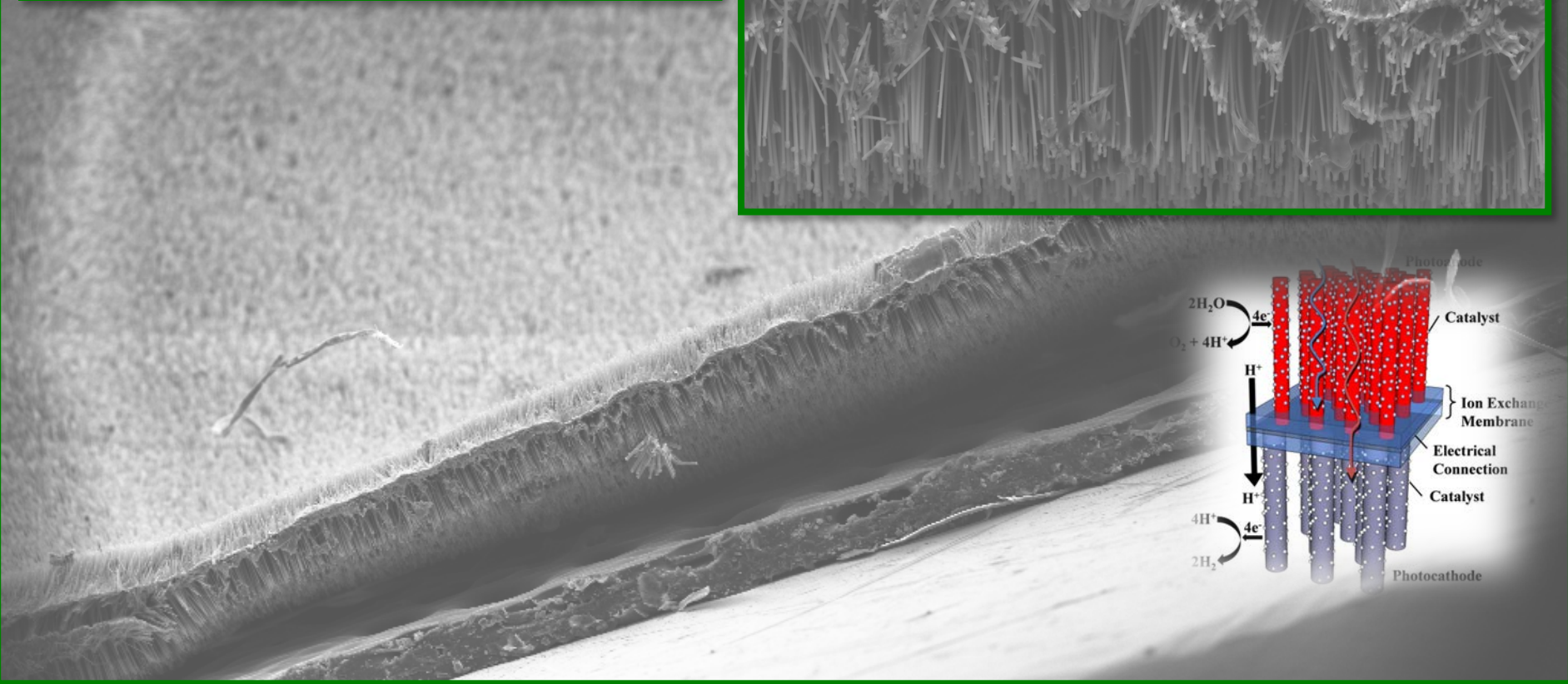
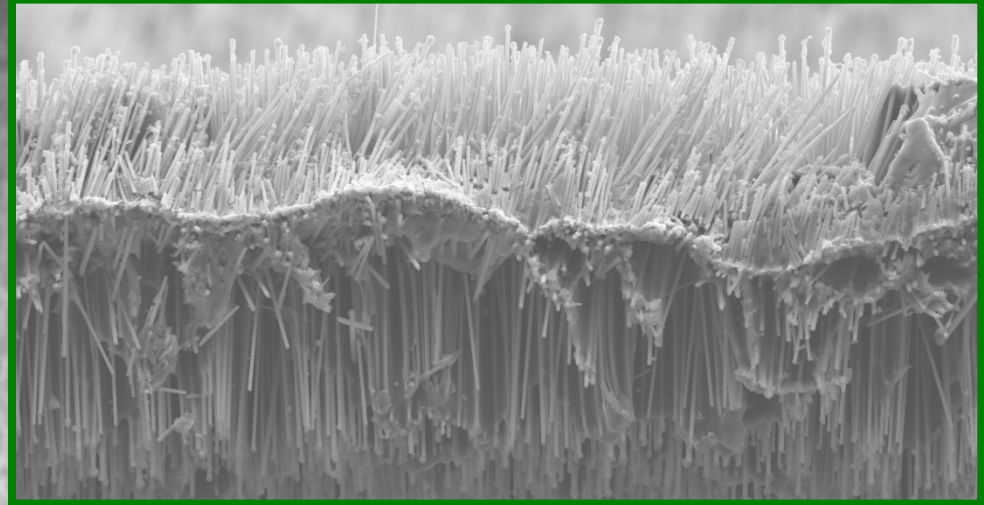
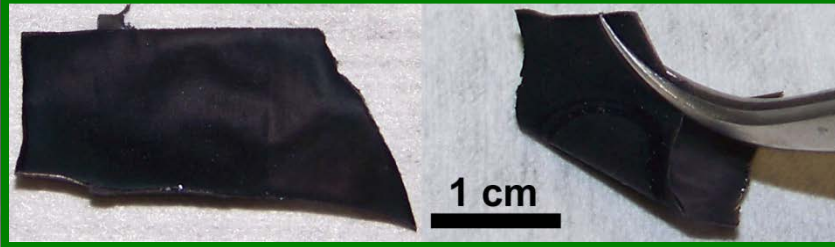
Test Membrane Fabrication



