

Free electron laser nitriding of metals: from basic physics to industrial applications

Peter Schaaf ^{a)}, M. Shinn ^{b)}, E. Carpene ^{a)}, J. Kaspar ^{c)}



GEORG-AUGUST-UNIVERSITÄT
GÖTTINGEN
Zweites Physikalisches Institut

a)

Jefferson Lab



b)

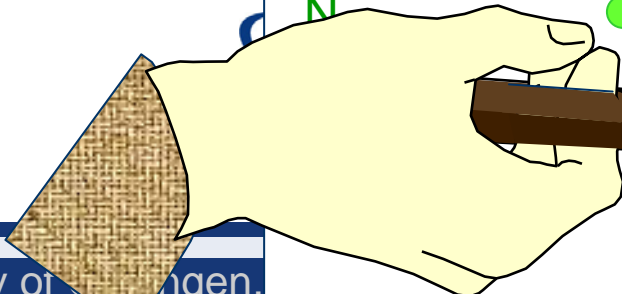
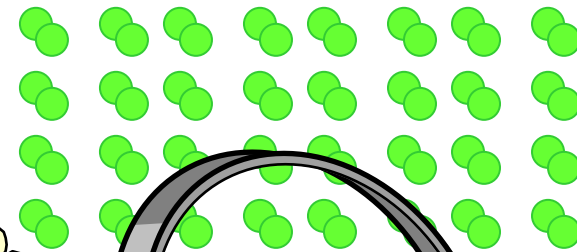
IWS

Fraunhofer
Institut
Werkstoff- und
Strahltechnik

c)

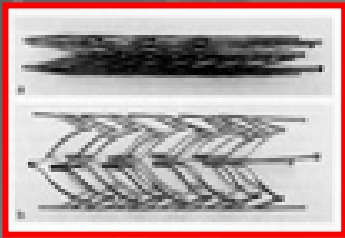
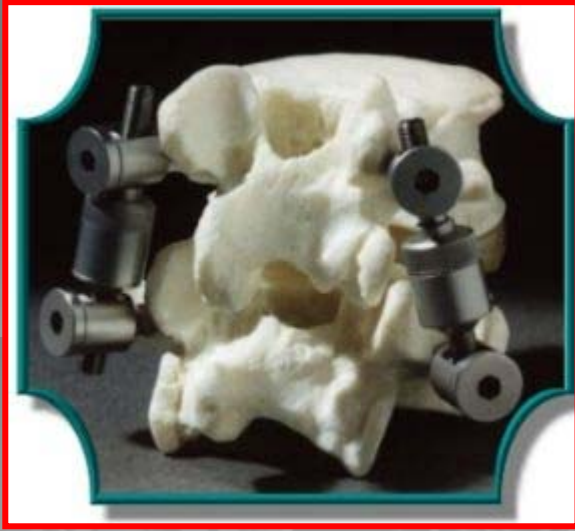
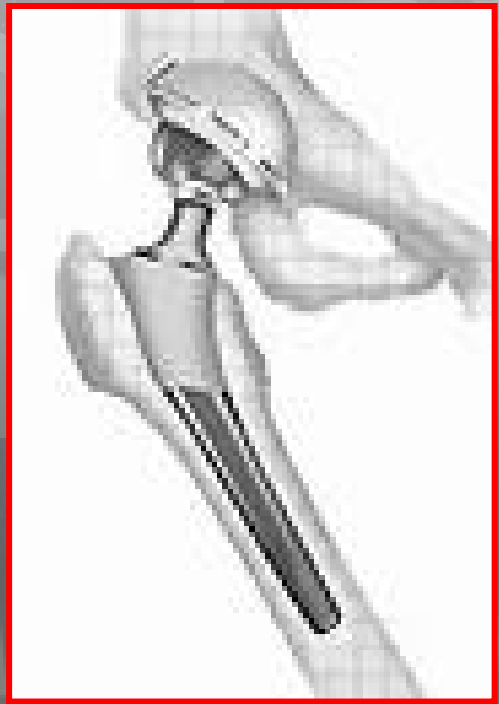
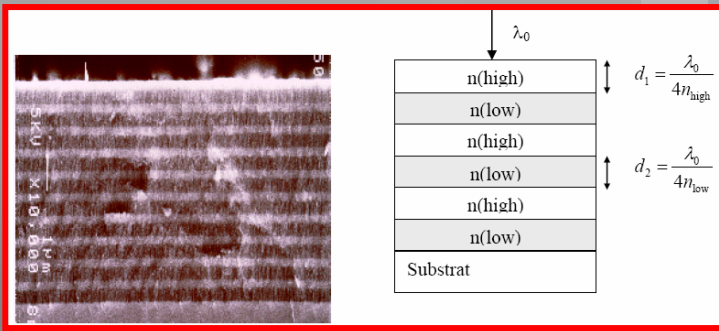
Laser Nitriding

Reactive or
non-reactive
atmosphere



Film or Functional
Coating: made fast

Substrate, work piece



Precision Components

- Laser Synthesis of Thin Films and Coatings (Nitriding, Carburizing, Hydriding): experimental principles, interactions, melt, plasma, dynamics, diffusion, solidification, ...
 - Fe-N and Fe-C,
 - Austenitic stainless steel
 - **TiN** and TiC
 - AlN and AlC
 - Si₃N₄ and SiC (IBM-Milliped)
 - Laser-Conditioning of Magnesium
 - Laser-Hydriding Ti-H
 - Production pc-a:Si(H) (TFT)
 - β-FeSi₂ (photovoltaics, optoelectr.)
 - Fe/Ag Multilayers by PLD (GMR, TMR)
 - Polymer-PLD (Applications)
 - Epitaxial recrystallisation (SiC, SiO₂)

- Excimer Laser 55 ns
- Nd:YAG Laser 8 ns
- **FEL** 1 ps
- Ti:Sapphire 150 fs

Basic Physics

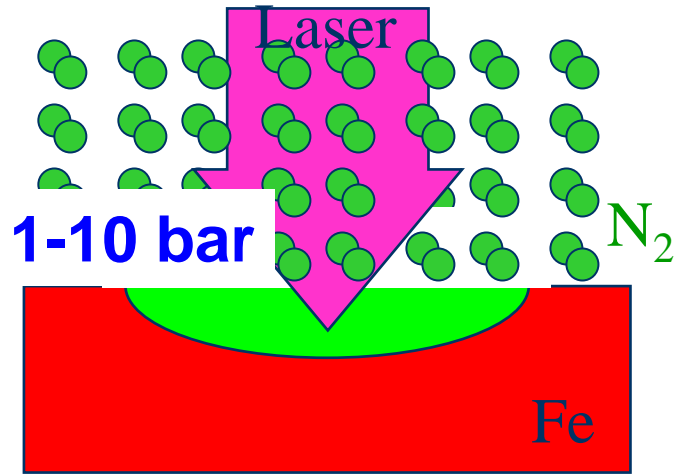
A large, faint watermark of the University of Göttingen seal is visible in the background on the right side of the slide. The seal features a central figure holding a staff, surrounded by a circular border with Latin text.



Laser Synthesis: temperature, plasma

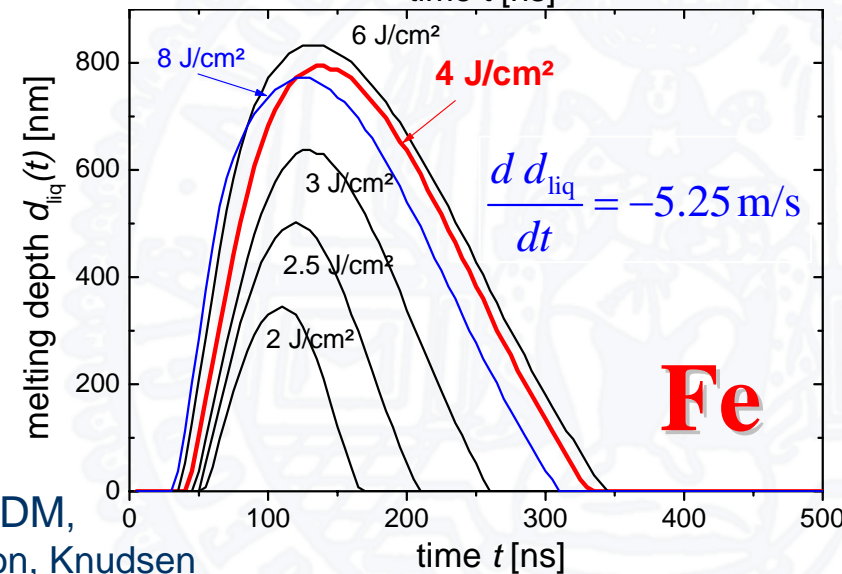
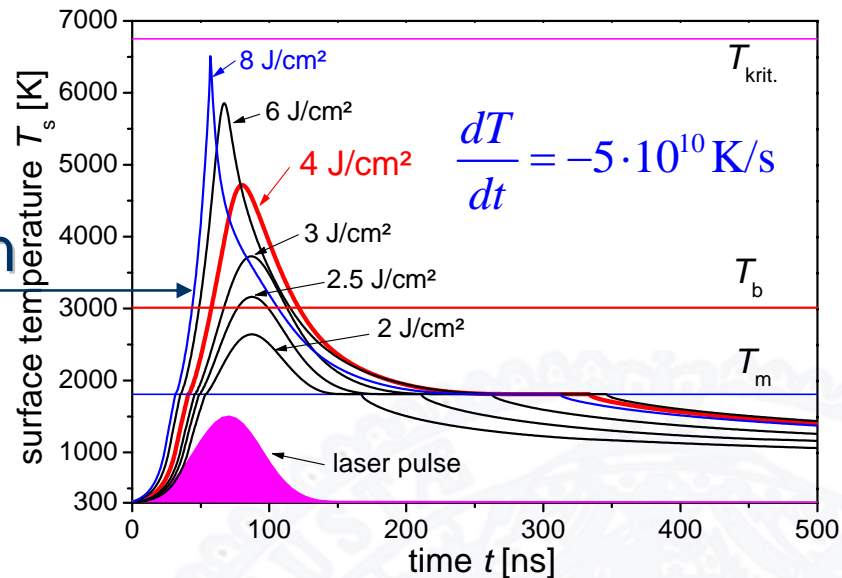
Excimer Laser: $\lambda=308$ nm (4.03 eV)
 $t_p=55$ ns (FWHM), 5×5 mm²,
 $H=4$ J/cm² ($I_0=70$ MW/cm²)

gas transparent for laser



evaporation
ablation
plasma-
formation
vacuum:
PLD

but: ambient atmosphere at high pressure
prevents ablation, causes high
pressures, chemical reactions,
take-up into liquid surface, re-solidification,
coating forms



Numerical: FDM,
incl. evaporation, Knudsen

Process: Principle of Laser Synthesis

Hom. Beam

0.1 MPa

0.9 MPa

10

dynamics
sure,
s (LSAW)
tion Wave
on Wave

Plasma (600 ns) :