Si wire arrays on both sides and no catalyst as a first generation test membrane

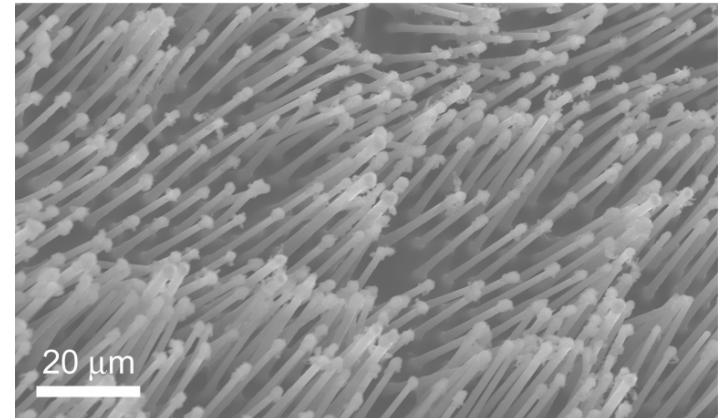


Dual Wire Array Membrane

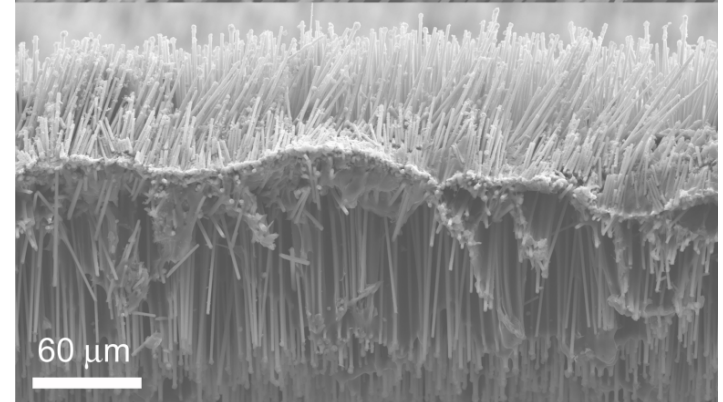


Si Wire/Ionomer Morphology

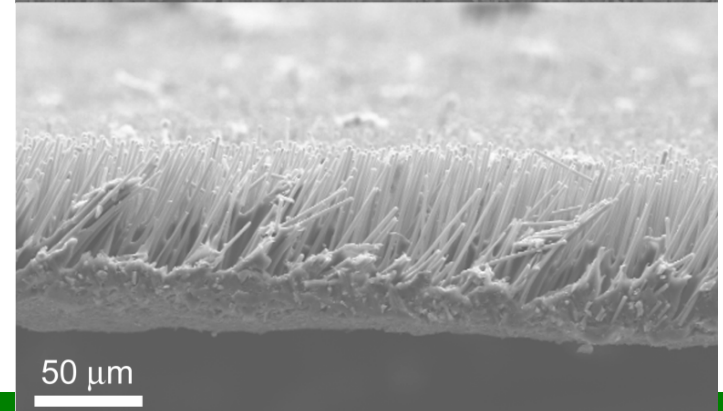
Dual (Si Wire Array/Nafion)/PEDOT-PSS



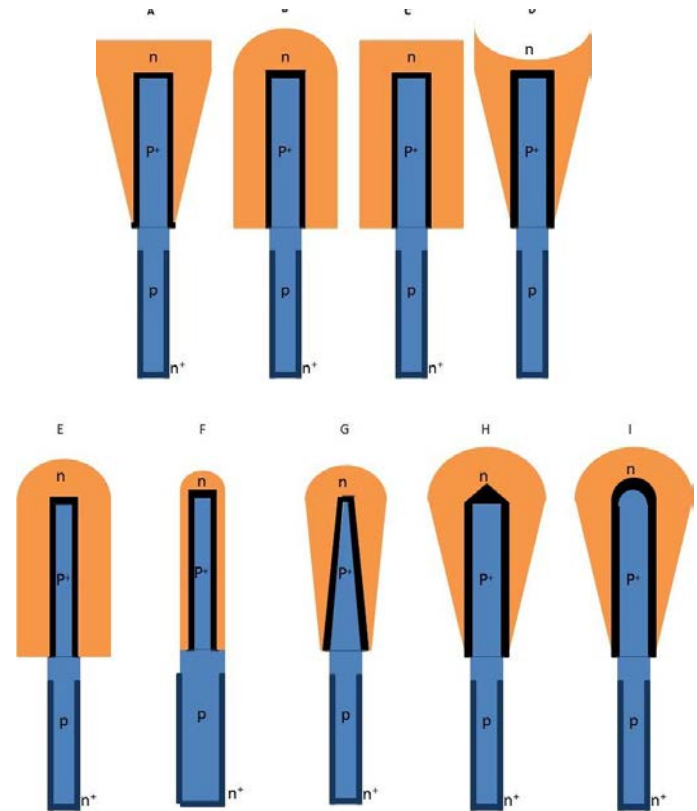
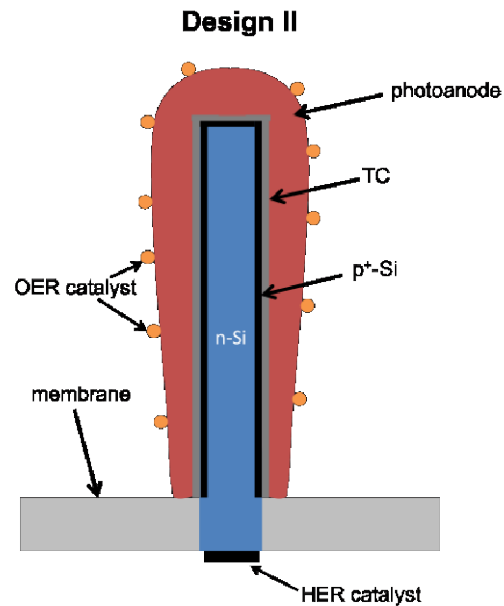
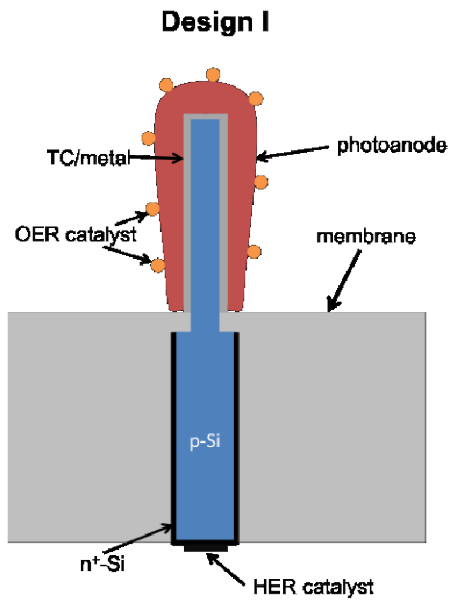
Dual (Si Wire Array/Nafion)/PEDOT-PSS