Laser pulse: 55 ns

Fe, 1 bar N₂, 4 J/cm²

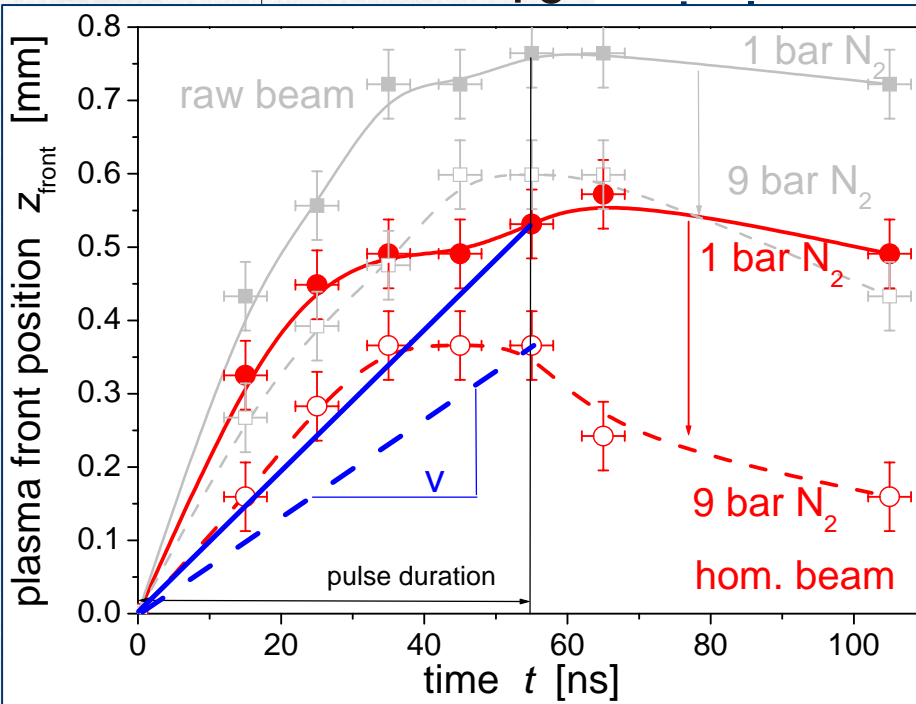
Fe, 9 bar N₂, 4 J/cm²

surface

surface

LSD-model

$$= \left[2(\gamma^2 - 1) \cdot \frac{I_0}{\rho_0} \right]^{1/3}$$



Theory

150

Experiment

1 bar : $v^{\text{LSD}} = 10.4$ km/s

9 bar : $v^{\text{LSD}} = 5.0$ km/s

350

1 bar : $v^{\text{LSD}} = 9.7(16)$ km/s

9 bar : $v^{\text{LSD}} = 6.6(16)$ km/s

increased gas pressure \Rightarrow plasma expansion more slowly – Plasma pressure higher and lasts longer!

p_p : 1 bar : 500 bar – 9 bar : 1050 bar



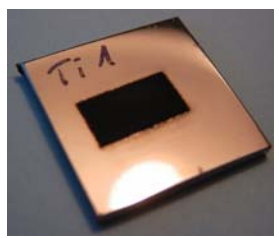
- Irradiation of Ti in N₂
- Free-Electron Laser FEL



Overview: TiN coatings

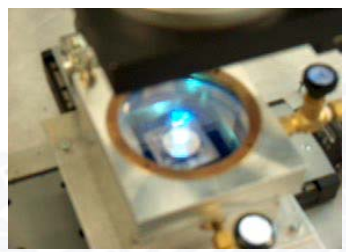
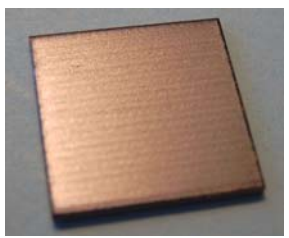
Ti:Sapphire+CPA

750 nm
150 fs



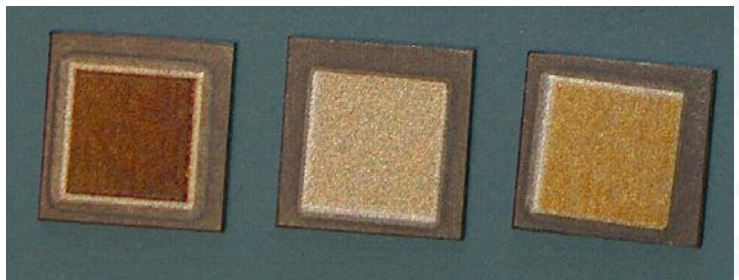
Excimer Laser

308 nm
55 ns



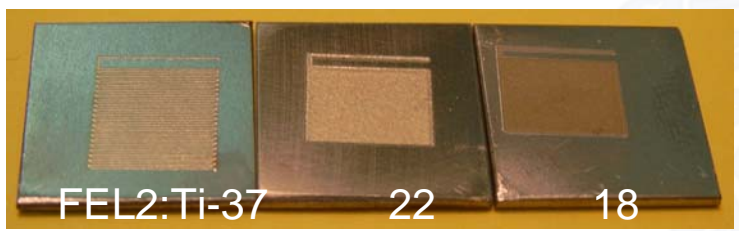
Nd:YAG

1064 nm, 532 nm
6 ns



FEL

3100 nm, 1050 nm
< 1 ps

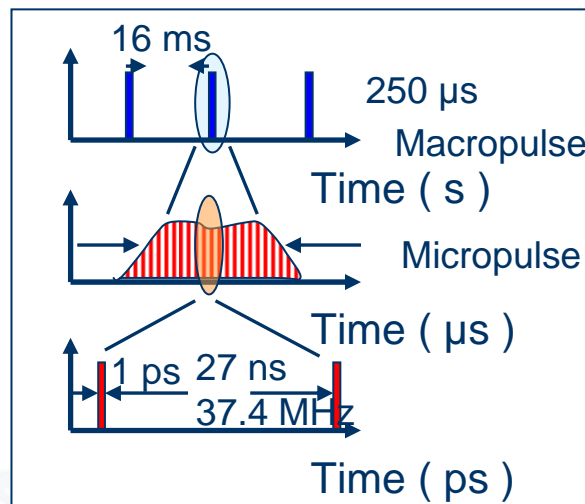
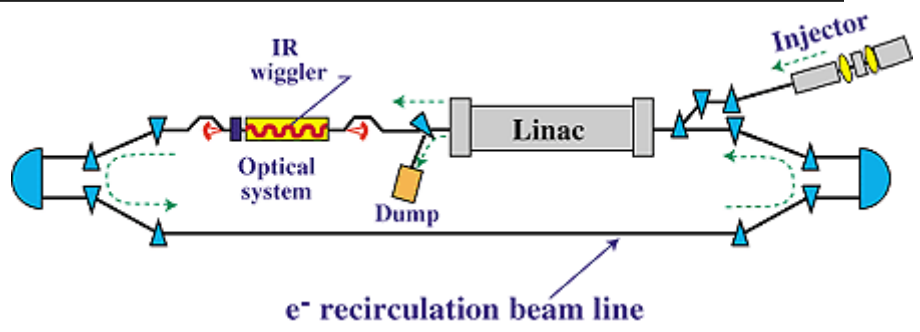
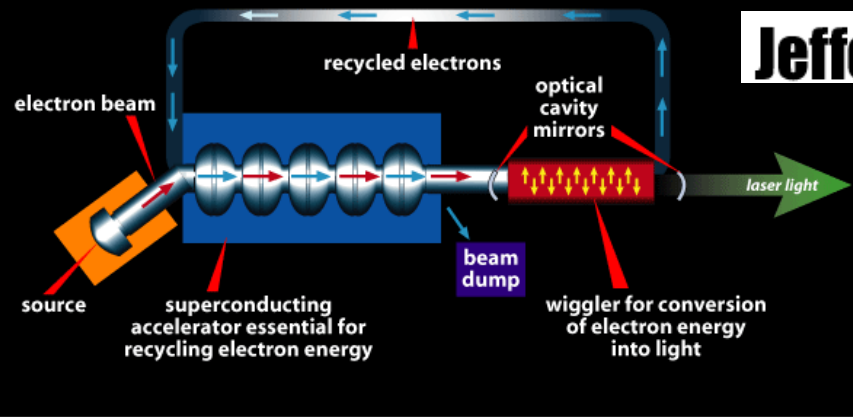




Faster and better with FEL ?

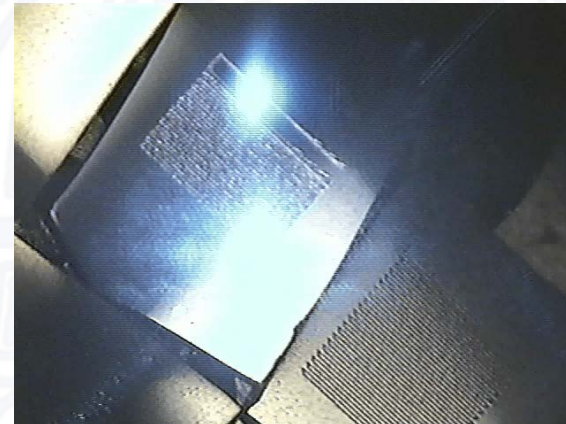
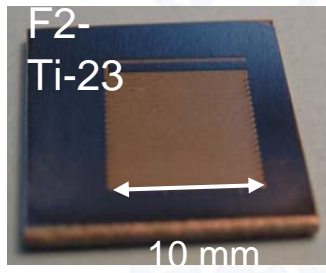
Jefferson Lab