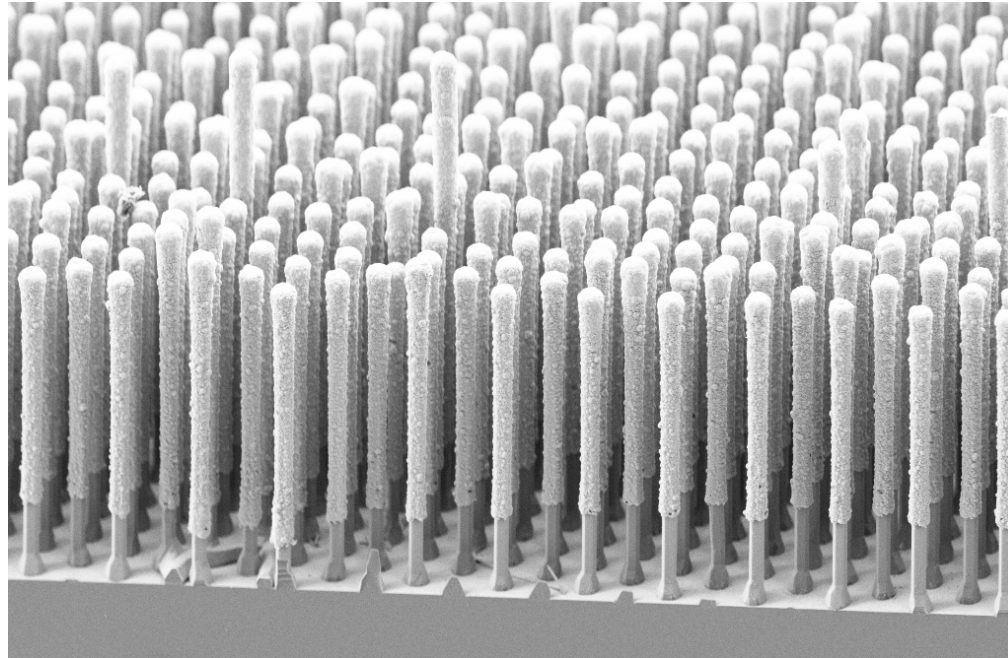
Si wire/QAPSF



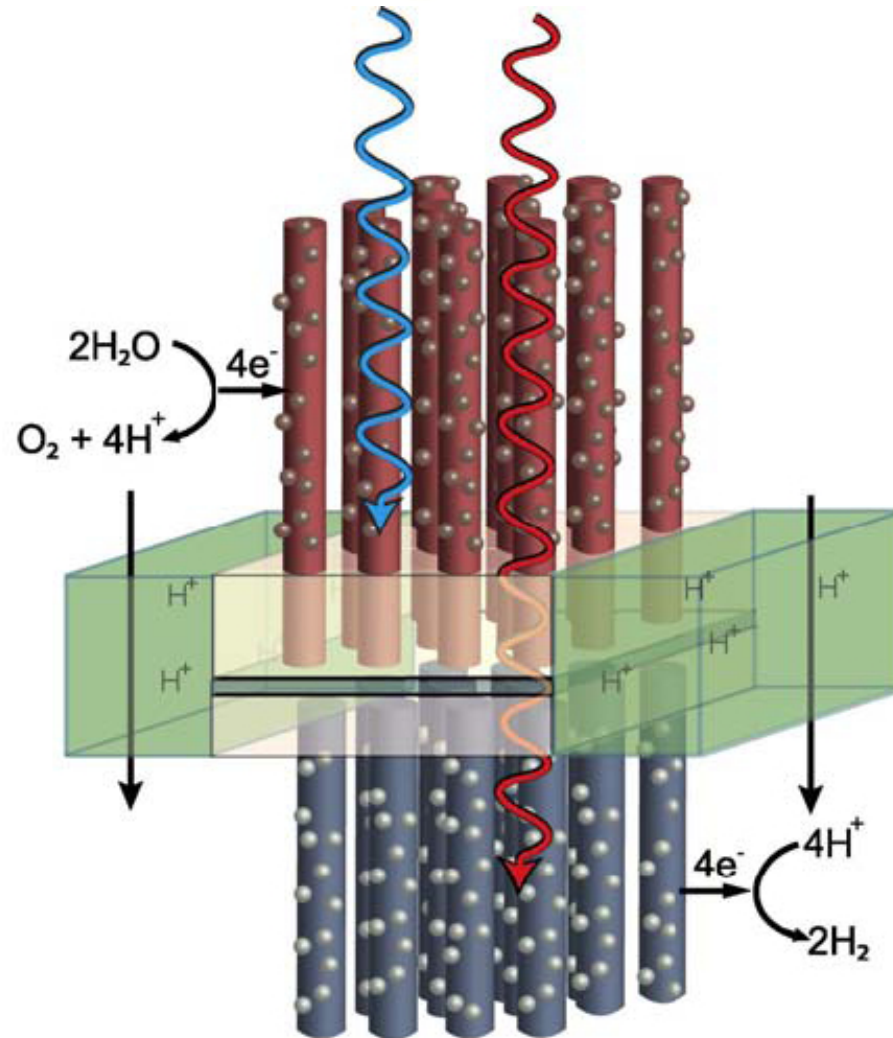
Self-Aligned Tandem Radial Junctions



p-n⁺ Si/ITO/WO₃ Tandem Radial Junctions



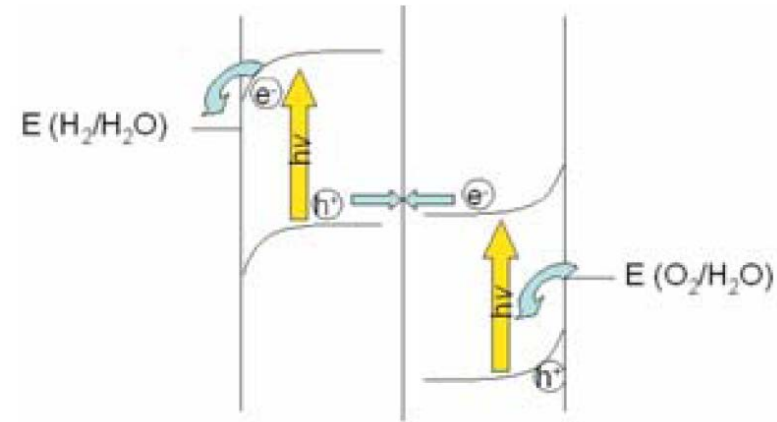
Constructing the Pieces of a Solar H_2 Fuel Generator



Dual Junction Nanorod Arrays



- 1.23 V needed to electrolyze water
- Requires a heterojunction or dual junction
- Single absorber: band gap of 2.0-2.6 eV that straddles the necessary potentials
- Photoanode and photocathode absorbers:

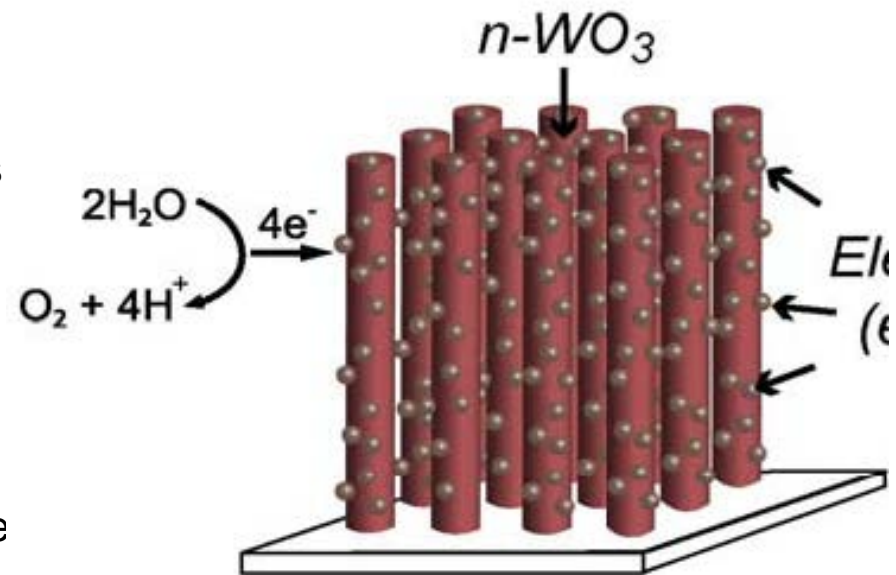


Band gaps can better match the solar spectrum (1.1-1.4 eV)

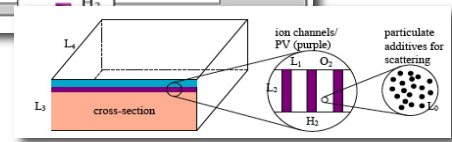
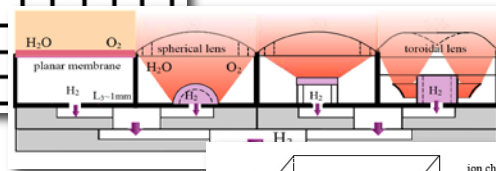
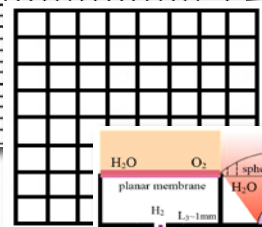
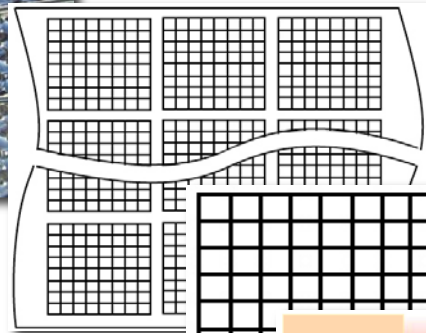
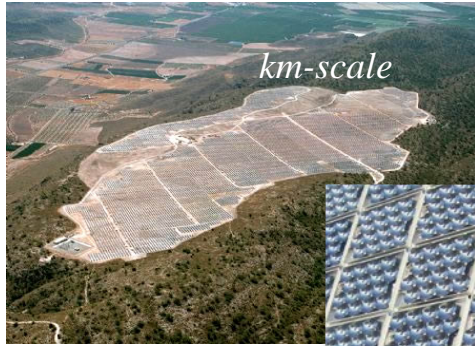
Allows for greater flexibility in materials design

Efficiency could approach 25%

- Optimal structure would have several light-absorbing layers absorbing different portions
- of the spectrum with an overall potential greater than 1.23 V



JCAP: DOE's Solar Fuels Energy Innovation Hub

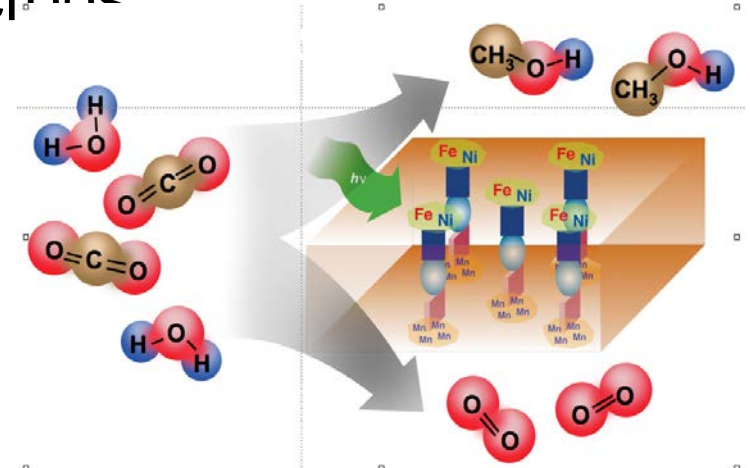
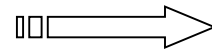
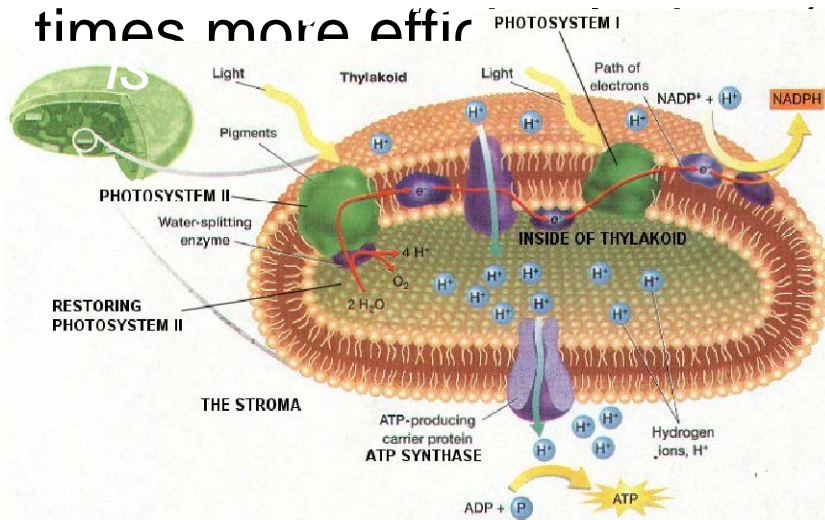


System/design/process level

Device/physics level

Mission of JCAP

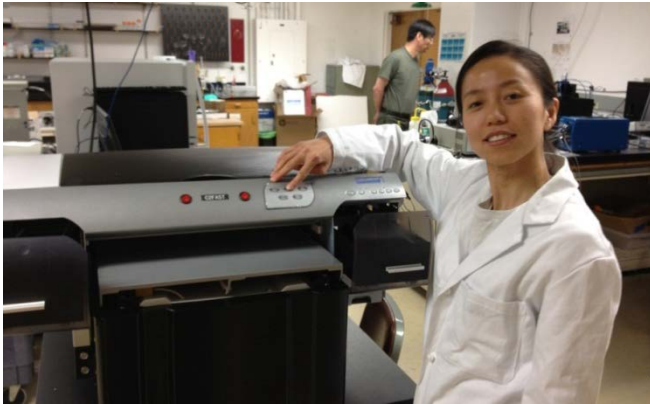
- **Melvin Calvin, 1982:** It is time to build an actual artificial photosynthetic system, to learn what works and what doesn't work, and thereby set the stage for making it work better
- **10-year JCAP Goal, 2010:** To demonstrate a manufacturably scalable solar fuel generator, using earth-abundant elements, that, with no wires, robustly produces fuel from the sun, 10 times more efficient (current) crops



Science Overview: Absorber Precursors



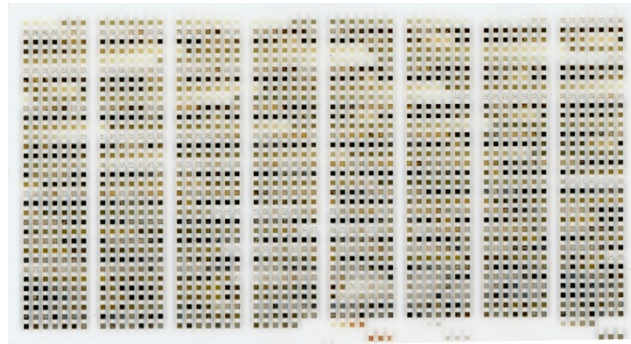
- Research Highlight – High-Throughput Experimentation (*continued*)



- Inkjet printing (generation 2)

- Reactive annealing
- Oxides, sulfides, nitrides, etc.