FEL
Newport News,
Virginia, USA

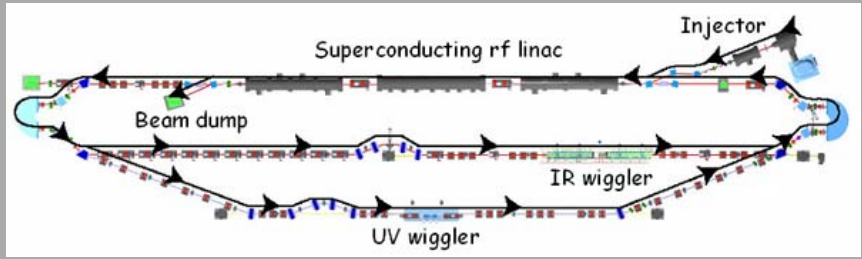


flexible in timing

3.1 μm
2 kW



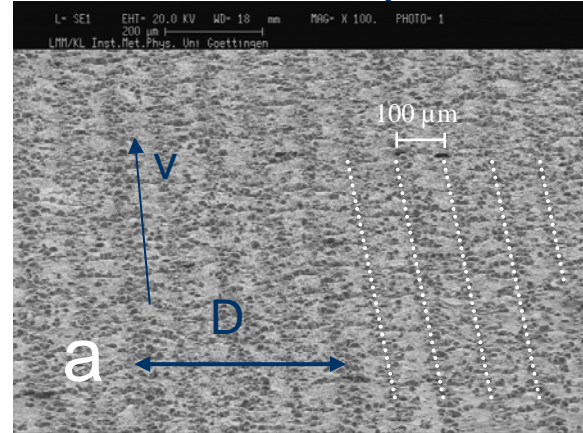
upgrade: 10 kW, UV (300 nm), 2005





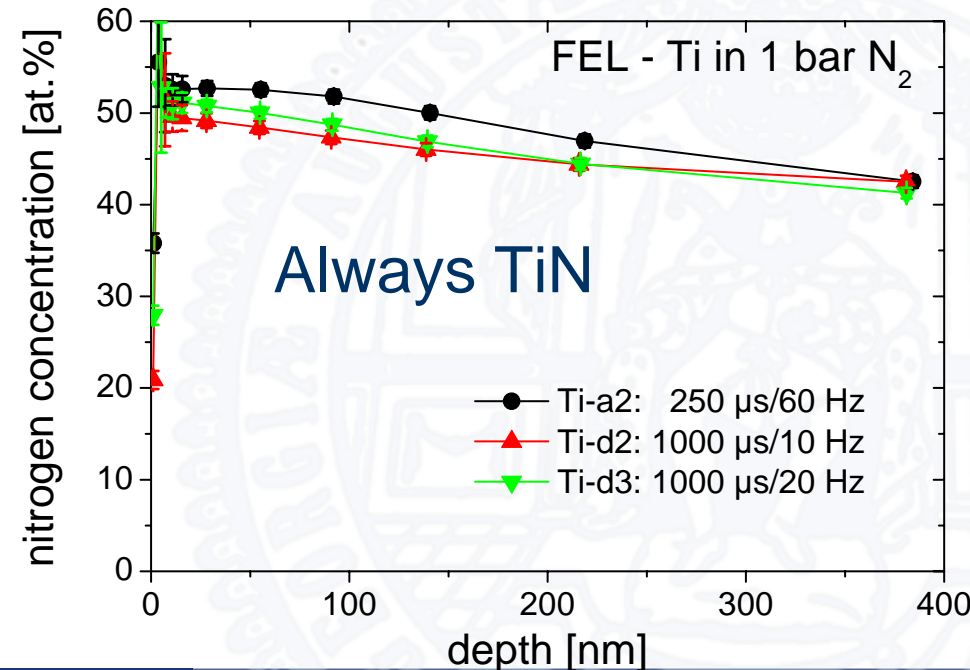
FEL: TiN Synthesis

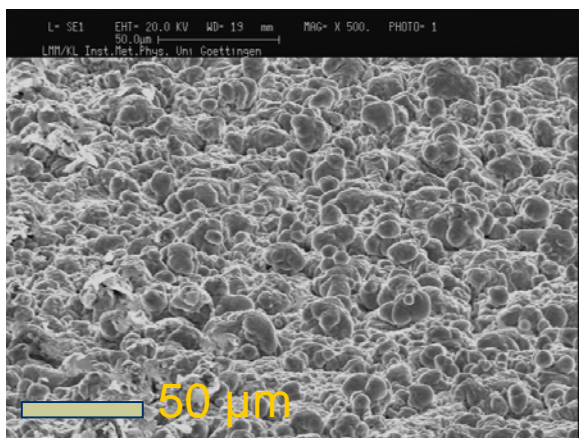
Line scan: velocity $u=0.5$ mm/s, line width $D=0.4$ mm, shift δ (50, 100, 200 μm)



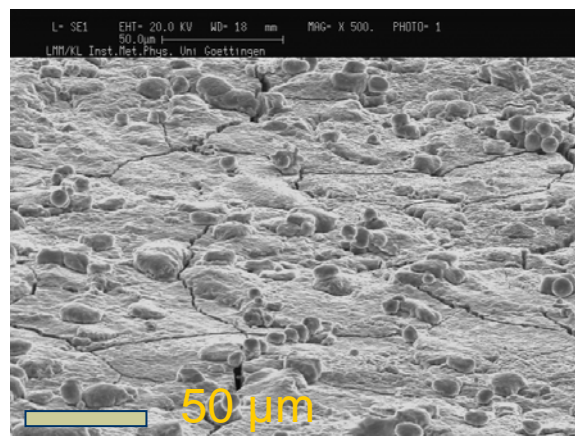
- formation of TiN
- concentration gradient
- independent of parameters
- structure of surface?

Sample	Macro t_m (μs)	Macro f_m (Hz)	shift δ (μm)	Fluence ϕ_m (J/cm^2)
Ti-a1	250	60	200	123
Ti-a2	250	60	100	123
Ti-a3	250	60	50	123
Ti-b1	500	30	100	246
Ti-c1	750	30	100	369
Ti-d1	1000	10	200	492
Ti-d2	1000	10	100	492
Ti-d3	1000	20	200	492
Ti-d4	1000	30	200	492
Ti-d5	1000	30	100	492

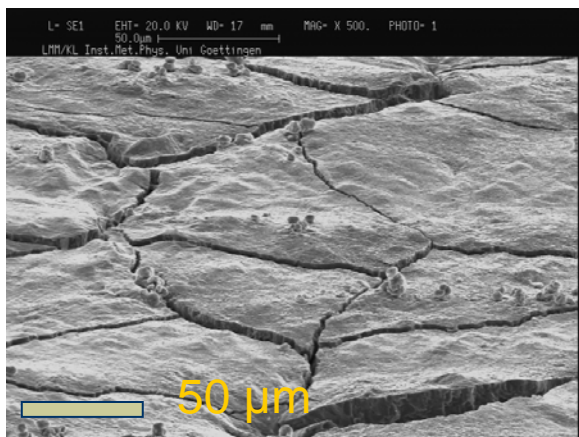




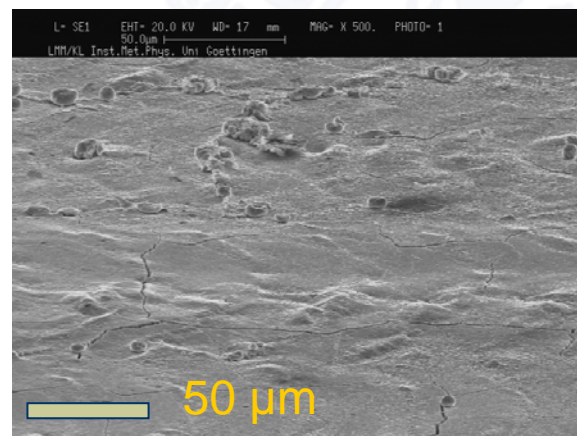
a2: 250 μ s, 60 Hz, 100 μ m



c1: 750 μ s, 30 Hz, 100 μ m

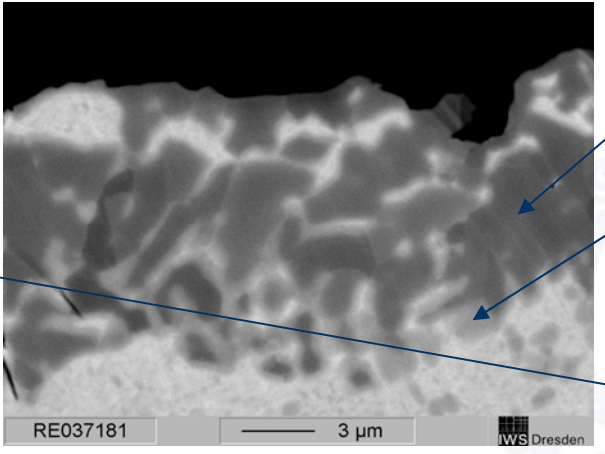
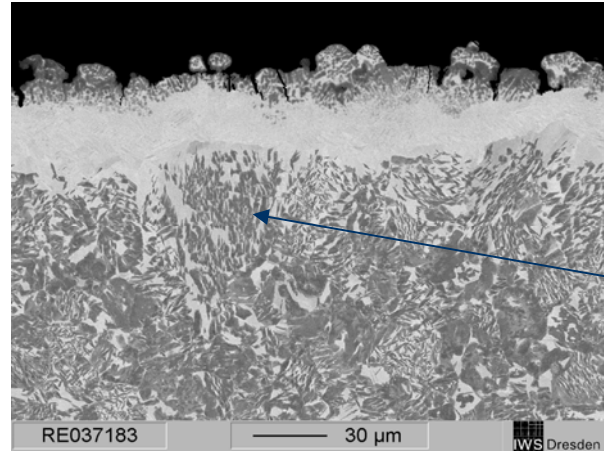
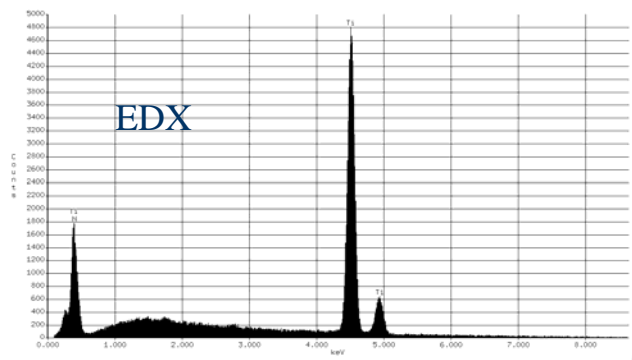
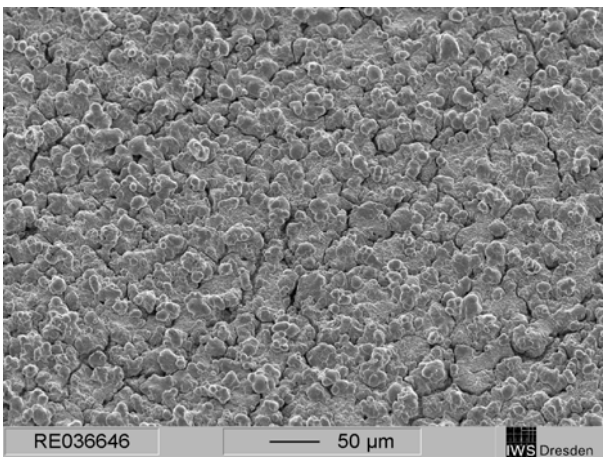
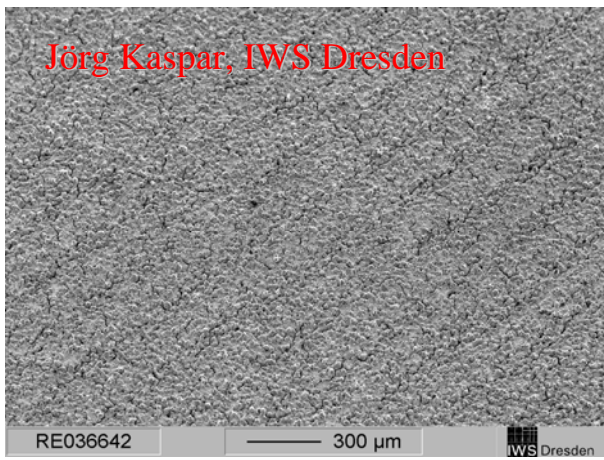