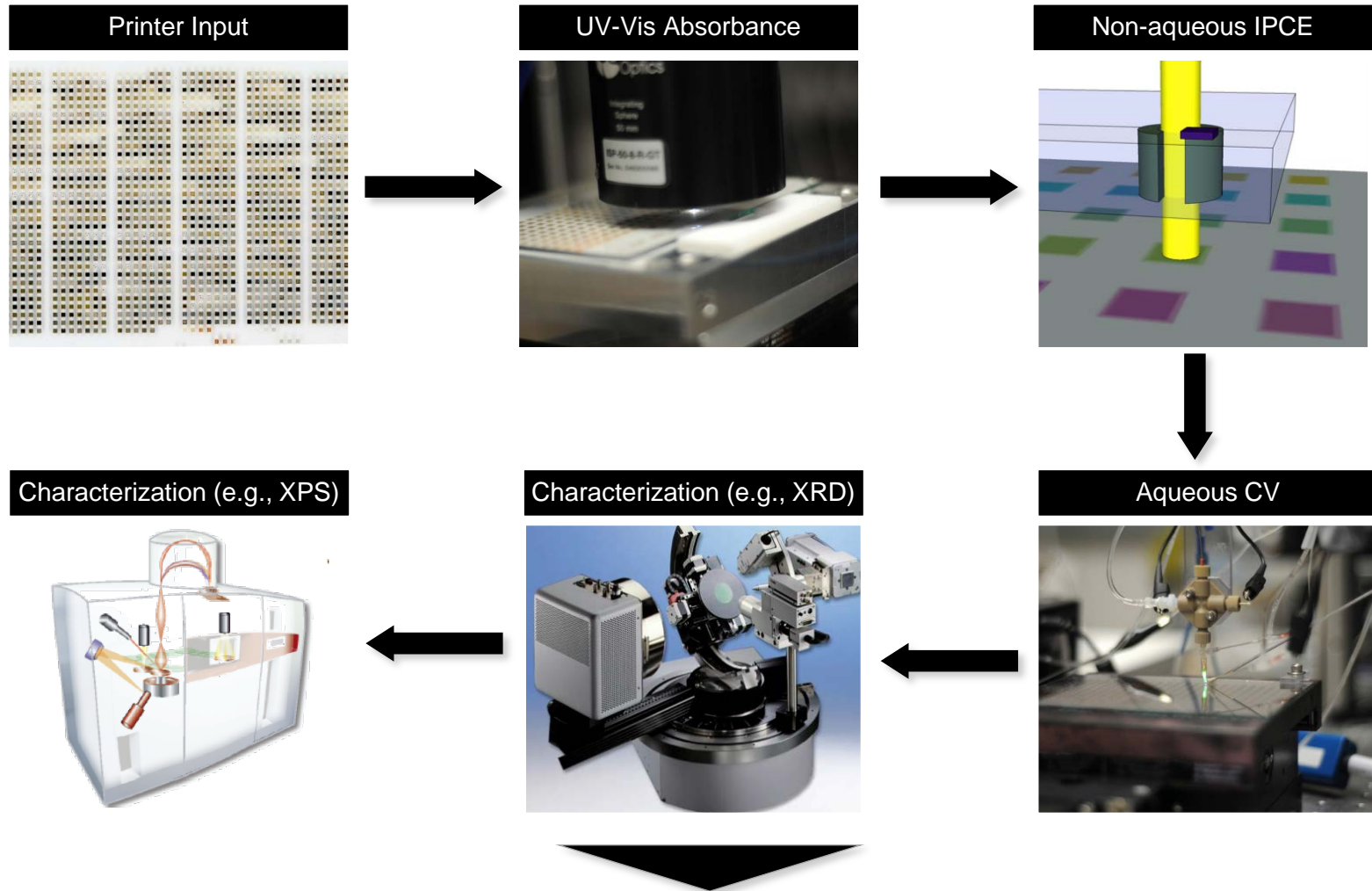
Co, Fe, Ti, Ni and baked: 1848 points 1mm²



Time:
5544 samples in 7 min.!
(highest resolution
at slowest speed)

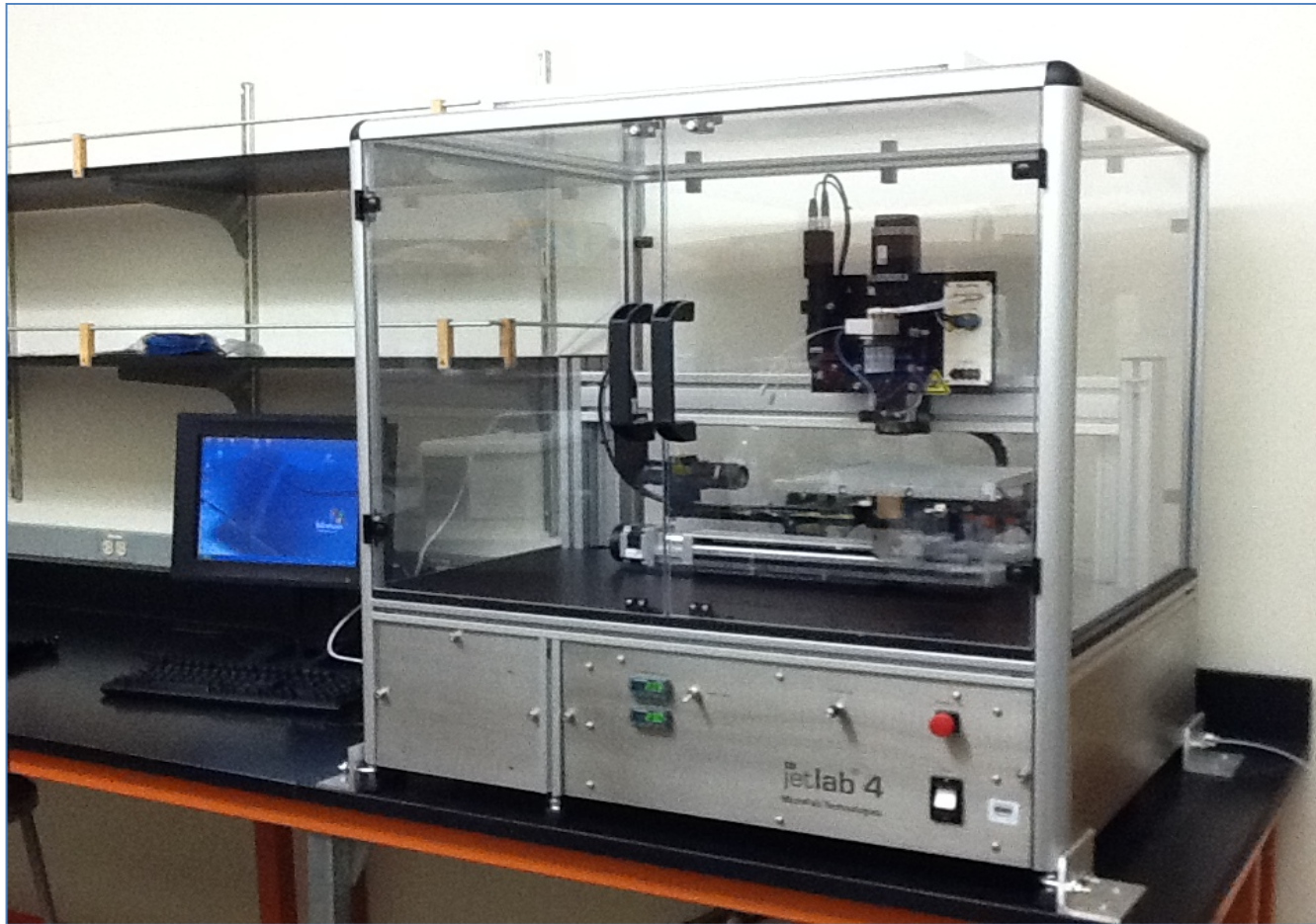
Inkjet printing methods can now produce
 10^4 - 10^5 samples /hour for high-
throughput fabrication

- Research Highlight – Establishment of Advanced Screening Systems



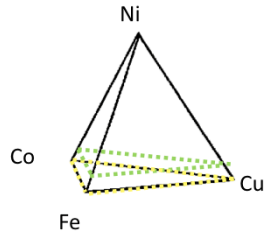
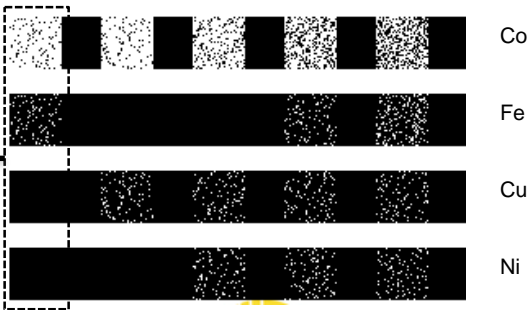
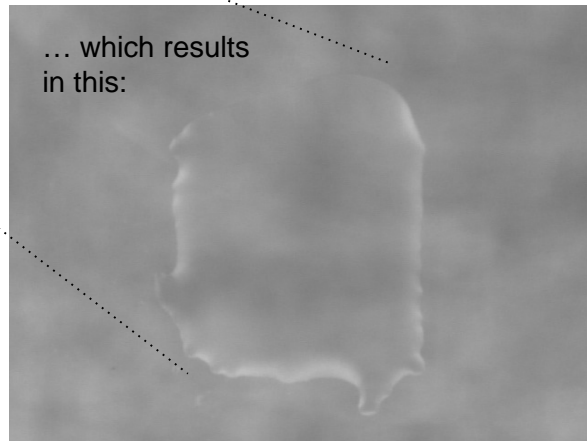
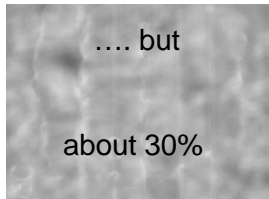
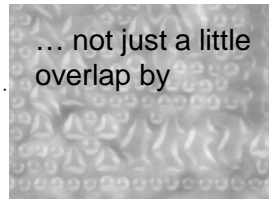
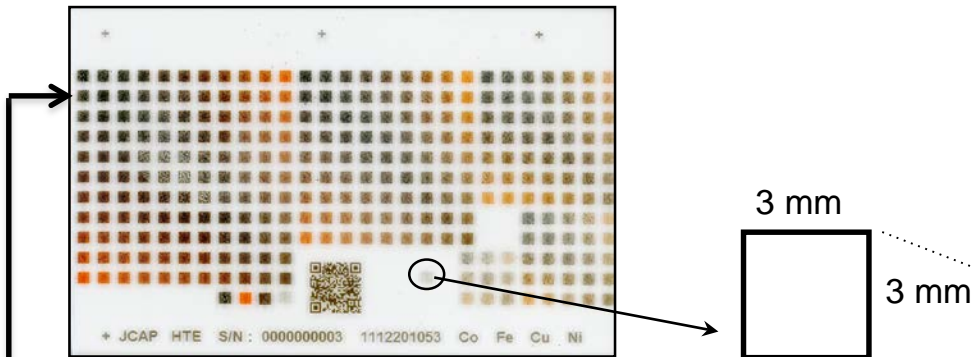
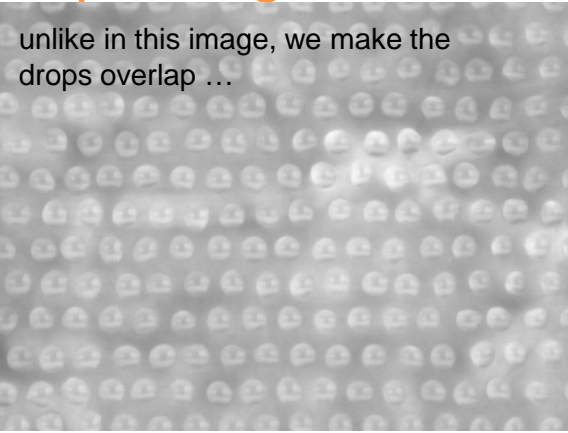
**Development of World-Class
Tools for High-Throughput Characterization and
Analysis**

MICROFAB printer



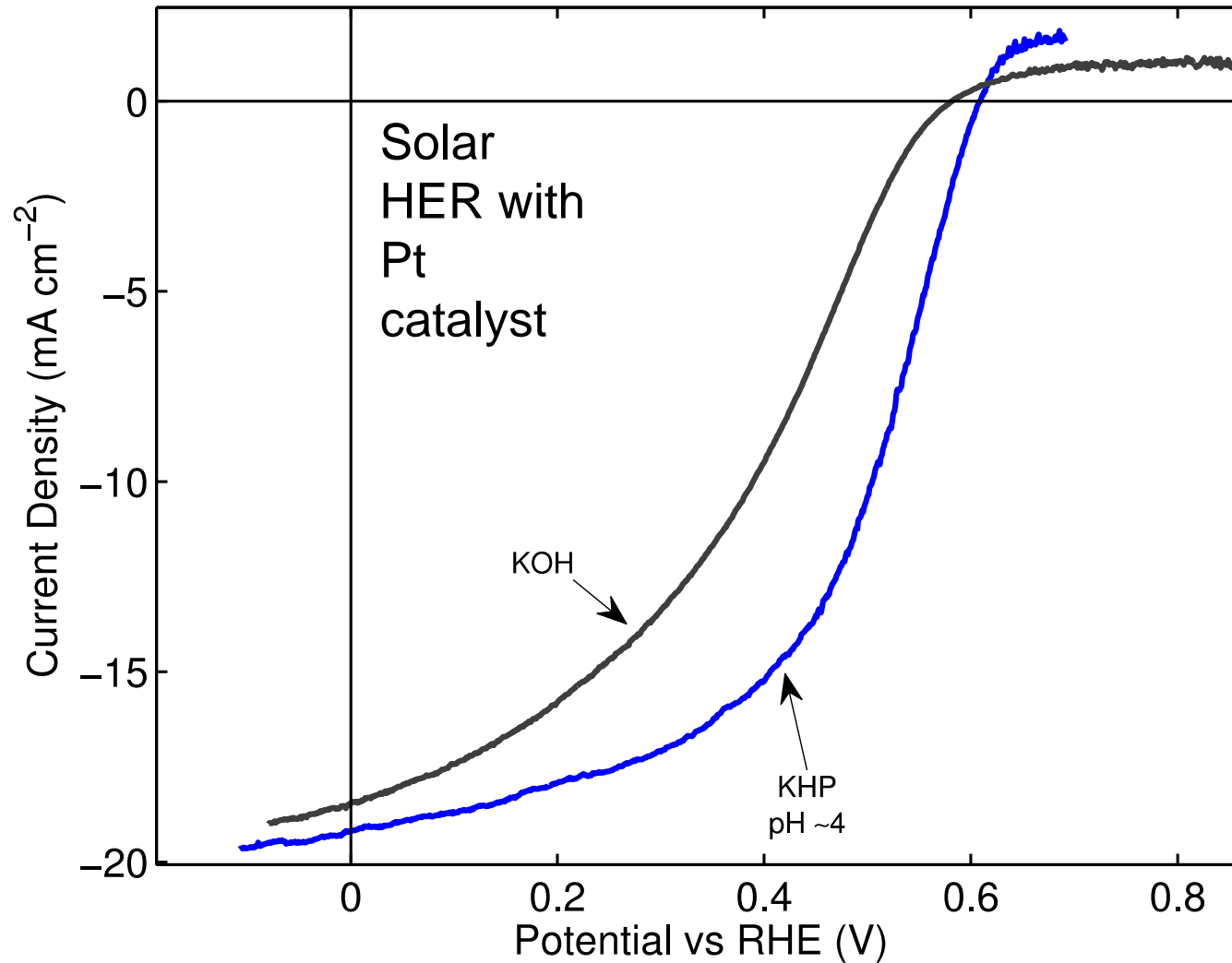
innovation no. 1 – printing methodology

- get medium throughput
- print 9 mm² spots
- print constant volume spots



Bitmaps for 5 spots on the plate (automation routines by Marty Marcin)