d5: 1000 μ s, 30 Hz, 100 μ m



d1: 1000 μ s, 10 Hz, 200 μ m

Jörg Kaspar, IWS Dresden



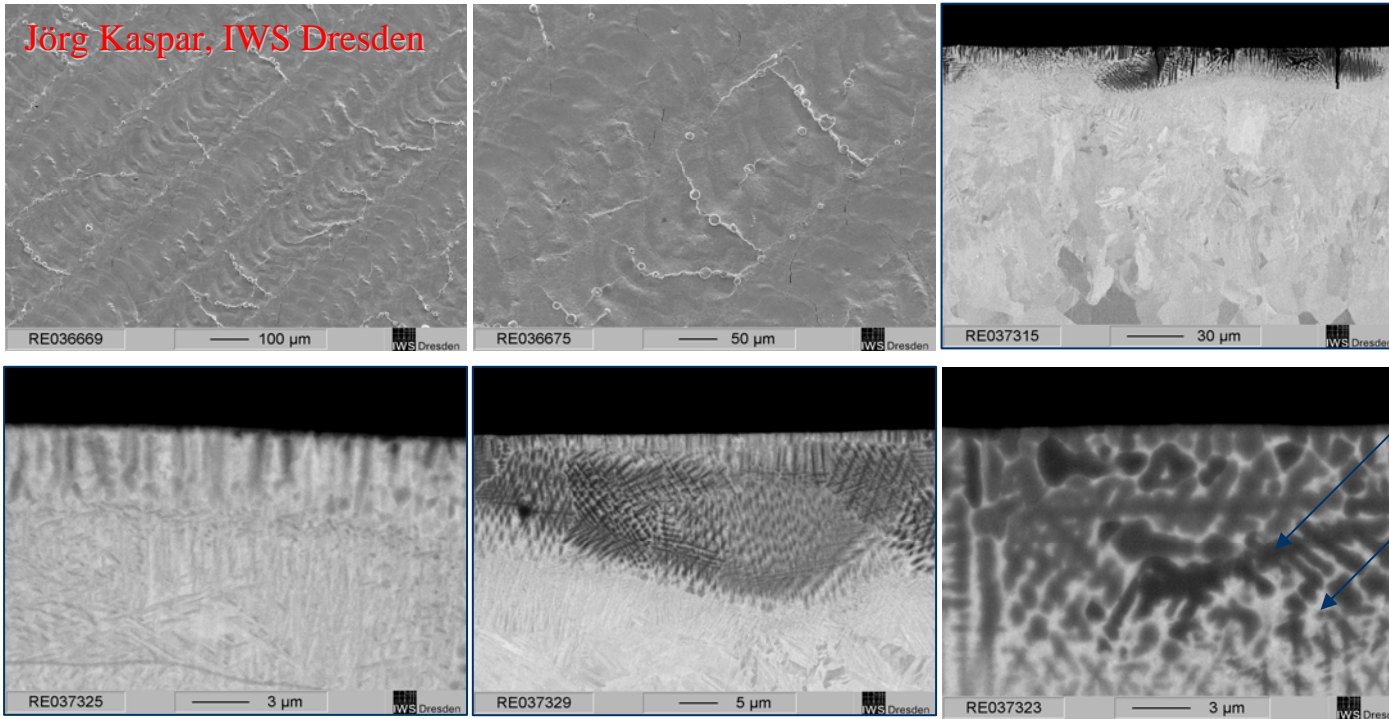
~ 23 at.% N

~ 16 at.% N

~ 8 at.% N

Surface very rough, melting pearls, network of fine cracks, melting depth 30-40 μ m, TiN 5-15 μ m, primary solidification of TiN at the surface, TiN has a nitrogen rich kernel and less nitrogen cover, α' -Martensite in between

Jörg Kaspar, IWS Dresden



EDX: ~30 at% N

EDX: ~10 at% N

melting zone 20-30 μ m, TiN 0-25 μ m,

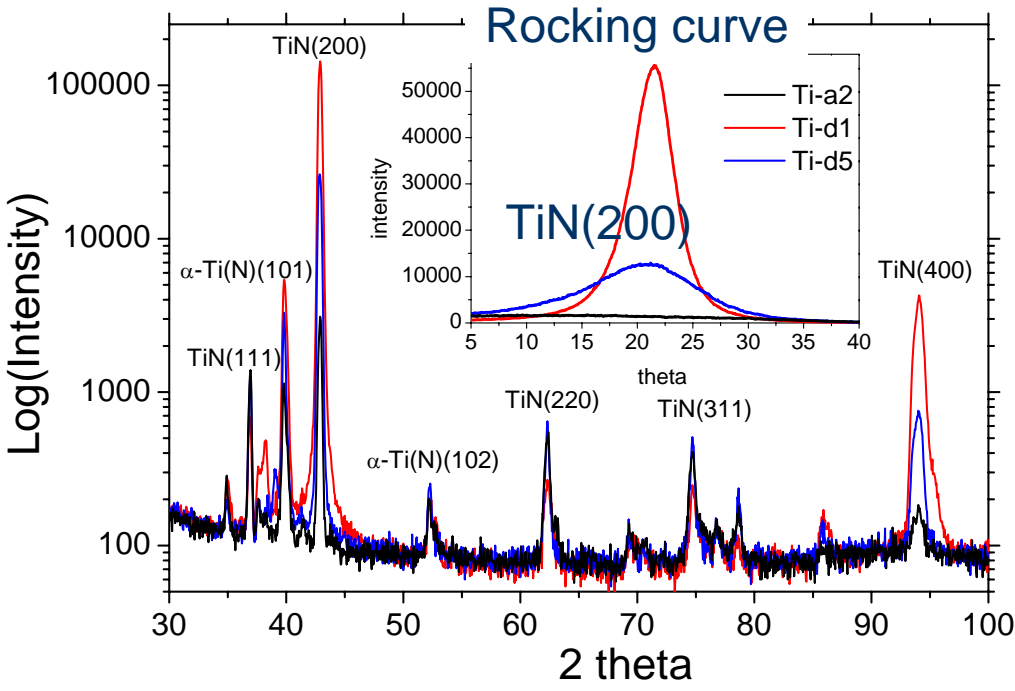
Very smooth surfaces, very few melt pearls, significant solidification lines, fine cracks.

cracks only within TiN. TiN cover smaller.

TiN perpendicular to the surfaces, dendritic solidification



GIXRD: Texture, Rocking curves



1000 μ s,
10 Hz,
200 μ m

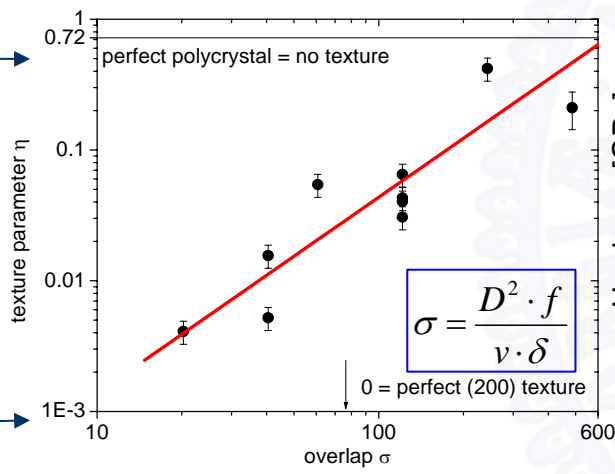
Strong texture for long macropulses and small overlap

1000 μ s,
30 Hz,
100 μ m

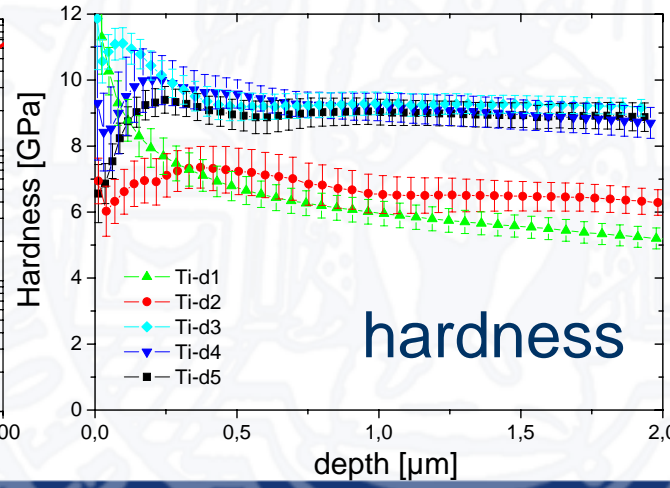
no texture for short macropulses and large overlap

250 μ s,
60 Hz,
100 μ m

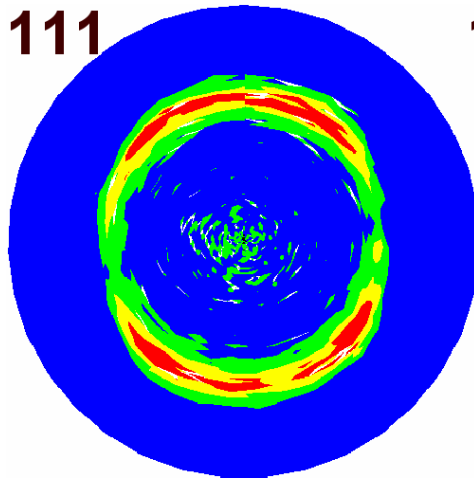
no texture (polycryst.) \rightarrow



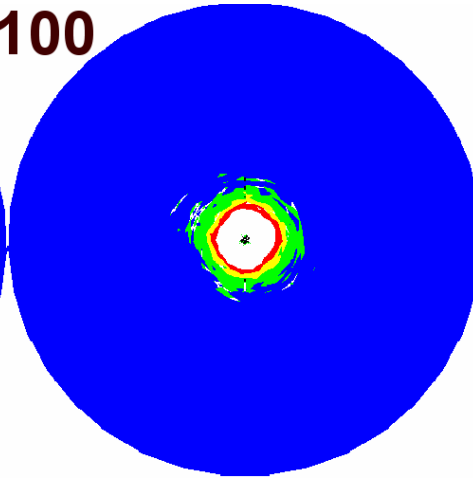
fully textured (single cryst.) \rightarrow



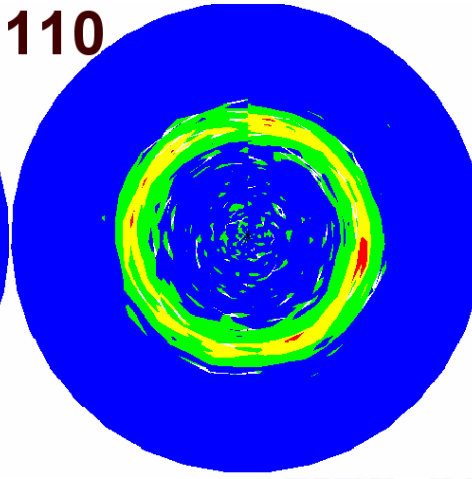
111



100



110

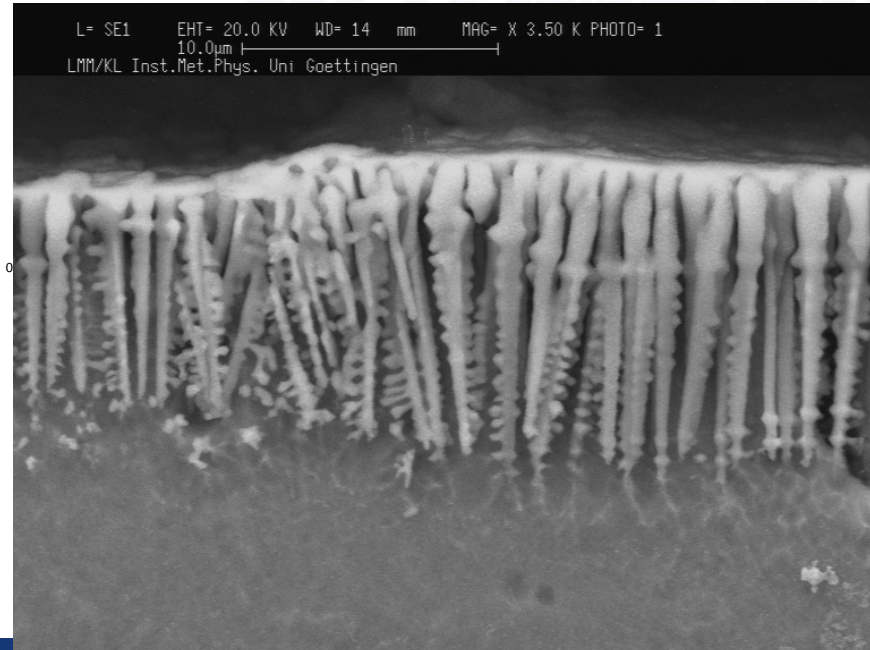
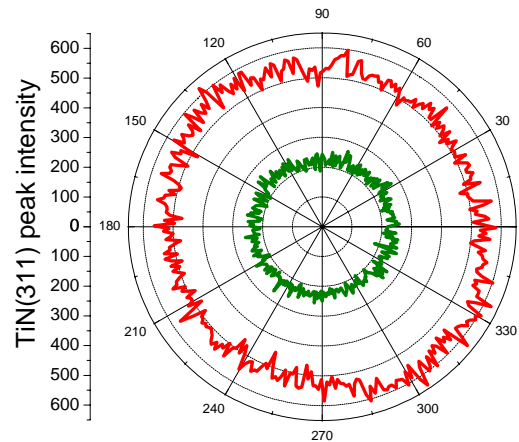
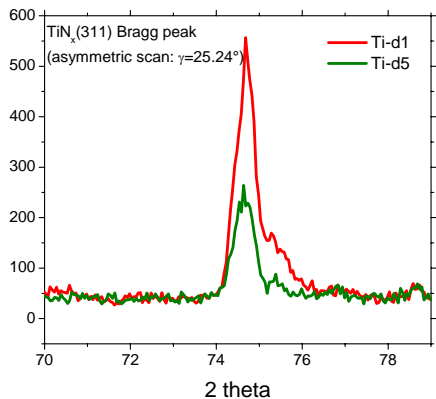


Ti-d3 (Ti-25)

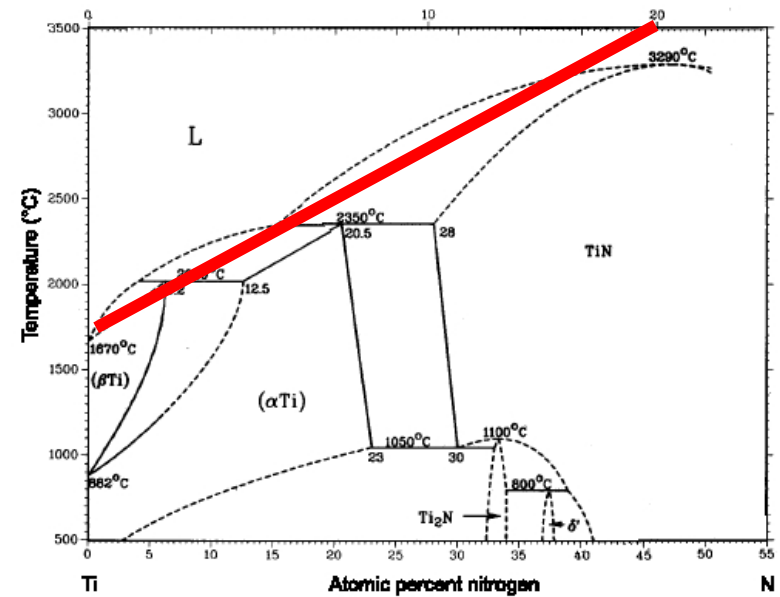
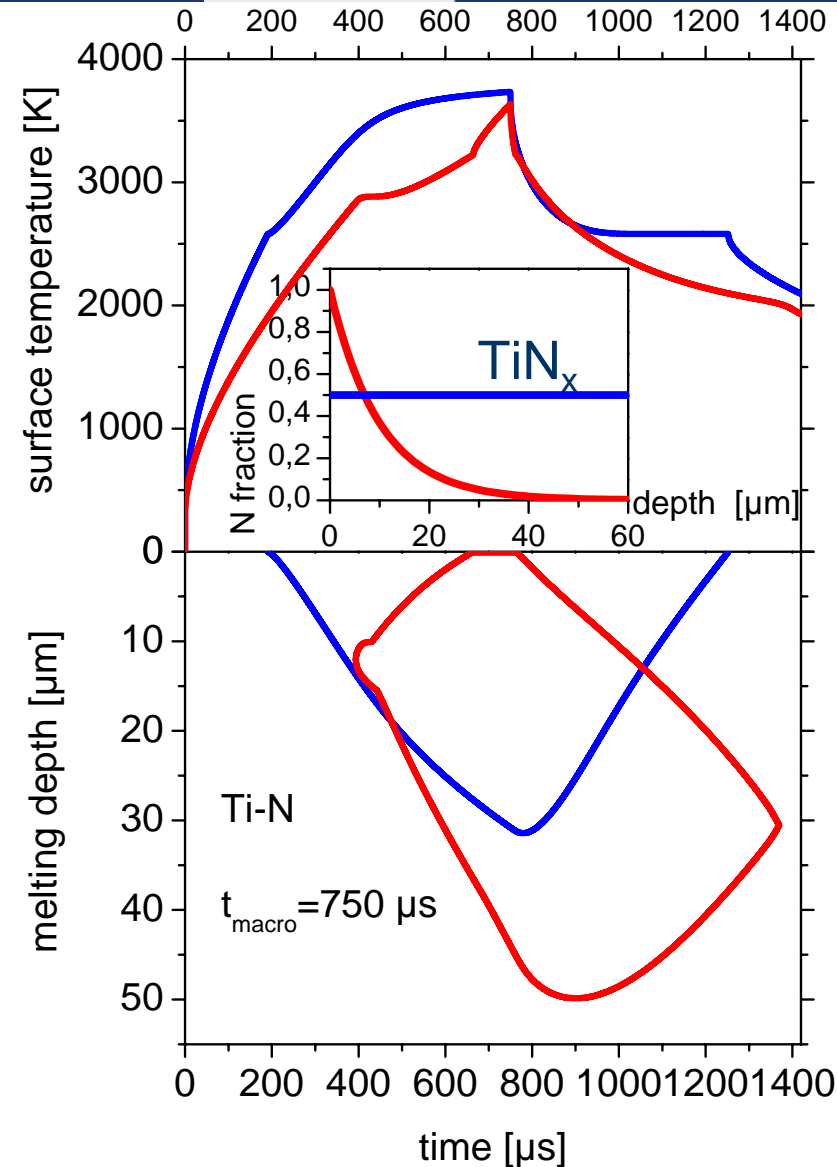
scan: $5^\circ \times 5^\circ$

φ : 0-360°

χ : 0-80°



Symmetric -> columnar growth, fiber texture
very strong texture, well aligned columns



Strong dependence of the melting temperature on the nitrogen content

nitrogen concentration gradient:

re-solidification starts at surface

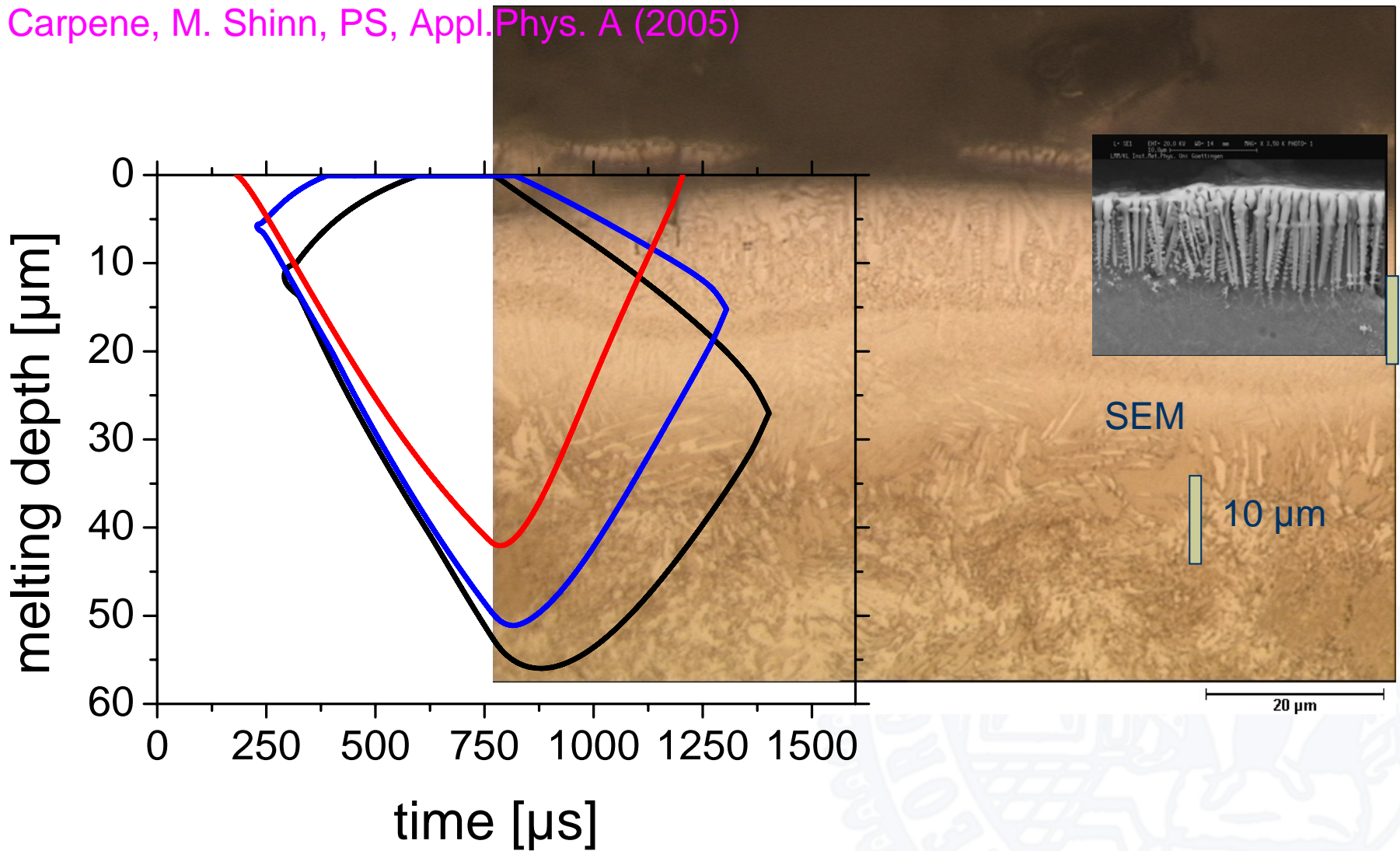
free (200) surface is most favorable



Comparison: Simulation and cross section

E. Carpene, PS, *MRS Proc.* **780** (2003) Y5.8.1

E. Carpene, M. Shinn, PS, *Appl. Phys. A* (2005)

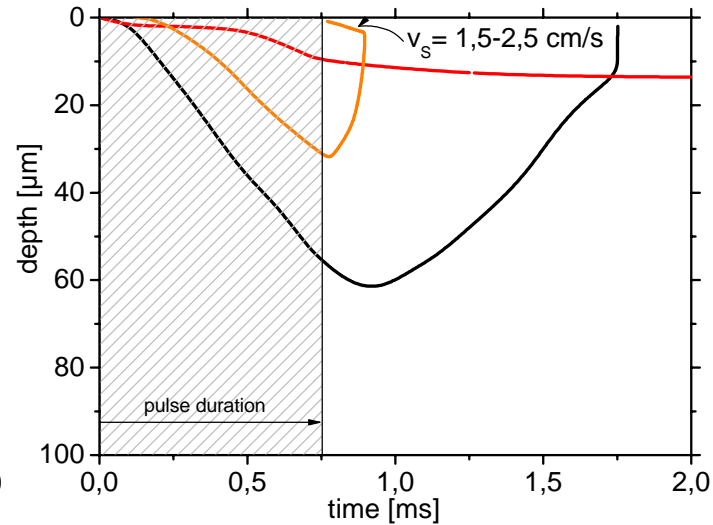
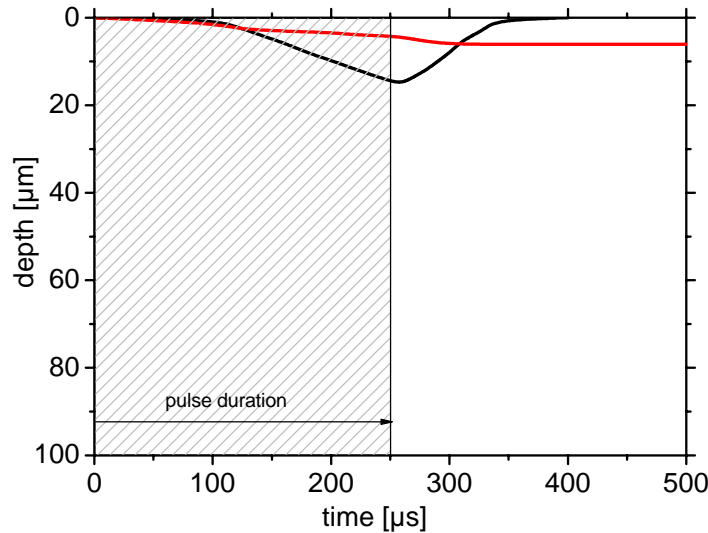


melting depth:

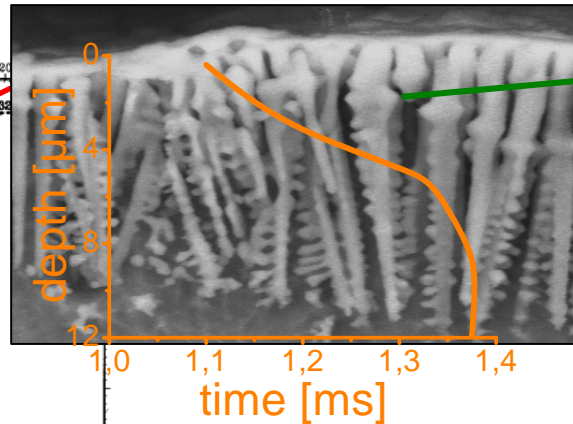
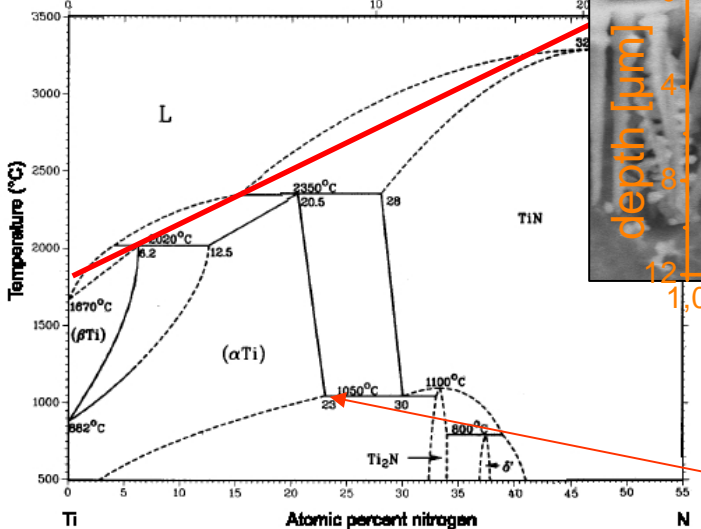
- Ti
- TiN

diffusion depth (1/e):

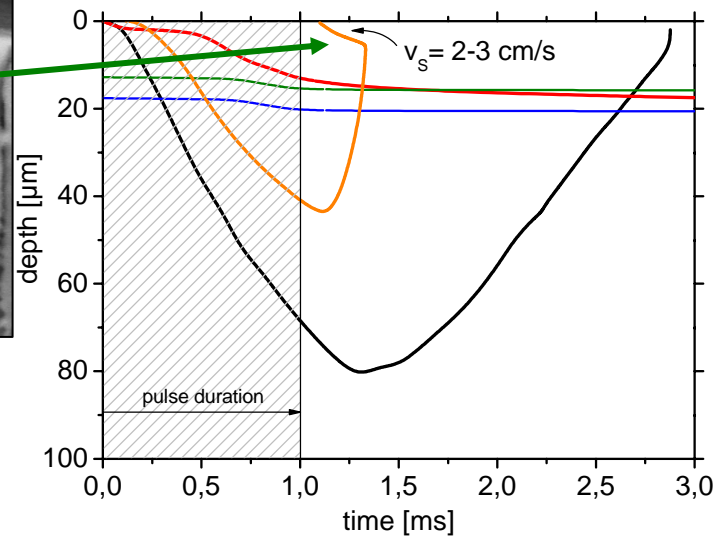
- 1 Pulse
- 2 Pulses
- 2 Pulses (corr.)



Phase diagramme:

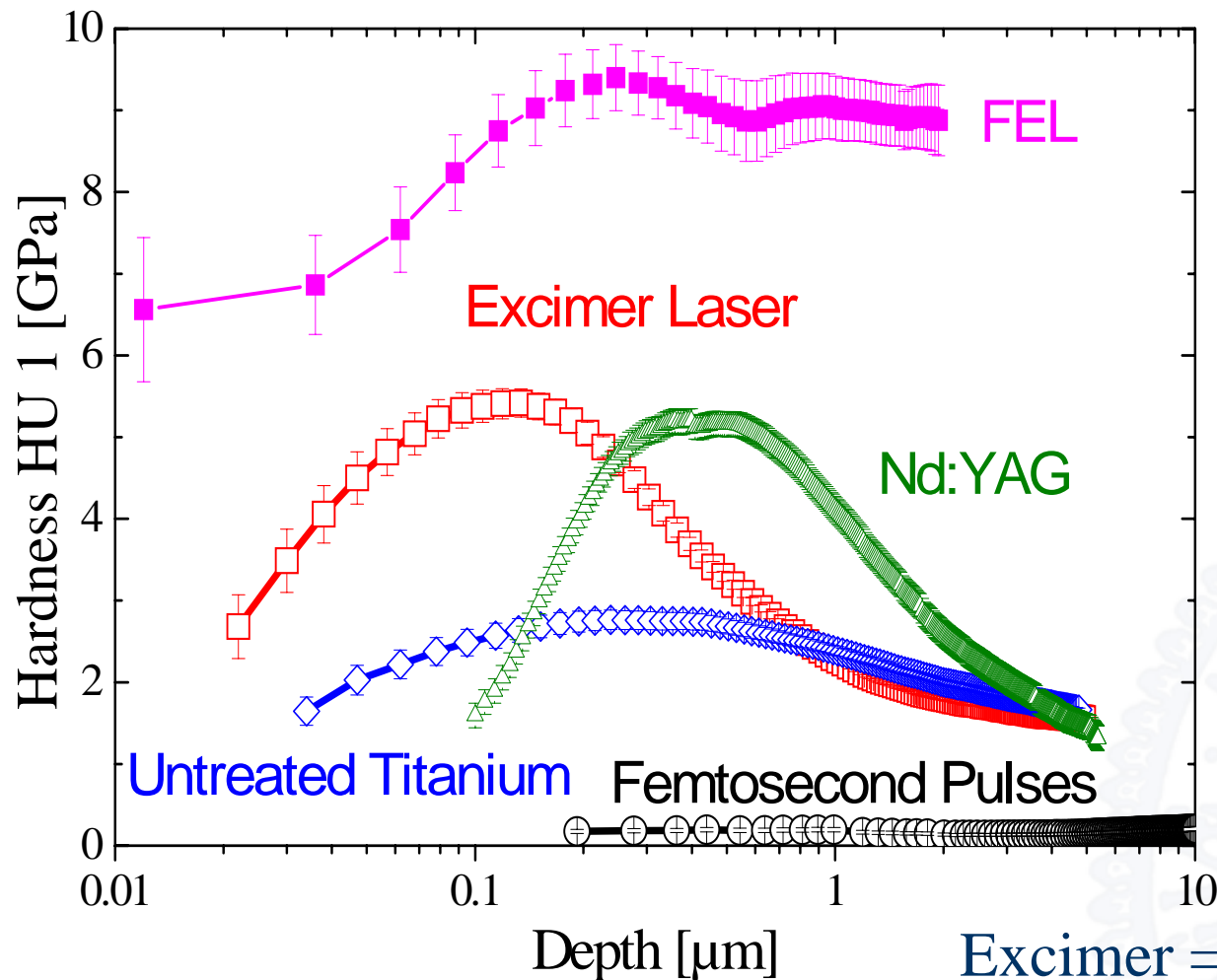


$T_m = f(N\%)$
23% solubility





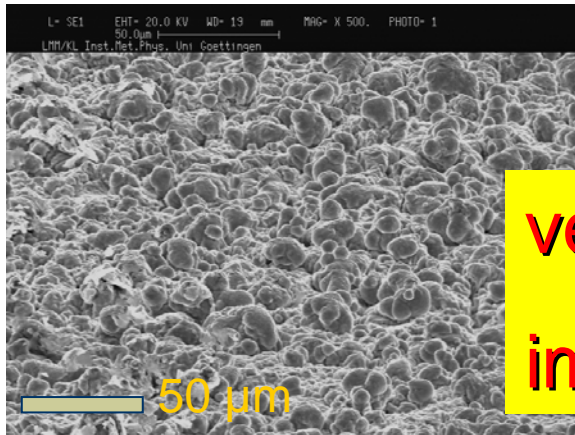
TiN: hardness – comparison laser



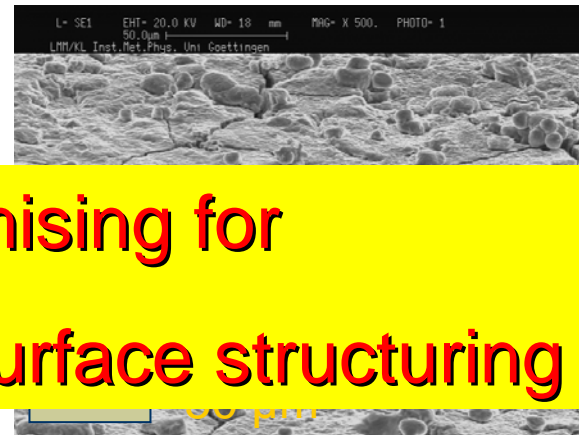
Femtoseconds – very soft (nanocluster)

FEL highest hardness and thickest coating

Excimer = „smooth“ ($R_a < 0.5 \mu\text{m}$)
Nd:YAG: $R_a \sim 1\text{-}2 \mu\text{m}$
FEL: smooth (cracks)



a2: 250 μ s, 60 Hz, 100 μ m

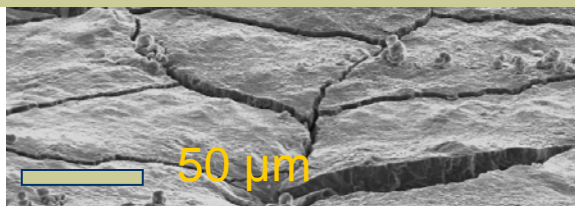


c1: 750 μ s, 30 Hz, 100 μ m

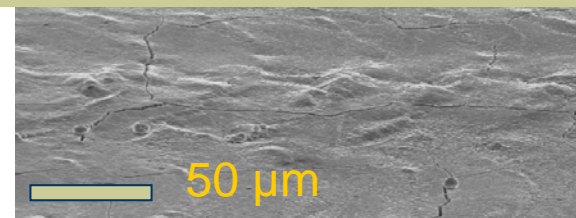
**very promising for
implant surface structuring**

Kieswetter K, Schwartz Z, Hummert TW, Cochran DL, Simpson J, Dean DD, Boyan BD.
Surface roughness modulates the local production of growth factors and cytokines by osteoblast-like MG-63 cells.