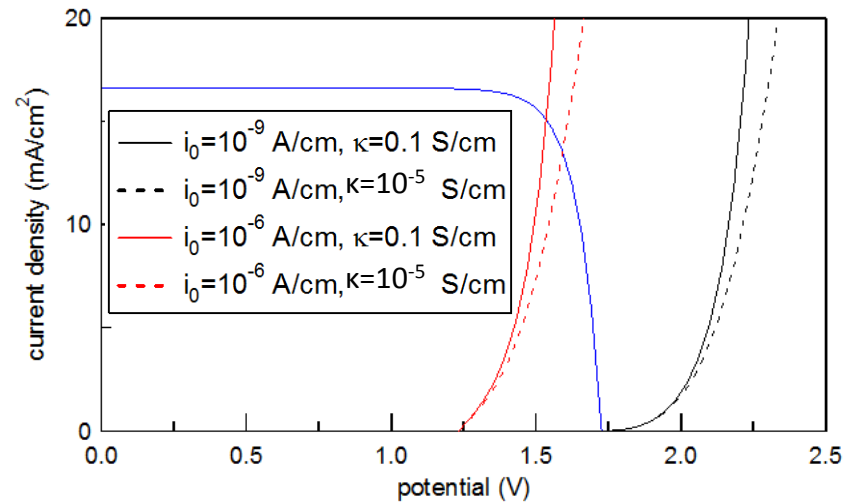
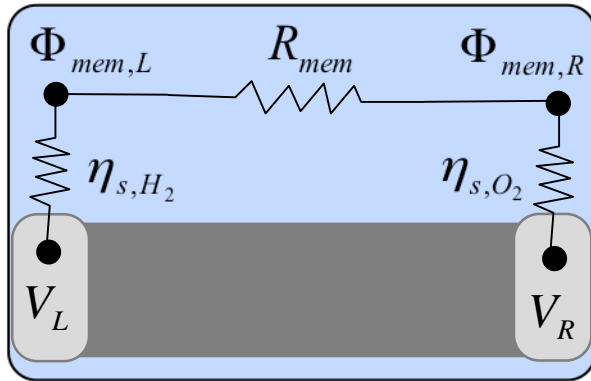
WSe₂



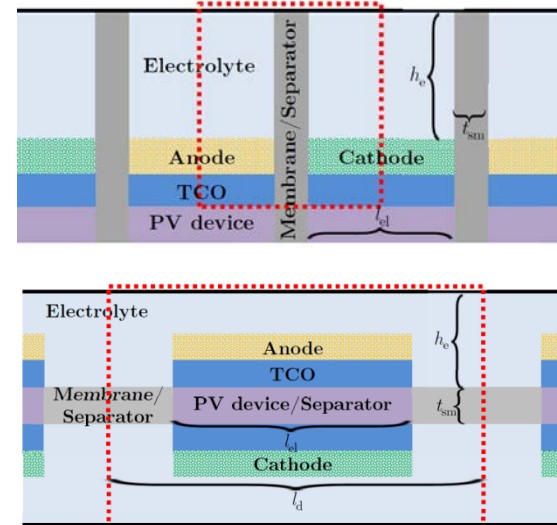
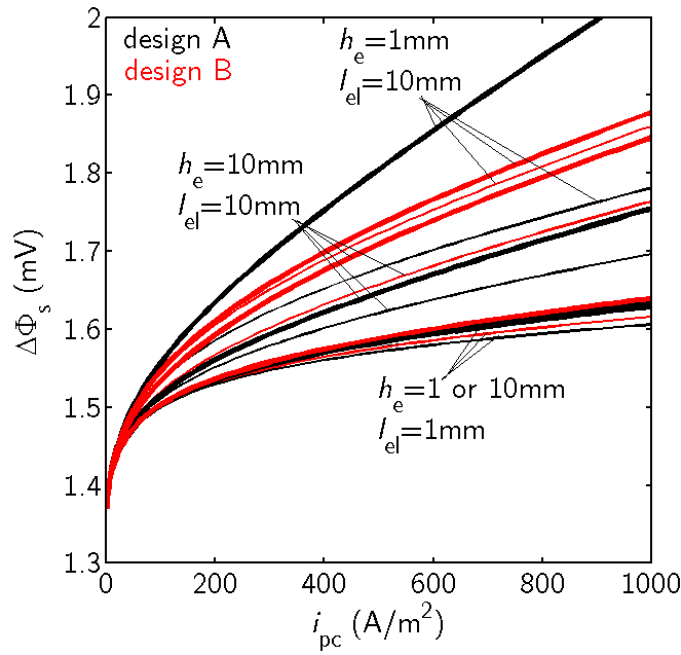
Generates H₂ at 3-5% conversion efficiency in acid or base!

Identifying design requirements through device simulation

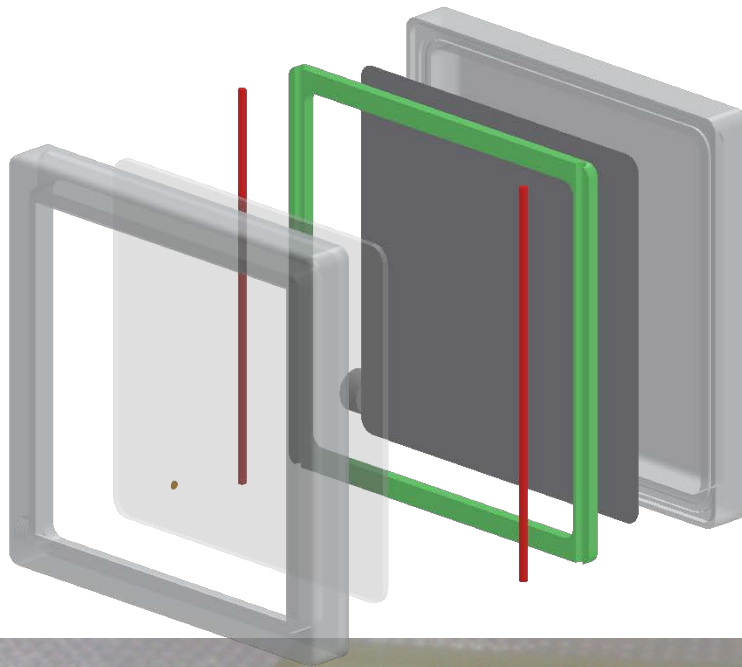
Sensitivity analysis determines acceptable range of membrane conductivity