Journal of Biomedical Material Research 1996; 32 (1): 55-63.



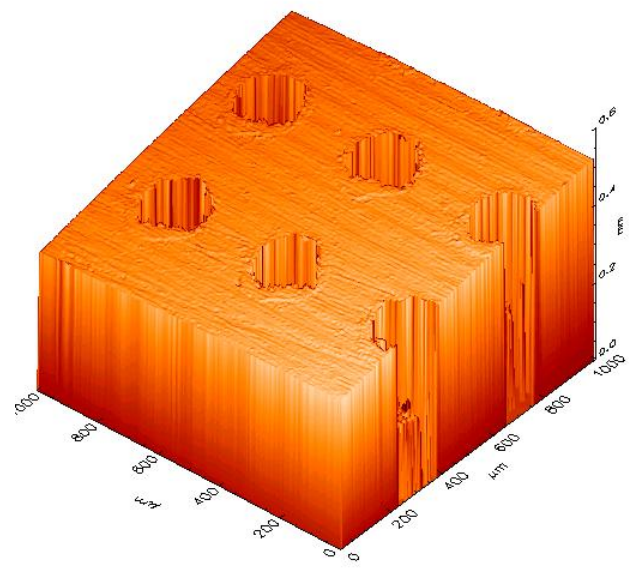
d5: 1000 μ s, 30 Hz, 100 μ m



d1: 1000 μ s, 10 Hz, 200 μ m



Laser-Structuring of an hip-joint



3D image of a laser structured hip joint (drilling holes of $D=200 \mu\text{m}$)

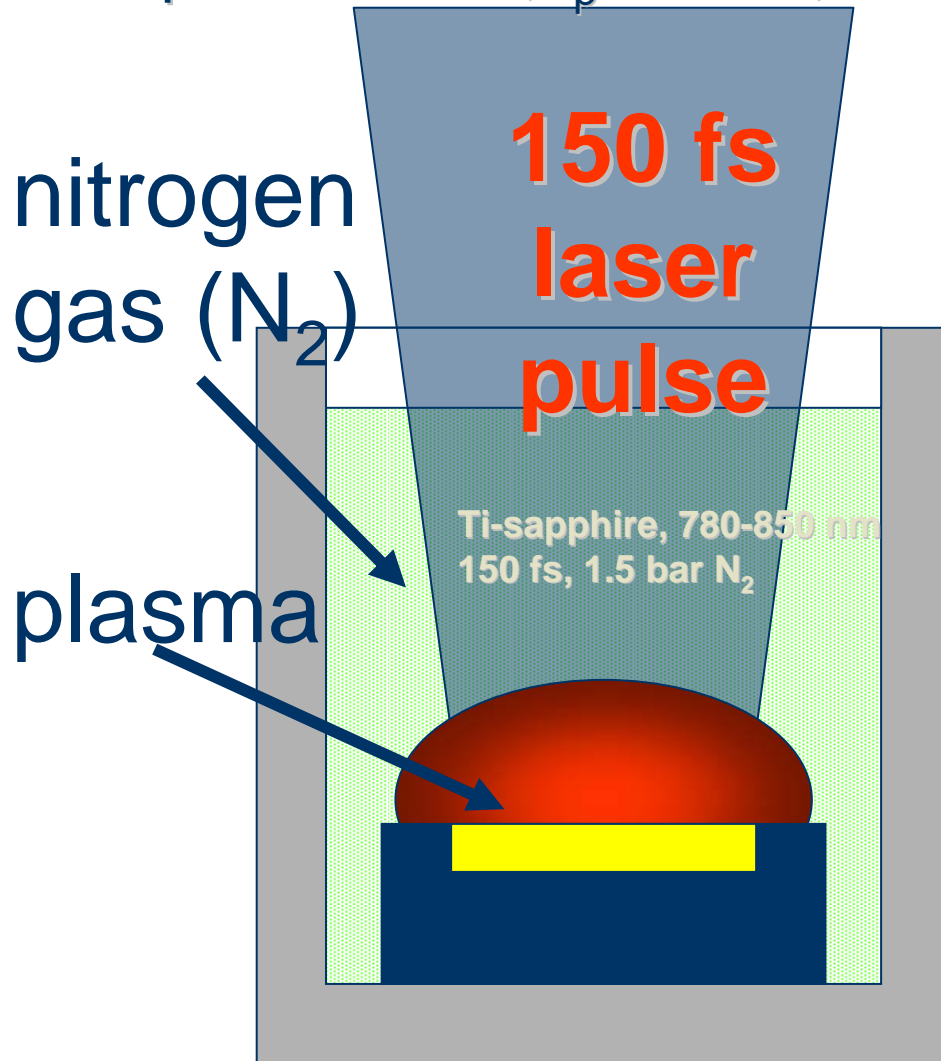


Aim: durable osseointegration and implant stability

Way: Surface must be a good stimulus for bone ingrowth (good microcontacts=osseointegration) very stable bone-implant-connection

chemical modification for chemical resistivity

Ti:Saphir mit CPA, $t_p=150$ fs, $\lambda=800$ nm



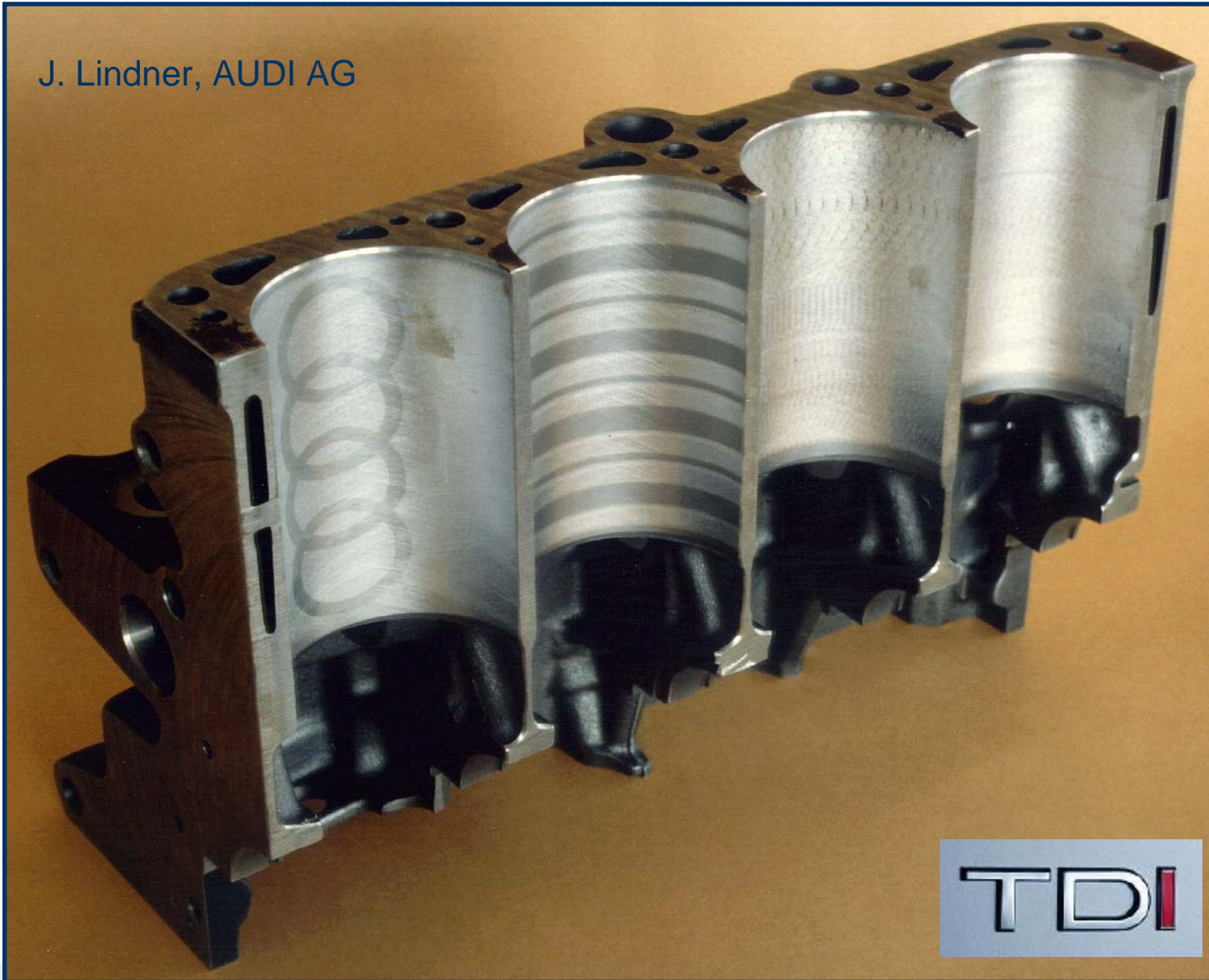
$t_p=1.5 \cdot 10^{-13}$ s (pulse duration):
 \Rightarrow non-thermal treatment
 (Coulomb explosion)

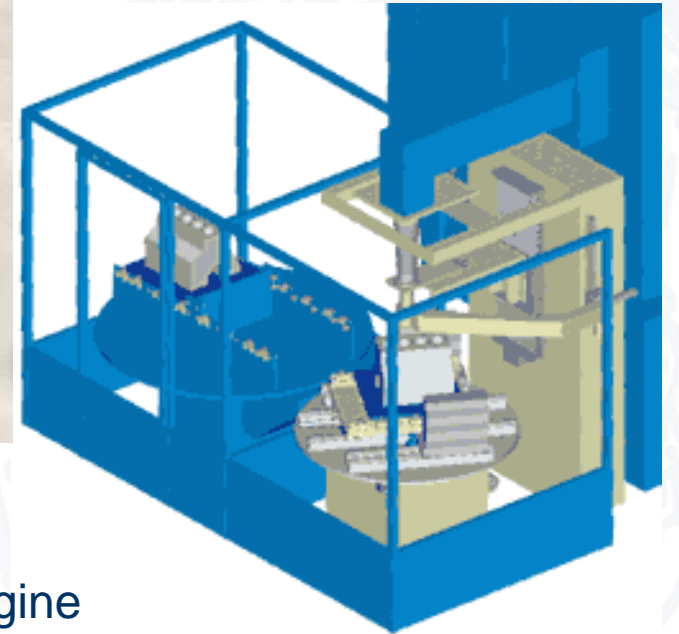
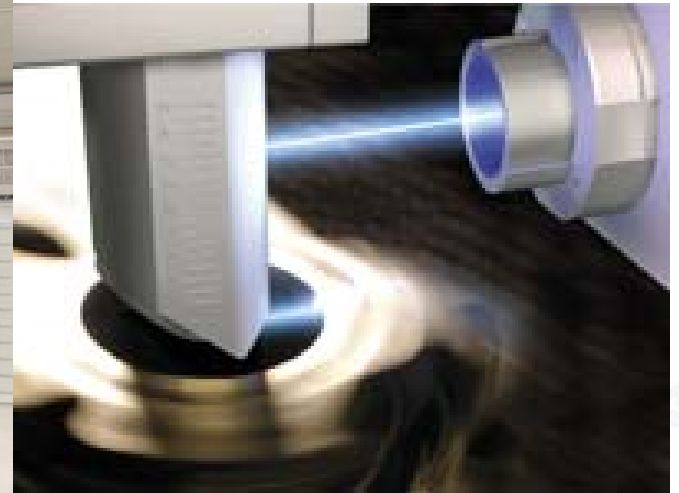
$$t_p \ll t_e \Rightarrow T_{\text{elec}} \gg T_{\text{latt}}$$