Comparison and optimization of proposed device architectures



Absorber-in-Membrane Design



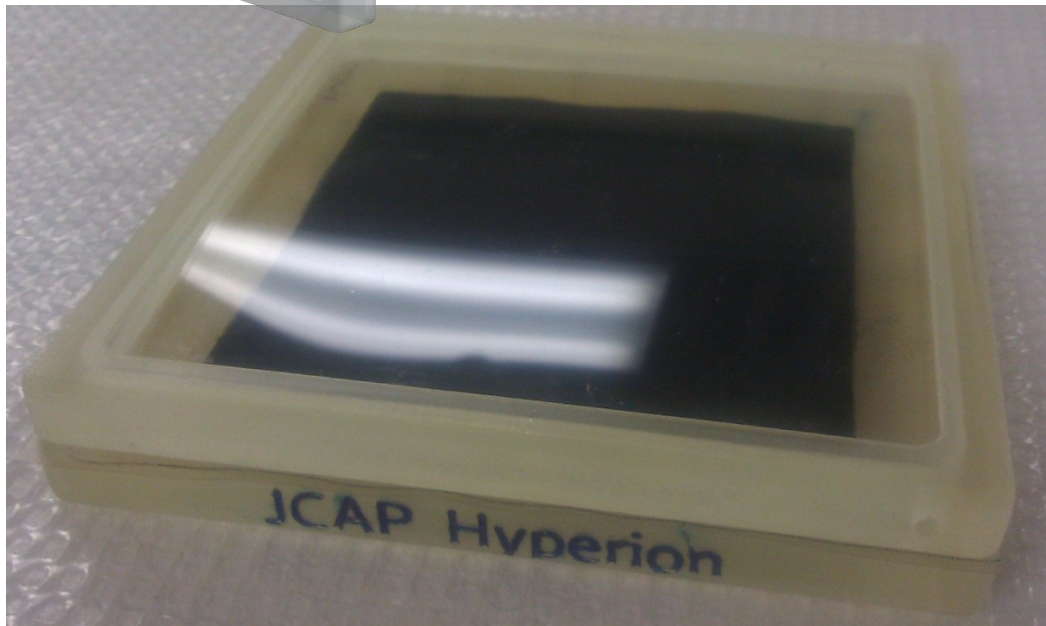
Membrane: H_2 -reduced or metal doped $SrTiO_3$ dispersed in Nafion. $SrTiO_3$ on H_2 side metallized with thin Pt film.

Based on (for example):

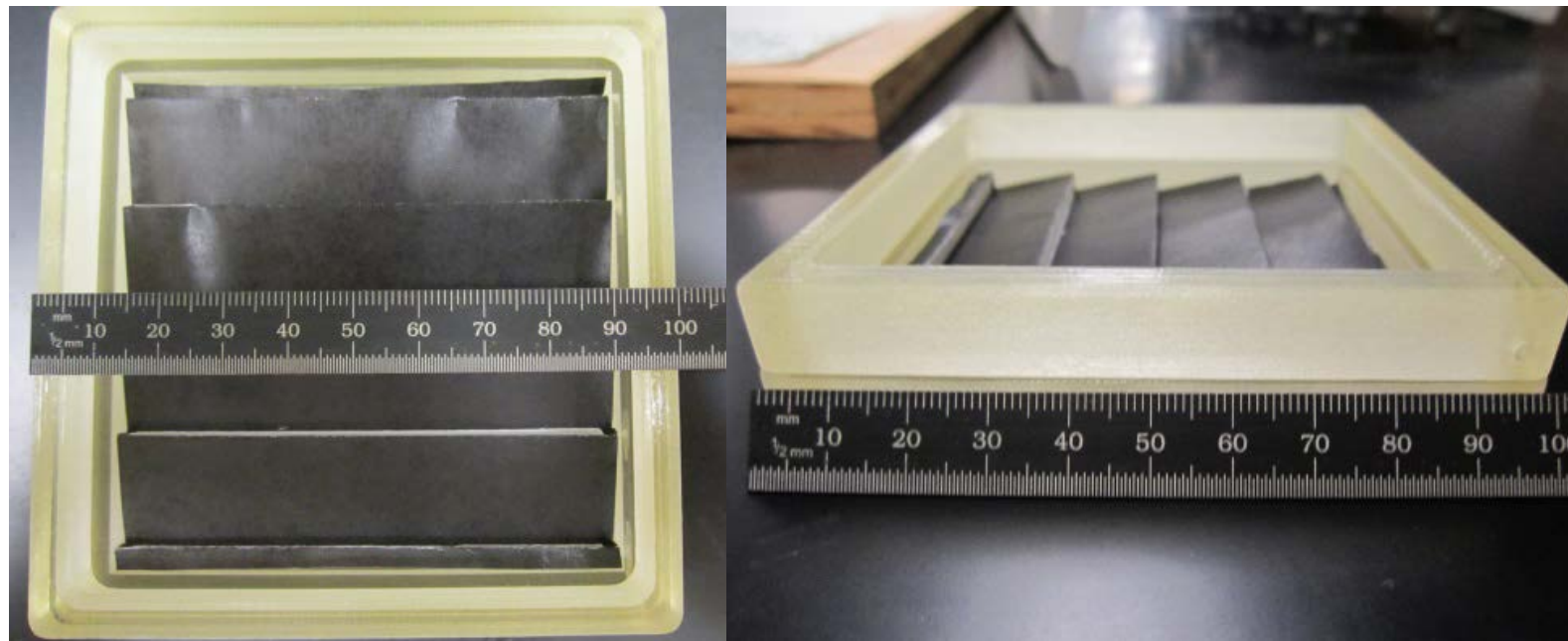
“Photoelectrolysis of water in cells with $SrTiO_3$ anodes”
J. G. Mavroides, J. A. Kafalas, and D. F. Kolesar, *Appl. Phys. Lett.*, **28**, 241 (1976).

Absorber-in-membrane chassis with commercial MEA (left).

JCAP-prepared H_2 -reduced $SrTiO_3$ -Nafion membrane (below).



- Top and side view of the shared square chassis of the absorber-in-membrane and the PV-based prototype (PV-based interior shown)
- Ruler indicates the prototypes housed in this chassis will be 10x10cm



Top (shown left) and side (shown right) view of the PV-based prototype measured with a ruler

Foresightful Energy Analysis



- **We are like tenant farmers chopping down the fence around our house for fuel when we should be using Nature's inexhaustible sources of energy — sun, wind and tide. ...**
- “Sunshine is spread out thin and so is electricity. Perhaps they are the same, but we will take that up later. Now the trick was, you see, to concentrate the juice and liberate it as you needed it. The old-fashioned way inaugurated by Jove, of letting it off in a clap of thunder, is dangerous, disconcerting and wasteful. It doesn’t fetch up anywhere. My task was to subdivide the current and use it in a great number of little lights, and to do this I had to store it. And we haven’t really found out how to store it yet and let it off real easy-like and cheap. Why, we have just begun to commence to get ready to find out about electricity. This scheme of combustion to get power makes me sick to think of—it is so wasteful. It is just the old, foolish Prometheus idea, and the father of Prometheus was a baboon.”

Foresightful Energy Analysis



- I'd put my money on the sun and solar energy. What a source of power!
- I hope we don't have to wait until oil and coal run out before we tackle that..

Foresightful Energy Analysis



- **I'd put my money on the sun and solar energy. What a source of power!**
- **I hope we don't have to wait until oil and coal run out before we tackle that..**

- **Thomas A. Edison, 1931**

Acknowledgments