- affected depth ~ 10 nm
- plasma only **after** laser pulse
- highly ionized vapor

Industrial Applications

J. Lindner, AUDI AG



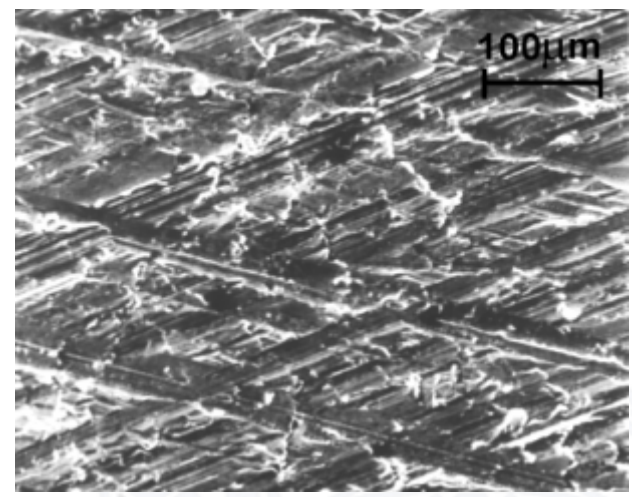


Treatment: mirror inside cylinder;
rotating engine block,
in series production, 5 Excimer simultaneous, 2 min/engine

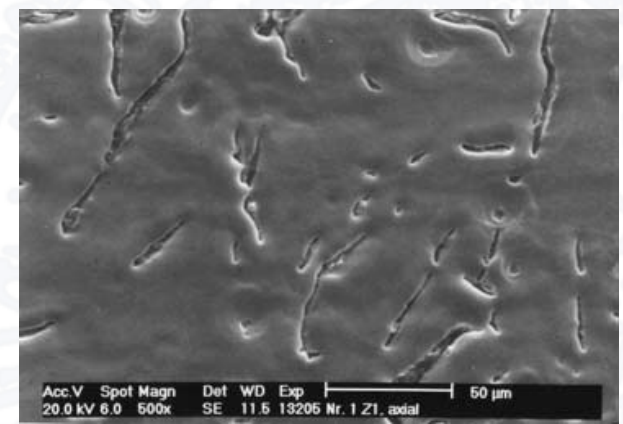




Foto: Audi Schilling Spezialbeschichtungen GmbH (re.)

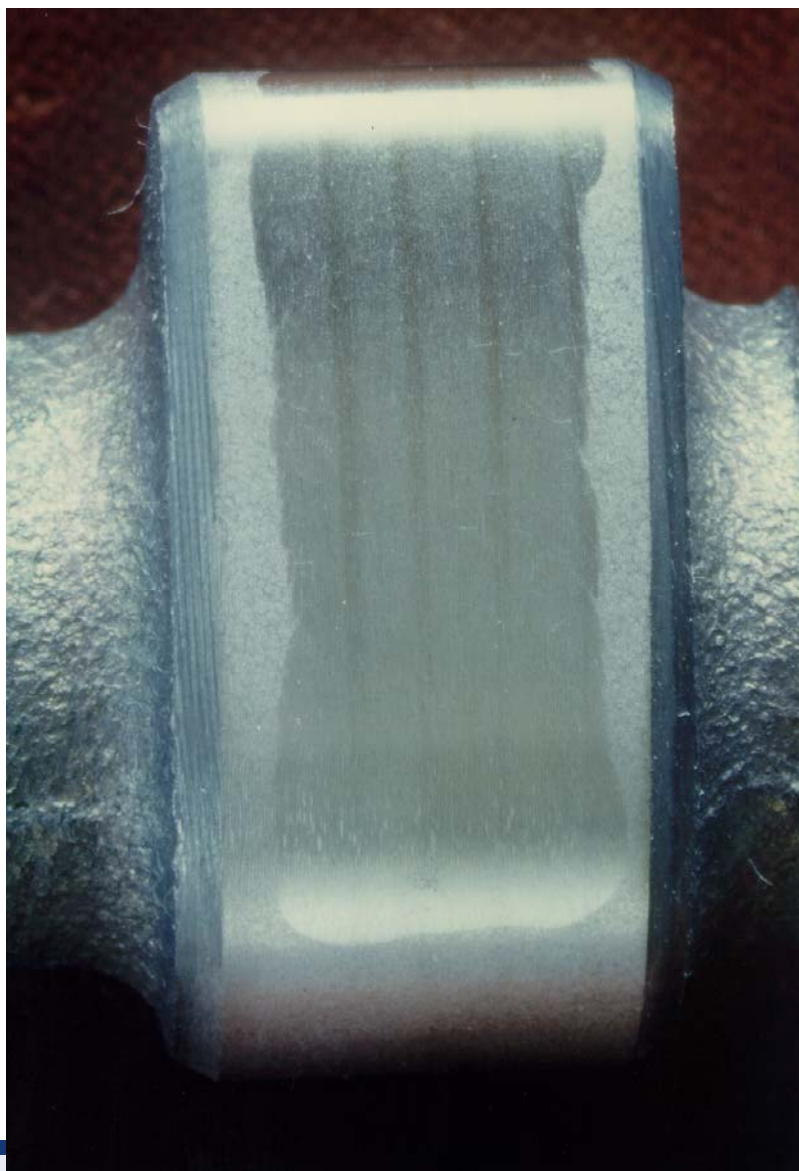
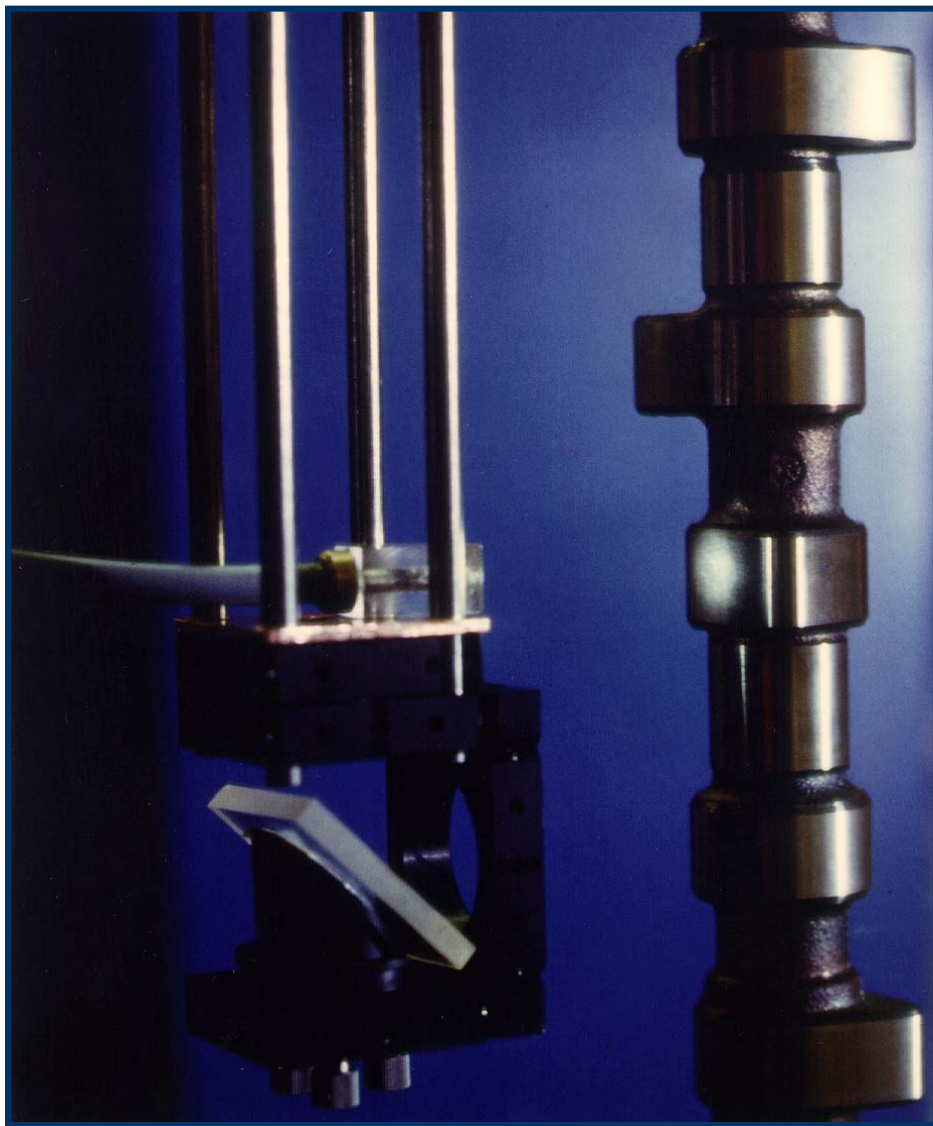


After honing



After laser treatment

Reduction of oil consumption (30x)
increase in efficiency and power



Summary



- Reactive Laser treatments enable flexible, clean and fast ways for the production of new materials, thin films and coatings
- Easy and fast modification and functionalizing of thin films and coatings by laser beams.
- But: sensitive adaptation of material, laser, and laser treatment for the specific application.
- Combination of several methods for resolving complicated processes and optimization of processing necessary.
- FEL is very attractive for fast (competitive) surface treatments
- Nanostructuring, Pulse tailoring
- Many Perspectives for thin films



- FEL, Jefferson Lab, Newport News, Virginia, USA
- AUDI AG, Daimler Benz, INA Wälzlager, IBM, Stihl
- Prof. H.-W. Bergmann (†), Uni Bayreuth, Metall. Werkstoffe
- Prof. A. Emmel, FH Amberg-Weiden, FB Maschinenbau
- Prof. M. Bamberger, Dr. W.D. Kaplan, Technion Haifa, Israel
- Prof. J. Wilden, TU Ilmenau, Fertigungstechnik
- BIAS Bremen, LZH Hannover
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E. Carpene, M. Kahle,
A. Müller, F. Landry, M. Han, S. Wagner,
C. Illgner, S. Dhar, S. Cusenza,

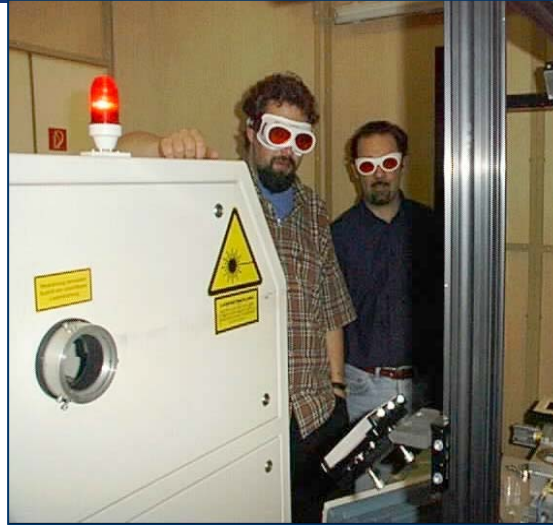
IWT Dresden: J. Kaspar

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Kevin Jordan

- **Your interest and patience**





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